Intramolecular Oxidative Phenol Coupling and Hydrazone-Initiated C-H Bond Insertion in Stemodane Synthesis:

A Reproduction of Work by Samir Chatterjee

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Abstract

The replication of the report entitled "Regioand Stereo-controlled Total Synthesis of the Stemodane Nucleus; an Unusual Diterprene Skeleton, a suspect publication by Dr. Samir Chatterjee", has been initiated. The Goal of this project is to evaluate the reproducibility of the synthetic pathway reported to produce the stemodane diterprene structre in figure 1, proposed by Dr. Chatterjee in 1979.

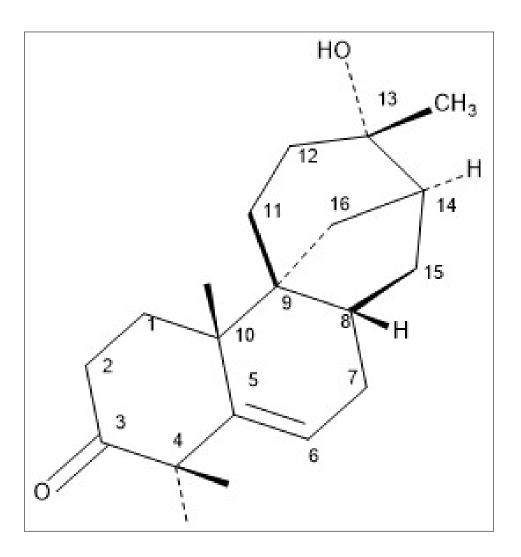


Figure 1: The desired Stemodane Nucleus [1].

In addition to other reports from this author having been refuted in published work[1][2], key elements of this stemodane report are believed to be inaccurate and improbable.

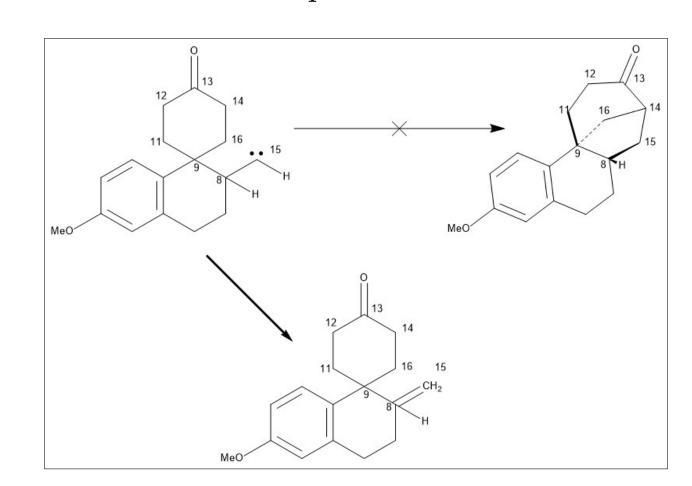


Figure 2: The suspect pathway [1].

Discussion

Based on the 3D renderings of the carbene intermediate, Figure 2, the geometric strain of this structure is evident. this suggests that C-15 to C-14 carbene bond formation is highly improbable.

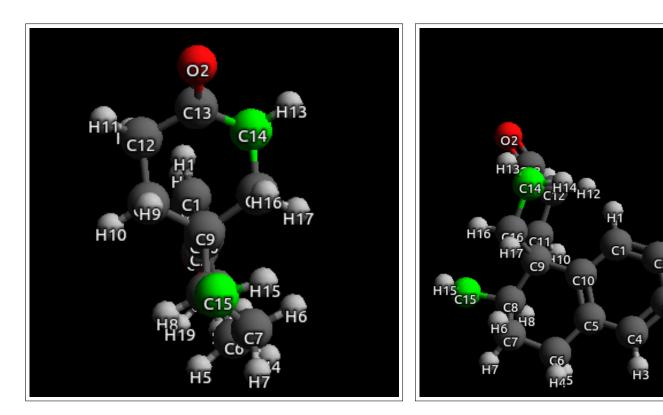
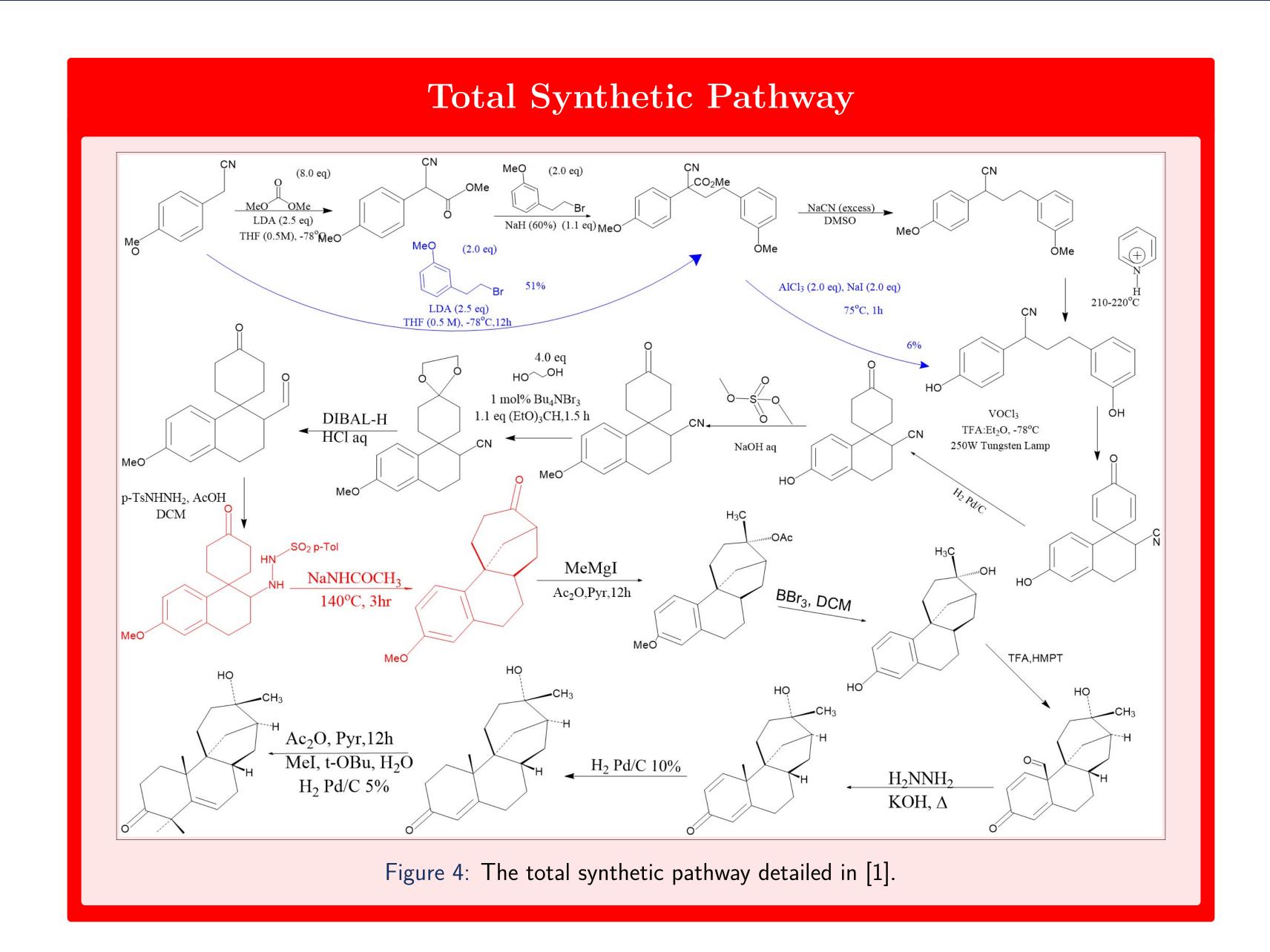


Figure 3: Left: The side view of the carbene intermediate Right: The front view of the carbene intermediate.



Initial Reactions

*Initial Conditions(Black)

Reaction	Conditions	Results
26	KHMDS (2.0 eq)	-No desired product formation
	THF (75-80°C) 17h	21% Starting material recovered
27	LDA (1.5 eq)	-Starting material recovered
	THF (75-80°C) 17h	
28	KHMDS (2.5 eq)	-Starting material recovered
	THF (75-80°C) 17h	
52	Py-HCl (2.5 eq)	-Starting material recovered
	(180°C) 3-8h	

Altered Reactions

*Altered Conditions (Blue)

 /		
Reaction	Conditions	Results
49	LDA (2.5 eq)	56% Product yield,
	THF (0.5M), (-78°C) 17h	Product NMR confirmed
50	LDA (2.5 eq)	44% Product yield,
	THF (0.5M), (-78°C) 17h	Product NMR confirmed
69	$BBr_3(3.0 eq)$	4% Product yield,
	DCM, (40°C) 12h	Product IR confirmed
75	$AlCl_3(2.0 eq), NaI (2.0 eq)$	6% Product yield,
	(75°C) 1h	Product IR confirmed

Figure 4 presents a mechanism in which the carbene intermediate at C-15 is stabilized by the electron inductive effects of the surrounding alkane carbons. However due to the highly reactive nature of carbenes leads to a hydride shift resulting in the formation of an olefin between C-8 and C-15; rather than proceeding with the theorized Hydrazone-initiated C-H insertion as the olefin reaction is significantly faster than the C-H insertion.

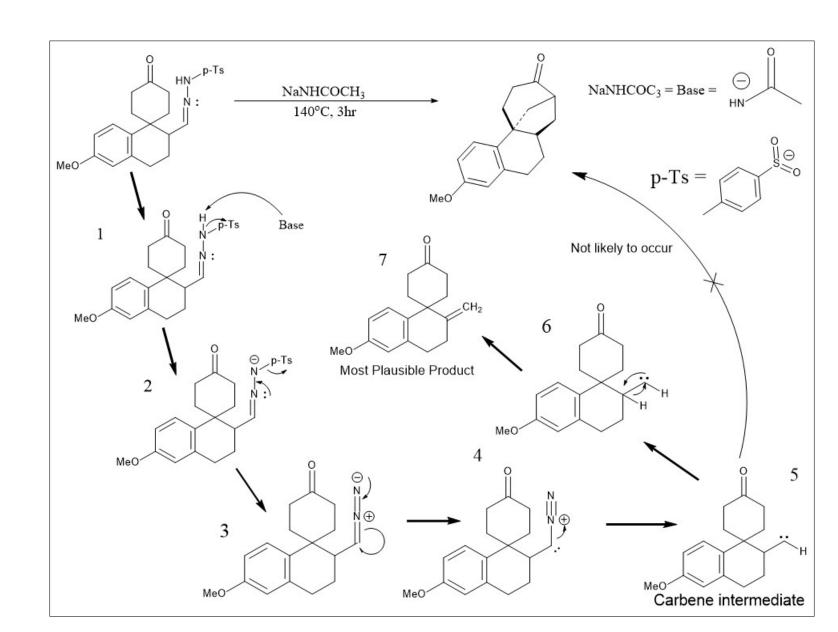


Figure 5: The mechanistic steps of the hydrazone initiated C-H insertion

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