# THE THERMAL AND MOMENTUM STRUCTURE 

OF AN EMERGING PLUME

A Thesis
Presented to

The Faculty of the Department of Chemical Engineering University of Houston

In partial Fulfillment<br>of the Requirements for the Degree Master of Science in Chemical Engineering

> by
> Shun-Kwok Tse
> December, 1977

## ACKNOTVLEDGEMENT

The author wishes to express deep appreciation to Professor H.W. Prengle, Jr. for his advice and guidance throughout this research.

Special thanks are due to Dr. Uday Mahagaokar, Professor F.L. Worley, Jr., and Dr. A.A. Siddiqi for their suggestion and inspiration during the courses of the project.

Many friends in the University helped by making the stay at $U$ of $H$ a pleasant and valuable experience.

The financial support from the Department of
Chemical Engineering is appreciated.
Finally the author wishes to give special thanks to his family for their care and encouragement.

# THE THERMAL AND MOMENTUM STRUCTURE 

OF AN EMERGING PLUME

## An Abstract of a Thesis Presented to

The Faculty of the Department of Chemical Engineering University of Houston

In Partial Fulfillment of the Requirements for the Degree Master of Science in Chemical Engineering
by
Shun-Kwok Tse
December, 1977

## ABSTRACT

An analytical study of the thermal and momentum structure of a compressible, axially symmetric turbulent plume was conducted. Experimental data from the literature and recent remote sensing data were analyzed. A finitedifference numerical technique was used to solve the equations of conservation of mass, momentum, and energy for the axial decay and radial distribution of temperature and velocity. Modifications of the dynamic eddy transfer coefficient given in the literature were used in the solutions. The numerical solutions show good agreement with the literature data and the remote sensing data.

The initial exit velocity and temperature ratio were found to be the most significant properties characterizing the variations of temperature and velocity. The axial temperature and velocity were found to decay faster as the temperature ratio is increased and as the initial velocity is decreased. The flat initial radial profiles were found to change gradually to a Gaussian profile, the characteristic radius describing the temperature being greater than that for the velocity distribution. Correlations for the eddy transfer coefficient, the radial temperature and velocity distribution coefficient, and the length of the core regions were obtained.

A generalized model for the axial decay of temperature and velocity was proposed consisting of, l) the core regions, where the exit temperature and velocity along the center line remain unchanged; and 2) the fully developed region where the behavior follows that of Priestly's model. The generalized model differs from Priestly:s model by addition of the core region, and agrees well with the literature data.

The most significant parameter of the model, the spreading coefficient, was correlated.

Emissions of pollutants as a function of excess air for combustion of gas, oil, and coal were studied. Literature data indicate that the concentration of unburned fuel, and carbon monoxide decrease exponentially as excess air increases from 0-20\% excess air, and above $20 \%$ most of unburned fuel and CO are negligible. $\mathrm{NO}_{\mathrm{x}}$ concentration increases approximately linearly as a function of excess air from 5-20\%, and above 25\% increases very slowly or remains constant. $\mathrm{SO}_{2}$ emissions primarily depend on the sulfur content of the fuel.

Signal processing of the remote sensing infrared inter-ferometry-spectroscopy data was studied and improved. The well-known CO P-R infrared structure was simulated from theory, and also the spectrogram was simulated in a manner corresponding to the field experiment. Interferences, not expected from the predicted composition of the plume, were found at the edges of the radiation envelope; a new frequency range was selected to minimize this effect. A new average value for the absorption coefficient was obtained from the theoretical intensity distribution of the spectral lines. These improvements produced a $10 \%$ increase in concentration for a high CO run.

## TABLE OF CONTENTS

Chapter Page
NOMENCLATURE ..... 1
I INTRODUCTION AND PURPOSE ..... 3
II THEORY OE EMERGING PLUME ..... 9
A. INTRODUCTION ..... 9
B. REVIEW OF LITERATURE ..... 19
C. MATHEMATICAL MODELING AND NUMERICAL SOLUTION ..... 21
III AXIAL TEMPERATURE AND VELOCITY DECAY ..... 34
A. ANALYSIS AND CORRELATION ..... 34
B. GENERALIZED EQUATIONS ..... 46
IV RADIAL TEMPERATURE AND VELOCITY DISTRIBUTIONS ..... 58
A. RADIAL DISTRIBUTION OF TEMPERATURE AND ..... 58 VERTICAL VELOCITY
B. DISTRIBUTION OF RADIAL VELOCITY ..... 67
C. BIBLIOGRAPHY FOR CHAPTERS I, II, III, \& IV ..... 69
V DISCUSSIONS, CONCLUSIOMS \& RECOMMENDATIONS ..... 71
VI EMISSIONS AS A FUNC'IION OF EXCESS AIR ..... 74
A. COMBUSTION OF GAS, OIL AND COAL ..... 74
B. VARIATIONS WITH EXCESS AIR ..... 79
C. EQUIPMENT DEVELOPMENTS ..... 87
D. CONCLUSIONS ..... 91
E. LITERATURE CITED ..... 91
VII IMPROVED IRRS SIGNAL PROCESSING METHOD ..... 96
A. INTRODUCTION ..... 96
B. THE IR SPECTRUM OF CO ..... 99
C. EQUIPMENT CHARACTERISTICS ..... 103
D. CO CONCENTRATION FROM IRRS ..... 104
E. RECOMMENDATIONS ..... 114
F. NOMENCLATURE ..... 115
G. LITERATURE CITED ..... 117
APPENDIX
A. SOLUTION OF FINITE-DIFFERENCE EQUATIONS ..... 118
B. FINITE-DIFFERENCE PROGRAM AND RESULT ..... 125
C. ANALYTICAL SOLUTION OF AXIAL TEMPERATURE ..... 155 AND VELOCITY DECAY
D. NON-IINEAR LEAST FIT PROGRAM AND RESULT ..... 159 FOR THE SPREADING COEFFICIENT
E. COMPUTER PROGRAMS FOR IRRS SIGNAL PROCESSING ..... 166

## LIST OF FIGURES

Figure Page
2-1 Plume Geometry and the Field of View ..... 10
2-2 Schematic Sketch of Free Jet ..... 18
2-3 Schematic Diagram of the Finite-Difference ..... 32 Net Work
3-1 Comparision of Experimental Data with Analysis ..... 35 for the Axial Decay of Temperature and Velocity
3-2 Comparision of Experimental Data with Analysis ..... 36for the Axial Decay of Temperature and Velocity
3-3 Comparision of Experimental Data with Analysis ..... 37 for the Axial Decay of Temperature and Velocity
3-4 Comparision of Experimental Data with Analysis ..... 38 for the Axial Decay of Temperature and Velocity
3-5 Comparision of Experimental Data with Analysis ..... 40 for the Axial Decay of Temperature and Velocity
3-6 Axial Profile of the Relative Intensity of ..... 41
Area Integrated Temperature Fluctuations
3-7: Axial Profile of the Relative Intensity of ..... 42
Area Integrated Temperature Fluctuations
4-1 Comparision of Experimental Data with Analysis ..... 59
for the Radial Distribution of Temperature and Velocity
4-2 Comparision of Experimental Data with Analysis ..... 6.0
for the Radial Distribution of Temperature and Velocity
4-3 Comparision of Experimental Data with Analysis ..... 61 for the Radial Distribution of Temperature and velocity
4-4 Comparision of Experimental Data with Analysis ..... 62 for the Radial Distribution of Temperature and Velocity

## Figure

Page
4-5 Radial Temperature and Velocity Distribution at Short Distance ..... 63
4-6 Comparision of Experimental Data with Analysis ..... 64 for the Radial Distribution of Temperature
4-7 Distribution of Radial Velocity ..... 68
6-1 Effect of Excess Air on Unburned Combustibles ..... 80
6-2 Effect of Excess Air on Pollutant Emissions from Combustion of Natural Gas ..... 82
6-3 Effect of Excess Air on Pollutant Emissions from a Pulverised Coal Unit ..... 83
6-4 Theoretical NO vs. Flame Temperature ..... 84
6-5 $\mathrm{NO}_{\mathrm{x}}$ vs. Excess $\mathrm{O}_{2}$ in Flue Gas ..... 84
6-6 NO vs. Excess $O_{2}$ for Balanced Draft Heater ..... 85 wi 宏h Combustion Air Preheat
86
6-7 NO $\mathrm{N}_{\mathrm{x}}$ vs. Excess $\mathrm{O}_{2}$ for a Forced Draft Heater
88
6-8 $\mathrm{NO}_{\mathrm{x}}$ Emission vs. Excess $\mathrm{O}_{2}$ for Oil Fired Unit
6-9 NO ${ }_{x}$ Emission vs. Excess Oxygen for Diffusion ..... 88 Flames Coal Fired Unit
6-10 $N_{x}$ vs. Excess Air for Front Wall and ..... 89
Tangentially Coal Fired Unit
6-11 $\mathrm{NO}_{\mathrm{x}}$ Emissions vs. Excess Air for Different ..... 90 Coal Composition
6-12 Schematic Showing Method of Air Admission for ..... 92 Two Stage Combustion Process
6-13 Effect of Gas Recirculation and Two Stage ..... 93 Combution on Combustion Temperature
6-14 Effect of Gas Recirculation and Two Stage ..... 94 Combustion of $\mathrm{NO}_{x}$ Formation
7-1 Infrared Emission Spectrum of Plume from Furnace Operated with Zero Excess Air (a) $0 \mathrm{~cm}^{-1800 \mathrm{~cm}^{-1}}$
97
97
(b) $1800 \mathrm{~cm}^{-1}-3600 \mathrm{~cm}^{-1}$ ..... 98
7-2 The Vibration-Rotation Spectrum of $C O$ under ..... 100 High Resolution
7-3 Simulated Spectrogram of CO ..... 102
7-4 Simulation of Spectrogram ..... 107
7-5 Simulated Spectrogram of CO Resembling Experimental Condition with Starting Sampled Point at
(a) $1993.12 \mathrm{~cm}^{-1}$ ..... 109
(b) $1993.32 \mathrm{~cm}^{-1}$ ..... 110
7-6 CO Spectrogram from Experimental Data ..... 111
A-1 The Finite-Difference Grid System ..... 121

## LIST OF TABLES

Table Page
2-l Mahagaokar's Axial Temperature Data ..... 11
2-2 Mahagaokar's Radial Temperature Data ..... 12
2-3 Corrsin and Uberoi's Axial Temperature and ..... 13 Velocity Data
2-4 Corrsin and Uberoi's Radial Temperature and ..... 14 Velocity Data
2-5 Tomich and Weger's Axial Temperature and ..... 15 Velocity Data
2-6 Tomich and Weger's Radial Temperature and ..... 16 Velocity Data
3-1 Values of $Z_{c t}, Z_{c v}$ and $F(M)$ ..... 44
3-2 Values of $\alpha$ from Scarbororough's Non-Linear ..... 55 Least Square Fit Method
4-1 Variation of $\beta$ and $\eta$ with $u_{o}, t_{o} / t_{a}$, and $z$ ..... 66
6-1 Emission Factors for Natural Gas Combustion ..... 75
6-2 Emission Factors for Fuel Oil Combustion ..... 76
6-3 Emission Factors for Bituminous Coal ..... 77 Combustion
6-4 Emission Factors From Anthracite Coal ..... 79Combustion without Control Equipment

## Upper Case

| $\mathrm{A}_{1}, \mathrm{~A}_{2}$ | integration constants given by Eq. (3-21) and (3-22) |
| :---: | :---: |
| D | jet diameter |
| F | radial heat flux |
| M | initial jet Mach number |
| ${ }^{\mathrm{N}} \mathrm{Pr}$ | Prandlt number |
| $\mathrm{N}_{\mathrm{Ec}}$ | Eckert number, defined by Eq. (2-17) |
| $\mathrm{P}, \mathrm{P}_{\mathrm{a}}$ | dimensionless density, $\rho / \rho_{o}$, and $\rho_{a} / \rho_{0}$ respectively |
| . 2 | dimensionless radial coordinate, r/D |
| $\mathrm{R}_{\mathrm{m}}$ | the radial extent of the jet |
| T | dimensionless temperature, ( $\left.t-t_{a}\right) /\left(t_{0}-t_{a}\right)$ |
| $U_{Z}, U_{R}$ | dimensionless verticle and radial velocity respectively, $u_{z} / u_{o}, u_{r} / u_{o}$ |
| Z | dimensionless axial coordinate, z/D |
| $\mathrm{Z}_{\mathrm{cv}} \mathrm{Z}^{\mathrm{Z}} \mathrm{ct}$ | dimensionless core length for momentum and heat flux of the jet |

Lower Case
$c_{p}$
$r$
$t_{,} t_{a}, t_{o}$
$t_{m}$
$t_{A} r m s$
$t_{A}$
$u_{r}, u_{z}, u_{o}$
heat capacity
radial coordinate of the plume
temperature, ambient temperature, and initial jet temperature
temperature at $r=0$., centre line
root mean square of the fluctuating temperature component
average area integrated temperature
radial, vertical, and initial jet velocity respectively

| $u_{z m}$ | vertical velocity at $r=0$, centre line |
| :--- | :--- |
| $z$ | vertical coordinate of the jet |
| $z_{o}$ | $D / 2 \alpha$ |
| $z^{\prime}$ | vertical coordinate described by equation (3-19) |

## Greek Letters

a spreading coefficient of the jet
$\alpha_{t}, \alpha_{v} \quad$ spreading coefficient for temperature and velocity decay respectively

B constant for the radial distribution of temperature in Eq. (4-l)
$\Delta R \quad$ finite difference step size in radial direction
$\Delta t_{m} \quad t_{m}-t_{a}$
$\Delta t_{0} \quad t_{0}-t_{a}$
$\Delta Z \quad$ finite different step size in axial direction
$\varepsilon_{v} \quad$ eddy kinematic viscosity
$n$ constant for the radial distribution of velocity in Eq. (4-2)
$\rho_{r} \rho_{a} \rho_{0}$ density, ambient air density, and exit density of
$\tau$
radial shear stress

## CHAPTER I

## INTRODUCTION AND PURPOSE

In recent years considerable interest has been shown in the monitoring of gaseous pollutants from stationary emission sources by the use of electro-optical techniques. Prengle and coworkers [17] have conducted investigations of remote sensing infrared radiometry-spectroscopy (IRRS) techniques using Fourier transform spectroscopic equipment to determine the temperature and concentration of pollutants in an emerging plume.

A subject of great relevance to meterologists, engineers and physicists studying the behavior of gaseous pollutants is the thermal and momentum structure of an emerging plume into the atmosphere. Of particular interest is the problem of determining the mechanism by which the energy in the hot gases dissipates and contributes to the dispersion of pollutants into the atmosphere.

So far it has been extremely difficult to make thermal and momentum measurements on actual plumes of substantial size, and essentially impossible to follow the temperature and velocity gradients, axially and radially. Some measurements using a helicopter have been attempted, but these results have definite limitations since the process of measurement alters the flow pattern of the plume $[10,28,6]$. Hence no body of information exists on the phenomena involved, except
on emerging laboratory jets in which the flow is also disturbed by the measuring device [20].

Mahagaokar and Prengle [14] by the application of the aforementioned IRRS technique were able to obtain some preliminary results consisting of axial gradients, radial gradients, and turbulent temperature fluctuations within the plume.

Several models exist in the literature for the theoretical treatment of the behavior of a continuous plume of bouyant gas of smoke moving through various types of atmospheres. A thorough review and critique of these theoretical investigations has been done by Briggs [5] in the light of the usefulness of these treatments for calculating the height to which plumes in different conditions will rise.

The first analysis of this problem was done by Schmidt [25] who studied the behavior of convective plumes of air above steady point and line sources of heat in a uniform incompressible atmosphere. Schmidt observed that a turbulent plume is physically confined to a conical region and found the distribution of temperature and velocity by balancing the horizontal turjulent transfer of heat and monentum against the vertical transfer by convection making an allowance for the effect of buoyance. Schmidt assumed geometrical and mechanical similarity of the processes in horizontal sections of the plume, and used

Prandlt's mixing length theory of turbulence to find the complete forms of the velocity and temperature profiles. He experimentally verified the calculated results for the point source by using small electrically heated grids in air.

Later Rouse, Yip, and Humphreys [22] treated the same problem for heated plumes in calm air by assuming eddy viscosity diffusion of the buoyance and momentum by a process analogous to molecular diffusion. The treatments by Schmidt and Rouse, et al. resulted in solutions in which the radius of the plume $r(z)$ is proportional to the distance $z$ from the source, the vertical velocity $u_{z}$ on the plume axis is proportional to $z^{-1 / 3}$, and the temperature excess (on the plume axis) above the ambient temperature $\Delta t$ is proportional to $z^{-5 / 3}$.

About the same time, Sutton [26] developed a simple theory for a buoyant in a calm atmosphere. As in Schmidt's theory, Sutton assumed the excess temperature in the jet, $\Delta t$, to be small compared to the absolute temperature of the air as a whole and took the missing length to be proportional to the radius of the jet at any level. He departed from the Prandtl theory by supposing that the velocity fluctuations in the jet are proportional to the rate of decrease of velocity downstream. The rate of entrainment of air by turbulent mixing on the boundary of the jet is expressed as a function of the exposed area, the velocity fluctuation and the mixing length. Sutton obtained an analytical solution in terms of
an unknown constant which he determined from experimental observations, on smoke clouds. Contrasted with Schmidt's solution, $r(z)$ is proportional to $z^{0.88}, u_{z}$ is proportional to $z^{-0.29}$ and $\Delta t$ is proportional to $z^{-1.46}$. Sutton showed that his solution fits the experimental data of Schmidt slightly better than Schmidt's original solution.

The problem of convection from a point source was also studied by Batchelor [3] who used dimensional analysis and included the phenomenon of stratification of the environment, i.e., variation of the atmospheric temperature with height above the source. He found power law expressions for the mean plume velocity and temperature as functions of height in an unstable atmosphere whose potential temperature gradient is also approximated by a power law. Batchelor observed that the mass flow across a section of the plume is proportional to $z^{5 / 3}$ and this increasing mass flow is provided by entrainment of stationary air at the edge of the plume. He found that the required horizontal mean inflow velocity is proportional to the mean vertical velocity and the ratio of the two determine the cone angle of the plume.

A generalized model for the turbulent convection from maintained and instantaneous sources under different types stratification was developed by Morton, Taylor, and Turner [15]. They used Taylor's entrainment hypothesis and assumed. that the lateral profiles of vertical velocity and buoyance
are similar at all heights and that the fluids are incompressible and do not change volume on mixing, and that the local variations in density throughout the motion are small compared to some reference density. The governing equations are derived in nondimensional form for the conservation of volume, momentum and buoyance where the buoyance is related to the excess temperature $\Delta t$ through a coefficient of cubical expansion. A numerical solution is obtained for the case of the maintained source which leads to a prediction of the final height to which a plume of light fluid will rise in a stable stratified environment. The constant governing the rate of entrainment was estimated by comparing the theory with experimental results.

The problem of a convective plume from a finite circular source under different conditions of environmental stratification was also considered by Priestly and Ball [18]. Their treatment is similar to the one above except that Taylor's entrainment hypothesis is replaced by an energy equation involving an assumption above the magnitude and distribution of the turbulent stress.

All the above models assumed that the flow was fully developed and similarity of the profiles exists. The fact that the emerging plume only becomes fully developed at a certain number of nozzle diameters downstream from the stack exit was not considered. Mahagaokar's experimental data
(average area integrated temperature) show that near the stack exit, the axial gradient is small, whereas the above models predict a large gradient.

The purpose of the work described in this thesis is: (I) to study the thermal and momentum structure of an emerging plume; and (2) to propose a model for the axial and radial temperature and velocity decay of the emerging plume.

As already mentioned, temperature and velocity data for emerging plumes are very scarce, only laboratory free jet data are available. Several sets of free jet experimental data are used to verify the solutions for the temperature and velocity distribution of the emerging plume, and used for correlations. The term "free jet" is used most of the time in place of "emerging plume," as a more appropriate and convenient term.

Chapters reporting work on 1) Emissions as Function of Excess Air, and 2) Improved IRRS Signal Processing Method are included as a matter of record.

## CHAPTER II

THEORY OF EMERGING PLUME

## A. Introduction

In order to compare the average area integrated temperature, $t_{A}$, of the field of view (F.O.V.), see Figure 2-1, obtained by the IRRS technique [14] with that from theoretical mathematical solutions, it is necessary to develop a good general solution for velocity and temperature variations in a free jet. Mahagaokar's [14] experimental data for the axial and radial temperature distribution are summarized in Tables 2-1 and 2-2.

A computational technique based on finite difference solution of the boundary layers forms for conservation of mass, momentum and energy, was developed to calculate both the axial velocity and temperature decay and the radial distribution of velocity and temperature. The method can be extended to a variety of boundary conditions different than the ones used in the present work. Corrsin and Uberoi's [9] and Tomich and Weger's [29] experimental data as listed in Tables $2-3,2-4,2-5,2-6$ were used to verify the numerical technique and for correlation purposes.

The structure of the compressible, axially symmetric free jet can be described in terms of three flow regions;


Figure 2-1. Plume Geometry and the Field of View

Table 2-1 Mahagaokar's Axial Temperature Data

| $\begin{aligned} & =3.5 \mathrm{ft} \\ & =37.3 \\ & \mathrm{O}=511.0 \\ & =292.3 \\ & \mathrm{a} / \mathrm{t}_{\mathrm{a}}=1 . \end{aligned}$ | /sec |  |  |
| :---: | :---: | :---: | :---: |
| z/D | $\begin{gathered} \mathrm{t}_{\mathrm{A}} \\ \left({ }^{\circ} \mathrm{K}\right) \end{gathered}$ | $\begin{aligned} & t^{\prime} \mathrm{A} \mathrm{rms} \\ & \left({ }^{\circ} \mathrm{K}\right) \end{aligned}$ | $\frac{t^{\prime} \mathrm{A} r m s}{t_{\mathrm{A}}}$ |
| 0.47 | 507 | 7.38 | 0.0146 |
| 0.76 | 502 | 8.05 | 0.0160 |
| 1.05 | 488 | 8.29 | 0.0170 |
| 1.34 | 472 | 10.2 | 0.0216 |
| 1.63 | 452 | 21.4 | 0.0474 |
| 2.21 | 436 | 18.4 | 0.0422 |
| 2.50 | 425 | 15.4 | 0.0362 |

$D=3.5$
$\mathrm{u}_{\mathrm{O}}=37.3 \mathrm{ft} / \mathrm{sec}$
$t_{0}^{\circ}=511.0^{\circ} \mathrm{K}$
$t^{\circ}=292.3^{\circ} \mathrm{K}$
$t_{o}^{a} / t_{a}=1.74$

| $z / D$ | $t_{A}$ <br> $\left({ }^{\circ} \mathrm{K}\right)$ | $t^{\prime}{ }_{A}$ <br> $\left({ }_{\mathrm{K}}\right)$ | $\frac{t^{\prime}{ }_{A} \mathrm{rms}}{t_{A}}$ |
| :---: | :---: | :---: | :---: |
| 0.47 | 502 | 7.68 | 0.0153 |
| 0.88 | 500 | 8.66 | 0.0173 |
| 1.22 | 476 | 9.58 | 0.0201 |
| 1.57 | 455 | 17.1 | 0.0376 |
| 1.92 | 442 | 20.7 | 0.0468 |
| 2.26 | 430 | 17.8 | 0.0414 |
| 2.61 | 421 | 15.7 | 0.0373 |

Table 2-2 Mahagaokar's Radial Temperature Data

| $\begin{aligned} & D=3.5 \\ & u_{o}^{o}=31.5 \\ & t_{0}^{o}=508^{\circ} \mathrm{K} \\ & t_{a}^{o}=297.2^{\circ} \mathrm{K} \\ & t_{o}^{a} / t_{a}=1.71 \end{aligned}$ |  |  |  |
| :---: | :---: | :---: | :---: |
| At $z / D=0.6$ |  |  |  |
| $r / D$ | $\begin{gathered} t_{A} \\ \left(K^{\circ}\right) \end{gathered}$ | $\begin{aligned} & t^{\prime} \mathrm{A} \text { rms } \\ & \left({ }^{\circ} \mathrm{K}\right) \end{aligned}$ | $\frac{t_{A}^{\prime} \mathrm{rms}}{t_{\mathrm{A}}}$ |
| 0.359 | 400 | 15.2 | 0.0380 |
| 0.293 | 457 | 13.1 | 0.0287 |
| 0.206 | 479 | 12.9 | 0.0269 |
| 0.115 | 495 | 10.5 | 0.0212 |
| 0 | 500 | 9.14 | 0.0183 |
| -0.115 | 490 | 10.6 | 0.0216 |
| -0.206 | 477 | 12.3 | 0.0258 |
| -0.293 | 453 | 14.1 | 0.0311 |
| -0.359 | 403 | 14.2 | 0.0352 |

$$
\begin{aligned}
& D=3.5 \\
& u_{o}=29.0 \\
& t_{0}^{0}=498^{\circ} \mathrm{K} \\
& t^{\circ}=296.2^{\circ} \mathrm{K} \\
& t_{o}^{a} / t_{a}=1.69 \\
& \text { At } z / D=0.6
\end{aligned}
$$

| $r / D$ | $t_{A}$ <br> $\left(K^{\circ}\right)$ | $t^{\prime}{ }_{A} r m s$ <br> $\left({ }^{\circ} \mathrm{K}\right)$ | $\frac{t^{\prime}{ }_{A r m s}}{t_{A}}$ |
| :---: | :---: | :---: | :---: |
| 0.359 | 406 | 13.8 | 0.0340 |
| 0.293 | 454 | 14.2 | 0.0313 |
| 0.206 | 475 | 11.2 | 0.0236 |
| 0.115 | 486 | 11.1 | 0.0228 |
| 0 | 491 | 9.36 | 0.0191 |
| -0.115 | 488 | 11.6 | 0.0238 |
| -0.207 | 477 | 13.4 | 0.0281 |
| -0.282 | 451 | 13.2 | 0.0293 |
| -0.368 | 401 | 15.6 | 0.0389 |

Table 2-3 Corrsin and Uberoi's Axial Temperature and Velocity Data

|  | $\begin{aligned} & 1 " \\ & =78 \mathrm{ft} \\ & =291{ }^{\circ} \mathrm{K} \\ & / t_{a}=2.0 \end{aligned}$ |  | $\begin{aligned} & \mathrm{D}=1 \\ & \mathrm{u}_{0}= \\ & \mathrm{t}_{0}^{0} \\ & \mathrm{t}_{\mathrm{o}}^{\prime} \end{aligned}$ | $\begin{aligned} & \mathrm{ft} / \mathrm{se} \\ & 1 \mathrm{o}^{\circ} \mathrm{K} \\ & =1.05 \end{aligned}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $z / D$ | $\frac{t-t_{a}}{t_{0}-t_{a}}$ | $\frac{u_{z}}{u_{0}}$ | $\mathrm{Z} / \mathrm{D}$ | $\frac{t-t_{a}}{t_{0}-t_{a}}$ | $\frac{\mathrm{u}_{\mathrm{z}}}{\mathrm{u}_{\mathrm{o}}}$ |
| 0 | 1.0 | 1.0 | 0 | 1.0 | 1.0 |
| 1.0 | 1.0 | 1.0 | 1.0 | 1.0 | 1.0 |
| 2.0 | 1.0 | 1.0 | 2.0 | 1.0 | 1.0 |
| 3.0 | 0.96 | 1.0 | 3.0 | 1.0 | 1.0 |
| 4.0 | 0.90 | 0.97 | 4.0 | 0.96 | 1.0 |
| 6.0 | 0.67 | 0.80 | 6.0 | 0.80 | 0.94 |
| 8.0 | 0.50 | 0.60 | 8.0 | 0.60 | 0.80 |
| 10.0 | 0.40 | 0.48 | 10.0 | 0.50 | 0.67 |
| 12.0 | 0.32 | 0.39 | 12.0 | 0.41 | 0.58 |
| 14.0 | 0.26 | 0.33 | 14.0 | 0.32 | 0.48 |

Table 2-4 Corrsin and Uberoi's Radial Temperature and Velocity Data

$$
D=1 "
$$

$u_{0}=78 \mathrm{ft} / \mathrm{sec}$
$t^{0}=588{ }^{\circ} \mathrm{K}$
$t^{\circ}=291{ }^{\circ} \mathrm{K}$
$t_{o}^{a} / t_{a}=2.02$

At $z / D=15$

| $r / D$ | $\frac{t-t_{a}}{t_{z 0} t_{a}}$ | $\frac{u_{z}}{u_{z O}}$ |
| :---: | :--- | :--- |
| 0 | 1.0 | 1.0 |
| 1.0 | 0.95 | 0.87 |
| 1.5 | 0.78 | 0.70 |
| 2.0 | 0.60 | 0.42 |
| 2.5 | 0.25 | 0.35 |
| 3.0 | 0.15 | 0.14 |

$$
\begin{aligned}
& D=1 " \\
& u_{0}^{\prime}=78 \mathrm{ft} / \mathrm{sec} \\
& t_{0}^{0}=305{ }^{\circ} \mathrm{K} \\
& t_{a}^{0}=291{ }^{\circ} \mathrm{K} \\
& t_{o}^{a} / t_{a}=1.05 \\
& \text { At } z / D=15
\end{aligned}
$$

| $r / D$ | $\frac{t-t_{a}}{t_{z O}-t_{a}}$ | $\frac{u_{z}}{u_{z O}}$ |
| :---: | :---: | :--- |
| 0. | 1.0 | 1.0 |
| 0.5 | 0.9 | 0.86 |
| 1.0 | 0.70 | 0.62 |
| 1.5 | 0.5 | 0.38 |
| 2.0 | 0.30 | 0.20 |
| 2.5 | 0.12 | 0.07 |
| 3.0 | 0.03 | 0.02 |

Table 2-5 Tomich and Weger's Axial Temperature and Velocity Data

| $\begin{aligned} & D=1 / 4 " \\ & u_{o}=822 \mathrm{ft} / \mathrm{sec} \\ & t_{0}^{0}=785 \circ^{\circ} \mathrm{K} \\ & t_{0}^{0}=293 \circ^{\circ} \mathrm{K} \\ & t_{o}^{a} / t_{a}=2.68 \end{aligned}$ |  |  | $\begin{aligned} & D=3 / 8^{\prime \prime} \\ & u_{0}^{o}=687 \mathrm{ft} / \mathrm{sec} \\ & t_{0}=797^{\circ} \mathrm{K} \\ & t_{0}^{0}=293^{\circ} \mathrm{K} \\ & t_{0}^{\mathrm{a}} / \mathrm{t}_{\mathrm{a}}=2.72 \end{aligned}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| z/D | $\frac{t-t_{a}}{t_{0}-t_{a}}$ | $\frac{u_{z}}{u_{0}}$ | z/D | $\frac{t-t_{a}}{t_{0}-t_{a}}$ | $\frac{u_{z}}{u_{0}}$ |
| 0 | 1.0 | 1.0 | 0 | 1.0 | 1.0 |
| 1.0 | 1.0 | 1.0 | 1.0 | 1.0 | 1.0 |
| 2.0 | 0.98 | 1.0 | 2.0 | 1.0 | 1.0 |
| 3.0 | 0.97 | 1.0 | 3.0 | 0.94 | 0.98 |
| 4.0 | 0.92 | 0.98 | 4.0 | 0.90 | 0.96 |
| 5.0 | 0.85 | 0.95 | 5.0 | 0.83 | 0.91 |
| 6.0 | 0.78 | 0.87 | 6.0 | 0.76 | 0.88 |
| 8.0 | 0.64 | 0.72 | 8.0 | 0.64 | 0.72 |
| 10.0 | 0.54 | 0.59 | 10.0 | 0.52 | 0.57 |
| 12.0 | 0.45 | 0.48 | 12.0 | 0.41 | 0.46 |

Table 2-6 Tomich and Weger's Radial Temperature and Velocity Data

| At $z / D=4.0$ |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | At $z$ | 11.0 |  |
| $r / D$ | $\frac{t-t_{a}}{t_{z O}-t_{a}}$ | $\frac{u_{z}}{u_{z 0}}$ | $r / D$ | $\frac{\mathrm{t}-}{\mathrm{t}_{\mathrm{zO}}}$ | $\frac{\mathrm{u}_{\mathrm{z}}}{\mathrm{u}_{\mathrm{zO}}}$ |
| 0.0 | 1.0 | 1.0 | 0.0 | 1.0 | 1.0 |
| 0.1 | 0.99 | 0.99 | 0.30 | 0.95 | 0.90 |
| 0.2 | 0.96 | 0.99 | 0.60 | 0.82 | 0.70 |
| 0.3 | 0.93 | 0.92 | 0.95 | 0.63 | 0.49 |
| 0.4 | 0.89 | 0.61 | 1.30 | 0.50 | 0.30 |
| 0.6 | 0.62 | 0.44 | 1.60 | 0.30 | 0.18 |
| 0.8 | 0.40 | 0.20 | 1.90 | 0.20 | 0.11 |
| 1.0 | 0.22 | 0.06 | 2.20 | 0.10 | 0.06 |
| 1.2 | 0.11 | 0.02 | 2.60 | 0.05 | 0.03 |

$$
\begin{aligned}
& D=3 / 8^{\prime \prime} \\
& u_{o}=687 \mathrm{ft} / \mathrm{sec} \\
& t_{0}^{0}=797{ }^{\circ} \mathrm{K} \\
& t_{0}^{0}=293{ }^{\circ} \mathrm{K} \\
& t_{0}^{\mathrm{a}} / \mathrm{t}_{\mathrm{a}}=2.72
\end{aligned}
$$

At $z / D=11.0$

| $r / D$ | $\frac{t-t_{a}}{t_{z O}^{-t}}$ | $\frac{u_{z}}{u_{z O}}$ |
| :--- | :--- | :--- |
| 0.0 | 1.0 | 1.0 |
| 0.30 | 0.96 | 0.92 |
| 0.50 | 0.85 | 0.77 |
| 0.80 | 0.72 | 0.59 |
| 1.10 | 0.57 | 0.42 |
| 1.30 | 0.42 | 0.20 |
| 1.60 | 0.30 | 0.20 |
| 1.85 | 0.19 | 0.12 |
| 2.40 | 0.04 | 0.01 |

see Figure 2-2. The potential core region extends to the point where the turbulent mixing region reaches the jet axis. This region is, of course, not necessarily identical for velocity and temperature. The next region is a transition region where the spread of the jet in the radial direction and the axial decay of the center-line properties are very rapid, and is followed by the fully developed region where the radial spread of the jet is nearly linear with respect to the axial distance. The greater the compressible effects, the greater is the deviation from linearity.

There are many problems which arise in connection with any numerical approach to the problem of solving for temperature and velocity as a function of position in the compressible free jet. One is the non-existance of similarity [12] in the profiles (as distinct from the case for incompressible flow). The model must also take into account the fact that the free jet only becomes fully developed a certain number of nozzle diameters downstream from the nozzle. Therefore, a complete treatment must include the potential core region and the transition region, as well as the fully developed portion of the jet. The boundary conditions are also a source of difficulty. These include the prescribed conditions at the nozzle (i.e., flat or other types of profiles) and the conditions at an infinite radial distance (i.e., where the jet is exhausting into a moving gas stream of into quiescent surroundings).


Figure 2-2. Schematic Sketch of Free Jet

Because of these problems, most of the available calculational techniques have had to make use of various simplifying assumptions which has limited their usefulness. Much of the work has been restricted to the problem of axial velocity and temperature decay, while the question of the radial variation of these quantities has not been dealt with extensively. It is the intent of this numerical technique to overcome some of these limitations.

## B. Review of Literature

Hinze and van der Hegge Zijmen [11], Tollmien [50], Schlichting [24], and others have obtained solutions to the problem of determining the velocity and temperature profiles in incompressible, turbulent free jets. These solutions, however, are somewhat limited in that they do not apply to the region near the nozzle. Solutions for the velocity and temperature profiles in compressible, turbulent jets have been much less numerous than the solutions for the incompressible case.

Kleinstein [12] presented solutions for the velocity and temperature profiles in the fully developed region of a compressible, turbulent free jet. He applied von Mises transformation to the boundary layer equations and used a Taylor series expansion to linearised the equations. Kleinstein incorporated in his solution a new formulation for the eddy viscosity of a compressible free jet. The
solution agreed well with experimental data [9] for the axial decay of velocity and temperature. However, radial distributions of velocity and temperature were not shown or compared with experimental data. Pai [16] solved the boundary layer equations in the form of von Mises variables by using a finite difference technique. This solution was restricted to the case of a free jet exhausting into a moving external stream. No comparison has been made between the solution and the experimental data, probably due to the lack of data for this type of flow.

Warren [31] solved for the axial variation of velocity and temperature in a compressible turbulent free jet exhausting into a media at rest using an integral method. He used universal profiles to describe the radial variation of axial velocity in the potential core region and the fully developed region. He also proposed a formulation for the eddy viscosity of a compressible free jet. The solution, which was restricted to turbulent Prandtl number of one, agreed fairly well with experimental data except for the transition region.

Szablewski [27] developed an analysis of the core region of a compressible free jet. He used a similarity parameter and numerical integration of the boundary layer equations to obtain velocity and temperature profiles, but the solution was restricted to the potential core region.

Tomich and Weger's [29] modified Kleistein's eddy transfer coefficient for the core region and the fully developed region, and solved the boundary layer equations by a finite difference technique. The solution agreed farily well with experimental data $[18,29]$, however, their correlations for the eddy transfer coefficients and the length of the core region do not apply for lower velocities (usual plume velocity is between $20 \mathrm{ft} / \mathrm{sec}$ to $40 \mathrm{ft} / \mathrm{sec}$ ), and no buoyance effect has been taken into account in their boundary layer equations. Other investigators have also treated various aspects of the compressible jet problem [28,24].
C. Mathematical Modeling and Numerical Solution

The solution of the boundary layer equations of conservation of mass, momentum and energy was obtained by using a variation of the finite difference method due to Abbott [1]. No transformations are made; solutions are obtained for free jet injection into a medium at rest where initial free jet velocity and temperature profiles are assumed to be step functions. Dynamic eddy transfer coefficients of momentum, $\rho \varepsilon_{v^{\prime}}$ and heat flux, $k$, are defined for the turbulent compressible flow. These transfer coefficients are modifications of those introduced by Kleinstein [16] , and Tomich and Weger [15], and are functions of axial position only.

The finite difference technique can also be used to describe jets exhausting into a uniform external stream by changing the boundary conditions, jets with initial profiles of velocity and temperature of an arbitrary shape, and jets with variable fluid properties. Also, concentration profiles can be obtained as more is learned about the structure of free turbulent shear flows, and this finite difference technique can accommodate more advanced formulations of the eddy transfer coefficients which may be functions of both axial and radial position.

A schematic diagram of the free jet showing the coordinate system used is shown in Figure 2-2. The equations were derived from the conservation of mass, momentum, and energy in the turbulent, compressible, axially symmetric steady state free jet.

Conservation of mass (continuity):

$$
\begin{equation*}
\frac{\partial}{\partial z}\left(\rho u_{z}\right)+\frac{l \partial}{r \partial r}\left(r \rho u_{r}\right)=0 \tag{2-1}
\end{equation*}
$$

Conservation of momentum:

$$
\begin{equation*}
\rho u_{z} \frac{\partial u_{z}}{\partial z}+\rho u_{r} \frac{\partial u_{z}}{\partial r}=\frac{1}{r} \frac{\partial}{\partial r}\left(r \rho \varepsilon_{v} \frac{\partial u_{z}}{\partial z}\right)+\left(\rho_{a}-\rho\right) g \tag{2-2}
\end{equation*}
$$

Conservation of energy:

$$
\rho u_{z} c_{p} \frac{\partial t}{\partial z}+\rho u_{r} c_{p} \frac{\partial t}{\partial r}=\frac{1}{r} \frac{\partial}{\partial r}\left(r k \frac{\partial t}{\partial r}\right)+\rho \varepsilon_{v}\left(\frac{\partial u_{z}}{\partial r}\right)^{2}
$$

The heat capacity and the static pressure are assumed to be constant in the above equations. The assumption of constant static pressure has been shown to be a valid assumption for subsonic free jet [8]. All densities, velocities and temperatures have been time-smoothed for turbulent flow. The boundary conditions for this free jet flow are:

$$
\begin{array}{rlr}
\text { at } \begin{array}{rlr}
z & =0, & \\
u_{z} & =u_{0} & \\
t & =t_{0} & \text { for } r<\frac{D}{z} \\
u_{r} & =0 & \\
u_{z} & =0 &
\end{array} \\
t & \text { for } r \geq \frac{D}{2} \\
u_{r} & =0 &
\end{array}
$$

at $r=0$,

$$
\frac{\partial u_{z}}{\partial r}=0
$$

$$
\begin{aligned}
& \frac{\partial t}{\partial r}=0 \quad \text { for all } z \\
& u_{r}=0
\end{aligned}
$$

at $r \rightarrow \infty$,

$$
u_{z}=0
$$

$$
t=t_{a} \quad \text { for all } z
$$

$$
u_{r}=0
$$

Making Equations (1) - (3) dimensionless, they become:

$$
\begin{equation*}
\frac{\partial}{\partial Z}\left(P U_{Z}\right)+\frac{1}{R} \frac{1}{\partial R}\left(R P U_{R}\right)=0 \tag{2-4}
\end{equation*}
$$

$$
P U_{Z} \frac{\partial U_{Z}}{\partial Z}+P U_{R} \frac{\partial U_{Z}}{\partial R}=\frac{1}{u_{0} D} \frac{1}{R} \cdot \frac{\partial}{\partial R}\left(R P \varepsilon_{v} \frac{\partial U_{Z}}{\partial R}\right)
$$

$$
\begin{equation*}
+\frac{\mathrm{Dg}}{u_{o}^{2}}\left[\mathrm{P}_{\mathrm{z}}-\mathrm{P}\right] \tag{2-5}
\end{equation*}
$$

$$
P U_{z} \frac{\partial T}{\partial Z}+P U_{R} \frac{\partial T}{\partial R}=\frac{1}{\rho_{0} u_{o} D} \frac{I}{R} \frac{\partial}{\partial R}\left(R \frac{k}{c_{p}} \frac{\partial T}{\partial R}\right)
$$

$$
\begin{equation*}
+\frac{u_{o}}{D\left(t_{o}-t_{a}\right)_{C_{p}}} P \varepsilon_{v}\left(\frac{\partial U_{z}}{\partial R}\right)^{2} \tag{2-6}
\end{equation*}
$$

The transformed boundary conditions are:
at $Z=0$,

$$
\begin{array}{ll}
\mathrm{U}_{\mathrm{Z}}=1.0 \\
\mathrm{~T} & =1.0 \quad \text { for } \mathrm{R}<\frac{1}{2} \\
\mathrm{U}_{\mathrm{R}}=0 .
\end{array}
$$

$$
\mathrm{U}_{\mathrm{Z}}=0
$$

$$
T=0
$$

$$
\text { for } R \geq \frac{1}{2}
$$

$$
\mathrm{U}_{\mathrm{R}}=0
$$

at $R=0$,

$$
\begin{aligned}
& \frac{\partial U_{Z}}{\partial R}=0 \\
& \frac{\partial \Gamma}{\partial R}=0 \\
& U_{R}=0
\end{aligned}
$$

at $R \rightarrow \infty$,

$$
\begin{aligned}
\mathrm{U}_{\mathrm{Z}} & =0 \\
\mathrm{~T} & =0 \\
\mathrm{U}_{\mathrm{R}} & =0
\end{aligned}
$$

For incompressible jets, expressions for the eddy kinematic viscosity have been obtained semi-empirically in terms of parameters of the main jet flow using either Prandtl's constant exchange coefficient theory, von Karman's hypothesis, or Taylor's vorticity theory. Solutions incorporating these expressions have agreed well with experimental data. Therefore, the expressions for eddy kinematic viscosity derived using these theories have become generally accepted for the solution of incompressible free jet flows.

For compressible free jet flows, however the mechanism of the transport phenomena has not been semi-empirically determined, and few empirical formulations have resulted in solutions which compare well with experimental data. Kleinstein [12] has stated that a "dynamic eddy transfer coefficient,' $\rho \varepsilon_{v}$, must be used as a measure of the transport phenomena for compressible flows. Furthermore, Kleinstein, using an empirical approach to deduce this result, states that this transfer coefficient is a function of axial position only. Kleinstein's momentum transfer coefficient is as follows:

$$
\begin{equation*}
\rho \varepsilon_{v}=0.0183\left(\rho_{0} \rho_{a}\right)^{1 / 2} u_{o} \frac{D}{2} \tag{2-7}
\end{equation*}
$$

Tomich and Weger [29] modified Kleinstein eddy transfer coefficient to provide a better agreement with experimental data:

$$
\begin{equation*}
P \varepsilon_{v}=F(M) P_{a}^{1 / 2} P_{m} u_{o} \operatorname{Dfv}(Z) \tag{2-8}
\end{equation*}
$$

where $F(M)=0.00954-0.00782 M+0.00325 \mathrm{M}^{2}$

$$
(M=\text { Mach number })
$$

$$
\begin{array}{rlrl}
f_{V}(Z) & =0.2 & & \text { for } Z \leq Z_{c v} \\
& =1.0 \quad \text { for } Z>Z_{c v} \\
Z_{C V} & =4.73 \quad \mathrm{P}_{\mathrm{a}}^{-1 / 2} \tag{2-10}
\end{array}
$$

And they also suggested that if the turbulent Prandtl number is only a function of axial position, the eddy thermal conductivity may be expressed as

$$
\begin{equation*}
k=F(M) C_{p} P_{a}^{1 / 2} P_{m} \rho_{o} u_{o} D f_{t}(Z) / N_{p_{r}} \tag{2-11}
\end{equation*}
$$

where

$$
\begin{array}{rlrl}
\mathrm{f}_{\mathrm{t}}(\mathrm{Z}) & =0.2 & \text { for } \mathrm{Z} \leq \mathrm{Z}_{\mathrm{ct}} \\
& =1.0 & \text { for } \mathrm{Z} \geq \mathrm{Z}_{\mathrm{ct}} \\
\mathrm{Z}_{\mathrm{ct}} & =3.43 \quad \mathrm{P}_{\mathrm{a}}^{1 / 2}  \tag{2-12}\\
& \\
\mathrm{~N}_{\mathrm{p}_{\mathrm{r}}} & =1.0 & & \text { for } \mathrm{Z} \leq \mathrm{Z}_{\mathrm{ct}} \\
& =0.715 & \text { for } \mathrm{Z}>\mathrm{Z}_{\mathrm{ct}}
\end{array}
$$

The extent of the core regions for velocity and temperature $\left(Z_{C V}\right.$ and $Z_{c t}$, respectively) were defined by Kleinstein, but it was verified by Tomich and Weger that Kleinstein transfer coefficients are much too large in the core regions which he did not treat in his analysis. They introduced the functions $f_{V}(z)$ and $f_{t}(z)$, which modify the transfer coefficient for the fully developed region to fit the core regions. Also the turbulent Prandtl number in the temperature core region is unity, indicating equal turbulent transport rates of momentum and energy. The turbulent Prandtl number for the remainder flow field has been found to be roughly constant and very nearly equal to the laminar Prandtl number [9,12]. $F(M)$ was the best value found for each experimental Mach number [9,29], and a least squares method was used to obtain $F(M)$ as a polynomial in Mach number.

Substituting Equations (2-8) and (2-11) into Equations $(2-5)$ and $(2-6)$, we have

$$
\begin{align*}
P U_{Z} \frac{\partial U_{Z}}{\partial Z}+P U_{R} \frac{\partial U_{R}}{\partial R} & =A_{V} \frac{1}{R} \frac{\partial}{\partial R}\left(R \frac{\partial U_{Z}}{\partial R}\right) \\
& +\frac{D_{g}}{u_{o}^{2}}\left[P_{A}-P\right] \tag{2-13}
\end{align*}
$$

where

$$
\begin{align*}
& A_{v}=\frac{P^{\varepsilon} v_{v}}{u_{o} D}=F(M) P_{a}^{l / 2} P_{m} f_{v}(Z)  \tag{2-14}\\
& P U_{Z} \frac{\partial T}{\partial Z}+P U_{R} \frac{\partial T}{\partial R}=A_{t} \frac{1}{R} \frac{\partial}{\partial R}\left(R \frac{\partial T}{\partial R}\right) \\
&+N_{E_{C}} A_{v} \frac{\partial U_{Z}}{\partial R^{2}} \tag{2-15}
\end{align*}
$$

where,

$$
\begin{align*}
& A_{t}=\frac{k}{c_{p} \rho_{o} u_{o} D}=F(M) P_{a}^{1 / 2} P_{m} f_{t}(Z) / N_{P_{r}}  \tag{2-16}\\
& N_{E_{c}}=\frac{u_{o}^{2}}{c_{p}\left(t_{o}-t_{a}\right)} \tag{2-17}
\end{align*}
$$

Differentiating Equations (2-4), (2-13), and (2-15), the final form of the equations used in finite difference technique are obtained,

$$
\begin{equation*}
\underline{U_{Z}} \frac{\partial P}{\partial Z}+\underline{P} \frac{\partial U_{Z}}{\partial Z}+\underline{U_{R}} \frac{\partial P}{\partial R}+\underline{P} \frac{\partial U_{R}}{\partial R}+\frac{P U_{R}}{\underline{R}}=0 \tag{2-18}
\end{equation*}
$$

The finite-difference solution used here is a variation of the fully implicit method of Abbott [15] for constant temperature, axially symmetric flows. The following finite difference approximations for derivatives at ( $Z+\Delta Z, R$ ) were used:

$$
\begin{align*}
& \frac{\partial U_{Z}}{\partial Z}=\frac{U_{Z}(Z+\Delta Z, R)-U_{Z}(Z, R)}{\Delta Z}  \tag{2-21}\\
& \frac{\partial U_{Z}}{\partial R}=\frac{U_{Z}(Z+\Delta Z, R+\Delta R)-U_{Z}(Z+\Delta Z, R-\Delta R)}{2 \Delta R} \tag{2-22}
\end{align*}
$$

$$
\frac{\partial^{2} U_{Z}}{\partial R^{2}}=\frac{U_{Z}(Z+\Delta Z, R+\Delta R)-2 U_{Z}(Z+\Delta Z, R)+U_{Z}(Z+\Delta Z, R-\Delta R)}{\Delta R^{2}}
$$

$$
\begin{align*}
& \frac{P U_{i}}{} \frac{\partial U_{Z}}{\partial Z}+\frac{P U_{R}}{\partial U_{Z}}=A_{v} \frac{\partial^{2} U_{Z}}{\partial R^{2}}+\frac{1}{R} \frac{\partial U_{Z}}{\partial R} \\
& +\frac{D g}{u_{o}}{ }^{2}\left[P_{A}-P\right]  \tag{2-19}\\
& \underline{P U_{Z}} \frac{\partial T}{\partial Z}+\underset{R}{P U_{R}} \frac{\partial T}{\partial R}=A_{t} \frac{\partial^{2} T}{\partial R^{2}}+\frac{1}{R} \frac{\partial T}{\partial R} \\
& +N_{E c} A_{V}\left(\frac{\partial U_{Z}}{\partial R}\right)^{2} \tag{2-20}
\end{align*}
$$

Similar expressions were used for the derivatives of $U_{R}$, $T$ and $P$. Another radial derivative form, which is used in the continuity equation in order to make the application of the boundary conditions easier is

$$
\begin{equation*}
\frac{\partial U_{R}}{\partial R}=\frac{U_{R}(Z+\Delta Z, R)-U_{R}(Z+\Delta Z, R-\Delta R)}{\Delta R} \tag{2-24}
\end{equation*}
$$

The finite difference approximations are substituted into Equations (2-18), (2-19) and (2-20), which along with Equations (2-14), (2-16), and (2-17), and the boundary conditions, comprise the system of nonlinear algebraic equations to be solved at each point of the finite-difference net work; see Figure 2-3. The system of algebraic equations was linearlized by taking the coefficients (shown as the underlined coefficients in Equations (2-18), (2-19) and (2-20)) of the nonlinear lines at their known value on the previous line of the net work or at $Z$. The resulting system of simultaneous, linear algebraic equations was solved by using Gaussian elimination method [13] for initial values of $U_{Z}, U_{R}$ and $T$ at each point of the line $Z+\Delta Z$. The computational accuracy was then improved with an iterative procedure until the deviations of the assumed and the newly calculated was less than $5 \%$ on $U_{Z}$ and $T$, and the solution is carried downstream to $Z+2 \Delta Z$, etc. A procedure developed by Wegstein [32]:


Figure 2-3 The Finite Difference Network

$$
\begin{equation*}
\bar{U}_{\mathrm{Z}}(\text { new })=0.33 \mathrm{U}_{\mathrm{Z}}(\mathrm{old})+0.67 \mathrm{U}_{\mathrm{Z}}(\text { new }) \tag{2-25}
\end{equation*}
$$

was used to accelerate the iterative process.
Near $\mathrm{Z}=0$, a smaller step size was required to converge to the correction solution than was required further downstream where the gradients are much smaller. The step sizes were increased downstream in order to speed up the numerical solution. A more detail description on the development of finite-difference equations and program set up are given in Appendix A. The computer program and the results are given in Appendix B.

## CHAPTER III

## AXIAL TEMPERATURE AND VELOCITY DECAY

## A. Analysis and Correlations

The results from the finite-difference method using Tomich and Weger's correlation $P \varepsilon_{v}, k, Z_{c V}$ and $Z_{c t}, ~ u s i n g$ Corrsin and Uberoi, and Tomich and Weger's experimental data are shown in Figures 3-1, 3-2, 3-3 and 3-4. For short distance the temperature and velocity decay at a shorter distance, as compared to the experimental data, and for long distance, the decay of temperature and velocity are about the same, which implies that the core lenghts $z_{c t}$ and $Z_{c v}$ should be adjusted.

New values of $Z_{c t}$ and $Z_{c v}$ were determined by a trial and error fit of experimental data. Beyond these core regions, the decay of temperature and velocity begins to drop much faster as observed from the experimental data.

In order to fit Mahagaokar's experimental temperature data, the values of $F(M)$ had to be adjusted and were determined by trial and error. The values of $Z_{c t}$ were also determined by trial and error. Since no velocity data were presented, the values of $Z_{c v}$ were determined using a factor of 1.28 of $Z_{c t}$, which is the average value of $Z_{c t} / Z_{c v}$, and the highest deviation is only 0.12 for the four sets of


Figure 3-1. Comparision of Experimental Data [9] with Analysis for Axial Decay of Temperature and Velocity


Figure 3-2. Comparision of Experimental Data [9] with Analysis for Axial Decay of Temperature and Velocity
© , $\Delta$ Experimental Data from Refernce [29]
$u_{0}=822 \mathrm{ft} / \mathrm{sec} . \quad \mathrm{t}_{\mathrm{o}} / \mathrm{t}_{\mathrm{a}}=1.05 \mathrm{D}=1 / 4^{\prime \prime}$


Figure 3-3. Comparision of Experimental Data [29] with Analysis for Axial Decay of Temperature and Velocity


Figure 3-4. Comparision of Experimental Data [29] with Analysis for Axial Decay of Temperature and Velocity
experimental data. The axial temperature and velocity decay, and the average tamperature of the "field of view," as calculated from the radial and axial temperature distributions obtained from the finite difference program, are shown in Figures 3-5 and 3-6.

Also from Mahagoakar's axial temperature fluctuation data, Figure 3-7 and 3-8, the temperature fluctuations start to rise approximately at $Z=1.2$, where the values of $Z_{\text {ct }}$ we determine are 1.4 and 1.3 , respectively. From the studies of turbulent fluctuations in pipe flow [4,21], the axial velocity and mass fluctuations are about 0.020 and 0.025 , respectively; Mahagoakar's temperature fluctuations in the core region are about 0.017. It can be concluded that the flow behavior of the emerging plume in the core region primarily resembles its behavior at the point of emergence. Also from the temperature fluctuation data, we can distinguish the three flow regions of the emerging plume as shown in Figures 3-7 and 3-8.

The values of $Z_{c t}, Z_{c v}$ and $F(M)$ determined in the present work as compared with those from Tomich and Weger's correlation are listed in Table 3-1. Correlations were derived for $F(M)$, $Z_{c t}$ and $Z_{c v}$ using the results obtained in this work. A least squares procedure was used to correlate $F(M)$ :

$$
\begin{equation*}
F(M)=0.0168-0.155 M+0.388 M^{2}-0.267 M^{3} \tag{3-1}
\end{equation*}
$$

> Average Temperature Data of Field of View [14] $u_{0}=37.3 \mathrm{ft} / \mathrm{sec} t_{0} / t_{a}=1.75 \mathrm{D}=3^{\prime \prime} 4^{\prime \prime}$

$\triangle$ Average Temperature Data of Field of View [14] $u_{0}=35.6 \mathrm{ft} / \mathrm{sec} \quad t_{0} / \mathrm{t}_{\mathrm{a}}=1.74 \mathrm{D}=3^{\prime} 4^{\prime \prime}$


Figure 3-6. Comparision of Experimental data [14] with Analysis for Axial Decay of Temperature and Velocity



Table 3-1 Values of $Z_{c t}, Z_{c v}$, and $F\left({ }^{(1)}\right.$

| $\begin{gathered} u_{\mathrm{zo}} \\ \mathrm{ft} / \mathrm{sec} \end{gathered}$ | $\frac{t_{0}}{t_{a}}$ | M | Tomich \& Weger's Correlation |  |  | Present Work |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\mathrm{Z}_{\text {ct }}$ | $\mathrm{Z}_{\mathrm{CV}}$ | $F(\mathrm{M})$ | $z_{c t}$ | $\mathrm{z}_{\mathrm{CV}}$ | $\mathrm{Z}_{\mathrm{ct}} / \mathrm{Z}_{\mathrm{cv}}$ | $F(M)$ |
| 822. | 2.68 | 0.755 | 2.10 | 2.89 | 0.00575 | 3.8 | 5.0 | 1. 32 | 0.00575 |
| 687. | 2.72 | 0.631 | 2.08 | 2.37 | 0.00617 | 3.5 | 4.2 | 1.20 | 0.00617 |
| 78.0 | 2.02 | 0.070 | 2.42 | 3.33 | 0.00921 | 3.0 | 3.9 | 1.30 | 0.00921 |
| 78.0 | 1.05 | 0.070 | 3.35 | 4.61 | 0.00921 | 4.2 | 5.4 | 1.29 | 0.00921 |
| 37.3 | 1.75 | 0.0343 | 2.59 | 3.58 | 0.00947 | 1.4 | 1.8 | 1. 28 * | 0.0120 |
| 35.6 | 1.74 | 0.0327 | 2.60 | 3.59 | 0.00948 | 1.3 | 1.7 | 1.28* | 0.0125 |

*This is the average value of $Z_{c t} / Z_{c v}$ for the first four set of data

The average absolute deviation is $5.36 \times 10^{-4}$ and the average percent deviation is 5.03\%.

For the correlations of $Z_{c t}$ and $Z_{c v}$, linear regressions in the forms of

$$
\begin{align*}
& Z_{c t}=A+B\left(\frac{t_{o}}{t_{a}}\right)+C M  \tag{3-2}\\
& Z_{c t}=A\left(\frac{t_{o}}{t_{a}}\right)^{B}(M) C \tag{3-3}
\end{align*}
$$

were tested. The latter form gives a smaller percent of deviation, and the results are as follows:

$$
\begin{equation*}
z_{c t}=14.4\left(\frac{t_{o}}{t_{a}}\right)^{-1.16} \mathrm{~m}^{0.455} \tag{3-4}
\end{equation*}
$$

Avg. \% deviation $=10.7 \%$

$$
\begin{equation*}
\mathrm{z}_{\mathrm{cv}}=17.5\left(\frac{\mathrm{t}_{\mathrm{O}}}{\mathrm{t}_{\mathrm{a}}}\right)^{-1.13} \mathrm{M}^{0.430} \tag{3-5}
\end{equation*}
$$

Avg. \% of deviation $=10.4 \%$

The values of $z_{c t}$ and $Z_{c v}$ are small compared with the total length of axial temperature and velocity decay, so a 10 percent error should be acceptable.

## B. Generalized Equations

As discussed previously, almost all the analytical solutions derived for the plume temperature and velocity decay are based on the similarity hypothesis and the Gaussian distribution of radial temperature and velocity. From the profiles of the experimental axial temperature and velocity decay, these analytical solutions agree well in the fully developed flow region; however, no account has been taken of the core region. A new model is proposed assuming that: 1) the core regions ( $Z_{c t}$ and $Z_{C V}$ ) exist where the exit temperature and velocity along the center line remain unchanged; and 2) in the fully developed region, the behavior follows that of Priestly's widely accepted model. In the latter model the most significant parameter, the spreading coefficient, $\alpha$, was correlated using the experimental data.

The value of $Z_{c t}$ and $Z_{c v}$ are given by the correlations derived previously. The analytical solution for the fully developed region is derived from the equations of conservation of mass, momentum and energy. They are the same equations as $(2-1),(2-2)$ and (2-3), except in the energy equation, the viscous energy dissipation term, $\rho \varepsilon_{v} \frac{\partial u_{z}}{\partial r} \frac{\partial u}{2}$ is neglected, and the shear stress term $\quad \rho \varepsilon_{v} \frac{Z_{r}}{\partial r}$ is replaced by $\tau$ to simplify the mathematics. The-equations then were integrated from $r=0$ to $x=\infty$ using the following boundary conditions,
at $r=0$,

$$
\frac{\partial u_{z}}{\partial r}=0
$$

$$
\frac{\partial t}{\partial r}=0
$$

for all $z$

$$
u_{r}=0
$$

at $r \rightarrow \infty$,

$$
\begin{aligned}
& u_{z}=0 \\
& t=t_{a}
\end{aligned}
$$

$$
\dot{u}_{r}=0
$$

$$
\frac{\partial t}{\partial r}=0
$$

for all z

$$
\frac{\partial u_{z}}{\partial r}=0
$$

$$
\frac{\partial u_{r}}{\partial r}=0
$$

which imply that at a sufficient radial distance there will be no momentum transfer and heat flow. The equations become,

$$
\begin{align*}
& \frac{d}{d z} \int_{0}^{\infty} r u_{z}^{2} \rho d r=\int_{0}^{\infty} r \frac{\Delta t}{t_{a}} \rho g d r  \tag{3-6}\\
& \frac{d}{d z} \int_{0}^{\infty} \frac{1}{2} r u_{z}^{3} \rho d r=\int_{0}^{\infty} r u_{z} \frac{\Delta t}{t_{a}} \rho g d r \\
& -\int_{0}^{\infty} r \tau \frac{\partial u_{z}}{\partial r} d r  \tag{3-7}\\
& \frac{d}{d z} \int_{0}^{\infty} r u_{z} \Delta t \rho d r=-\int_{0}^{\infty} r u_{z} \rho \frac{\partial t_{a}}{\partial z} d r \tag{3-8}
\end{align*}
$$

It is assumed that the atmospheric condition is isothermal, i.e., $\frac{\partial t_{a}}{\partial z}=0$, the shearing stress is a quadratic function of the relative velocity, i.e.,

$$
\begin{equation*}
\tau=\frac{l}{2} \rho u_{z m}^{2} j\left(\frac{r}{R}\right) \tag{3-9}
\end{equation*}
$$

and the similarity hypothesis is good for the radial profiles of $\Delta t$ and $u_{z}$. Following the argument of Sutton [26] that the profiles are Gaussian and the measures of dispersion are approximately at least the same for $\Delta t$ and $u_{z}$, for which experimental confirmation in the laboratory had been provided by Rouse et al. [22] and by Railston [20], $\Delta t$ and $u_{z}$ are taking as

$$
\begin{align*}
& \Delta t=\Delta t_{m} \exp \left(-\frac{r^{2}}{2 R_{m}^{2}}\right)  \tag{3-10}\\
& u_{z}=u_{z m} \exp \left(-\frac{r^{2}}{2 R_{m}^{2}}\right) \tag{3-11}
\end{align*}
$$

Substituting Equations (3-9), (3-10) and (3-11) into (3-6), (3-7) and (3-8), and then integrating gives,

$$
\begin{align*}
& \frac{d}{d z}\left(R_{m}^{2} u_{z m}^{2}\right)=R_{m}^{2} \frac{\Delta t_{m}}{t_{a}} g  \tag{3-12}\\
& \frac{d}{d z}\left(R_{m}^{2} u_{z m}^{3}\right)=3 R_{m}^{2} u_{z m} \frac{\Delta t_{m}}{t_{a}} g-\alpha R_{m} u_{z m}^{3}  \tag{3-13}\\
& \frac{d}{d z}\left(R_{m}^{2} u_{z m} \Delta t_{m}\right)=0 \tag{3-14}
\end{align*}
$$

$U_{z m}, \Delta t_{m}$ and $R_{m}$ are then obtained by an integrating factor method. The final solutions are as follows:

$$
\begin{equation*}
R_{m}=\alpha z^{\prime} \tag{3-15}
\end{equation*}
$$

$$
\begin{align*}
& \Delta t_{m}=\frac{A_{1}}{\alpha^{2} z^{\prime 2}}\left(\frac{3}{2} \frac{A_{1} g}{t_{a^{\alpha^{2}}} z^{\prime}}+\frac{A_{2}}{z^{\prime} 3}\right)^{-1 / 3}  \tag{3-16}\\
& u_{z m}=\left(\frac{3}{2} \frac{A_{1} g}{t_{a} \alpha^{2} z^{\prime}}+\frac{A_{2}}{z^{\prime} 3}\right)^{1 / 3} \tag{3-17}
\end{align*}
$$

where

$$
\begin{align*}
& z_{0}=D / 2 \alpha  \tag{3-18}\\
& z^{\prime}=z+z_{o}-z_{c t} \text { for } \Delta t_{m}  \tag{3-19}\\
& z^{\prime}=z+z_{o}-z_{c v} \quad \text { for } u_{z m}  \tag{3-20}\\
& A_{1}=\alpha^{2} z_{o}^{2} u_{o} \Delta t_{o}  \tag{3-21}\\
& A_{2}=z_{o}^{3}\left[u_{o}^{3}-\frac{3 A_{1}}{2 t_{a}^{2} z_{o}}\right] \tag{3-22}
\end{align*}
$$

A more detail description for the development of the analytical solution for $\Delta t_{m}, u_{z m}$ and $R_{m}$ is shown in Appendix C.

The linear spreading of the plume is derived as a necessary property of the solution and $\alpha$ is regarded as a spreading coefficient. The origin for $z^{\prime}$ is taken as the point of concurrence obtained by extending the plume outline below the level of the fully developed region, $z_{c t}$ or $z_{c v}$. $A_{1}$ and $A_{2}$ are the constants of integration determined from the initial or exit conditions of the plume.

Substituting Equation $(3-18), z_{o}=D / 2 \alpha$ into Equations (3-19) to (3-22), we have

$$
\begin{equation*}
z^{\prime}=\frac{\left(z-z_{c t}\right) \alpha+D / 2}{\alpha} \text { for } t_{m} \tag{3-23}
\end{equation*}
$$

$$
\begin{align*}
& z^{\prime}=\frac{\left(z-z_{C V}\right) \alpha+D / 2}{\alpha} \text { for } u_{m}  \tag{3-24}\\
& A_{1}=\frac{D^{2}}{4} u_{o} \Delta t_{o}  \tag{3-25}\\
& A_{2}=\frac{D^{3}}{8 \alpha^{3}}\left[u_{o}^{3}-\frac{3 D u_{o} t_{o} g}{4 t_{a} \alpha}\right] \tag{3-26}
\end{align*}
$$

Substituting Equations (3-23) to (3-26) into (3-16) and (3-17), $t_{m}$ and $u_{z m}$ can be expressed as a function of $\alpha$ and $z$, the vertical distance from the real source. After some arrangement the following form is obtained

$$
\begin{gather*}
\frac{M_{1}^{3}}{\Delta t_{m}^{3}}=\frac{M_{2}\left[D / 2+\left(z-z_{c t}\right) \alpha\right]^{5}}{\alpha}+M_{3} M_{4}\left[\frac{D}{2}+\left(z-z_{c t}\right) \alpha\right]^{3} \\
-\frac{M_{3} M_{5}\left[\frac{D}{2}+\left(z-z_{c t}\right) \alpha\right]^{3}}{\alpha}  \tag{3-27}\\
u_{z m}^{3}=\frac{M_{2}}{\alpha\left[\left(z-z_{c v}\right) \alpha+\frac{D}{2}\right]}+\frac{M_{3}}{\left[\left(z-z_{c v}\right) \alpha+\frac{D}{2}\right]} \\
x\left(M_{4}-\frac{M_{5}}{\alpha}\right) \tag{3-28a}
\end{gather*}
$$

where

$$
\begin{aligned}
& M_{1}=\frac{D^{2}}{4} u_{o} \Delta t_{o} \\
& M_{2}=\frac{3 D^{2} u_{o} \Delta t_{o} g}{8 t_{a}} \\
& M_{3}=\frac{D^{3}}{8} \\
& M_{4}=u_{o}^{3} \\
& M_{5}=\frac{3 D u_{0} \Delta t_{o} g}{4 t_{a}}
\end{aligned}
$$

Equations (3-27) and (3-28) are now in the form

$$
\begin{equation*}
Y=f(z, \alpha) \tag{3-29}
\end{equation*}
$$

and Scarborough's [30] non-linear least squares method can be used to obtain the best value of $\alpha$ from a given set of values of $z$ and $Y$. Following Scarborough, the residuals can be written as

$$
\begin{align*}
& D_{1}=\alpha^{\prime}\left(\frac{\partial \mathrm{f}_{\mathrm{i}_{1}}}{\partial \alpha}{ }_{0}+\mathrm{Y}_{1}^{\prime}-Y_{1}\right.  \tag{3-30}\\
& \mathrm{D}_{\mathrm{n}}=\alpha^{\prime}\left(\frac{\partial \mathrm{f}_{\mathrm{n}}}{\partial \alpha}{ }_{0}+\mathrm{Y}_{\mathrm{n}}^{\prime}-\mathrm{Y}_{\mathrm{n}}\right.
\end{align*}
$$

$Y_{i}$ is the experimental data point at $z_{i}$ and $Y^{\prime}{ }_{i}$ is the calculated value of $Y$ obtained from Equation (3-27) or (3-28)
using $z_{i}$ and an estimate of $\alpha$ denoted by $\alpha_{0}$. The partial derivatives are valued at $\alpha=\alpha_{0}$, and the corrected value of $\alpha$ is given by

$$
\begin{equation*}
\alpha=\alpha_{0}+\alpha^{\prime} \tag{3-31}
\end{equation*}
$$

from the condition for least squares,

$$
\sum \mathrm{D}_{\mathrm{i}}{ }^{2}=\min
$$

or

$$
\frac{\partial}{\partial \alpha^{1}} \sum_{i}^{n} D_{i}^{2}=0
$$

which gives

$$
\begin{equation*}
2 \sum_{i} D_{i} \frac{\partial D_{i}}{\partial \alpha^{r}}=0 \tag{3-32}
\end{equation*}
$$

Equations (3-29) to (3-32) are used to obtain the following solution for $\alpha$ :

$$
\begin{equation*}
\alpha=\alpha_{0}+\frac{\sum_{i}\left(Y_{i}^{\prime}-Y_{i}\right)\left(\frac{\partial f_{i}}{\partial \alpha}\right)_{0}}{2} \tag{3-33}
\end{equation*}
$$

A computer program was set up to solve $\alpha$ for $\Delta t_{m}$ and $\mathrm{u}_{\mathrm{zm}}$ using the experimental data given in Tables 2-1, 2-3, and 2-5. An iterative procedure was used until the absolute difference of the new $\alpha$ and the old $\alpha$ is less than $5 \%$ of the new $\alpha$. The computer program and the results are given in Appendix D.

Since both temperature and velocity data were used for two different equations, and a few assumptions have been made on the radial temperature and velocity profiles as discussed, different values of $\alpha$ for temperature and velocity should be expected and they are denoted by $\alpha_{t}$ and $\alpha_{v}$, respectively. The values of $\alpha_{t}$ and $\alpha_{v}$ calculated from the computer program are summarized in Table 3-2, and the values of $\Delta t_{m} / \Delta t_{o}$ are $u_{z m} / u_{z o}$ versus vertical distance are shown in Figures 3-1 to 3-6. Fairly good agreement is observed.

In order to predict the axial decay of temperature and velocity, correlations of $\alpha_{t}$ and $\alpha_{v}$ as a function of $t_{o} / t_{a}$ and $M$ were made by linear programming. The results are as follows:

$$
\alpha_{t}=0.0422\left(\frac{t_{\mathrm{O}}}{t_{\mathrm{a}}}\right)^{0.405}(\mathrm{M})^{-0.273}
$$

$$
\text { Avg. } \% \text { deviation }=5.6 \%
$$

Table 3-2 Values of $\alpha$ from Scarborough's Non-Iinear Least Square Method

| $\begin{aligned} & \mathrm{u} \\ & \mathrm{zt} / \mathrm{sec} \end{aligned}$ | $\frac{t_{0}}{t_{a}}$ | M | Z ct | $\alpha_{t}$ | Avg. \% Deviation | $\mathrm{Z}_{\mathrm{CV}}$ | $\alpha_{v}$ | Avg. $\%$ Deviation |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 822. | 2.68 | 0.755 | 3.8 | 0.070 | 5.2 | 5.0 | 0.082 | 5.5 |
| 687. | 2.72 | 0.631 | 3.5 | 0.073 | 4.8 | 4.2 | 0.085 | 5.7 |
| 78.0 | 2.02 | 0.070 | 3.0 | 0.097 | 4.9 | 3.9 | 0.105 | 6.5 |
| 78.0 | 1.05 | 0.070 | 4.2 | 0.088 | 3.2 | 5.4 | 0.092 | 4.8 |
| 37.3 | 1.75 | 0.0343 | 1.4 | 0.141 | 5.4 | 1.8 | 0.145 | 4.5 |
| 35.5 | 1.74 | 0.0343 | 1.3 | 0.147 | 4.4 | 1.8 | 0.150 | 4.3 |

$$
\alpha_{v}=0.0486\left(\frac{t_{0}}{t_{a}}\right)^{0.435}(\mathrm{M})^{-0.236}
$$

Avg. \% deviation $=7.3 \%$

The final form of the generalized equations for temperature and velocity decay are as follows:

$$
\begin{aligned}
& \Delta t_{m}=\Delta t_{0} \quad \text { for } z \leq z_{c t} \\
& \Delta t_{m}=\frac{A_{1}}{\alpha_{t}{ }^{2} z^{\prime 2}}\left(\frac{3}{2} \frac{A_{1} g}{t_{a^{\alpha} t^{2} z^{\prime}}}+\frac{A_{2}}{z^{\prime 3}}\right)^{-1 / 3} \text { for } z>z_{c t} \\
& u_{z m}=u_{z O} \quad \text { for } z \leq z_{c V} \\
& u_{z m}=\left(\frac{3}{2} \frac{A_{1} g}{t_{a} \alpha_{v}{ }^{2} z^{\prime}}+\frac{A_{2}}{z^{\prime 3}}\right)^{1 / 3} \quad \text { for } z>z_{c t} \\
& z_{c t}=14.4\left(\frac{t_{0}}{t_{a}}\right)^{-1.16}(\mathrm{M})^{0.455} \\
& z_{C V}=17.5\left(\frac{t_{O}}{t_{a}}\right)^{-1.13} \quad \text { (M) } 0.430 \quad D \\
& \alpha_{t}=0.0422\left(\frac{t_{0}}{t_{a}}\right) 0.405 \text { (M) }-0.273 \\
& \alpha_{v}=.0486\left(\frac{t_{0}}{\left(t_{a}\right.}\right) 0.435 \quad \text { (M) }-0.236
\end{aligned}
$$

$$
\begin{aligned}
& z_{o t}=D / 2 \alpha_{t} \\
& z_{o v}=D / 2 \alpha_{v} \\
& z^{\prime}=z+z_{o t}-z_{c t} \quad \text { for } \Delta t_{m} \\
& z^{\prime}=z+z_{o v}-z_{c v} \quad \text { for } u_{z m} \\
& A_{1}=\frac{D^{2}}{4} u_{z o} \Delta t_{o} \\
& A_{2}=z_{o t}^{3}\left(u_{z o}^{3}-\frac{3 A_{1} g}{2 t_{a}^{2} t_{o t}}\right) \\
& A_{2}=z_{o v}^{3}\left(u_{z o}^{3}-\frac{3 A_{1} g}{2 t_{a}{ }^{2} v_{o v}}\right)
\end{aligned}
$$

## CHAPTER IV

## RADIAI TEMPERATURE AND VELOCITY DISTRIBUTION

A. Radial Distribution of Temperature and Vertical Velocity The radial profiles of temperature and vertical velocity calculated by the finite difference program agrees well with Corrsin and Uberoi [9] data as shown in Figure 4-1 and 4-2, but do not agree so well with Tomich and Weger's data, as shown in Figure 4-3 and 4-4.

The radial profiles of temperature and velocity at short distances from the stack exit are shown in Figure 4-5. The radial profiles change gradually from a flat profile to the Gaussian shape of profiles. The edges of the profile are smoothing out in the Gaussian shape.

A comparison of the experimental radial average temperature of the field of view with that calculated from finite difference method is shown in Figure 4-6. The points near the center gives better agreement than the two points near the edges.

An attempt was made to correlate the radial profiles of temperature and velocity obtained from the literature and from the finite difference method using the similarity hypothesis as proposed by many pioneers $[20,22,11,24] ;$ i.e.,
© © Experimental Data [9]


Figure 4-1. Comparision of the Experimental Data [9] with the Analysis for the Radial Distribution of Temperature and Velocity


Figure 4-2. Comparision of the Experimental Data [9] with the Analysis for the Radial Distribution of Temperature and Velocity





Figure 4-6. Comparision of the Experimental Data [14] with the Analysis for the Radial Distribution of Temperature

$$
\begin{align*}
& \frac{\Delta t}{\Delta t_{m}}=\exp \left(-\beta \frac{r^{2}}{z^{2}}\right)  \tag{4-1}\\
& \frac{u_{z}}{u_{z m}}=\exp \left(-\eta \frac{r^{2}}{z^{2}}\right) \tag{4-2}
\end{align*}
$$

where $\beta$ and $\eta$ are constants. However, $\beta$ and $\eta$ were found to be approximately constant only when $z$ is constant, and $\beta$ and $\eta$ increase as $z$ increases. That is to say the radial distribution of temperature and velocity are Gaussian shape, however, the similarity hypothesis for the radial distributions doesnot hold for compressible free jet. The average value ${ }_{\lambda}$ of $\beta$ and $\eta$, and their average percent deviations at various
 simulated value from the finite difference method, are given in Table 4-1. The values of $\beta$ and $\eta$ are then correlated with the dimensionless variables $t_{o} / t_{a}, M$ (Mack number) and $z$ by the linear regression method; the results are as follows:

$$
\begin{equation*}
\beta=23.14 z^{0.680} M^{0.229}\left(\frac{t_{o}}{t_{a}}\right)^{-0.491} \tag{4-3}
\end{equation*}
$$

Avg. \% deviation $=7.15$

$$
\begin{equation*}
n=15.33 \mathrm{z}^{0.688} \mathrm{~m}^{0.0857}\left(\frac{t_{\mathrm{o}}}{t_{\mathrm{a}}}\right)^{-0.329} \tag{4-4}
\end{equation*}
$$

Avg. \% deviation $=6.89$

Table 4-1 Variation of $\beta$ and $\eta$ with $u_{z o}, t_{a} / t_{o}$, and $z$

| $\begin{gathered} u_{z o} \\ \mathrm{ft} / \mathrm{sec} \end{gathered}$ | $\frac{t_{0}}{t_{a}}$ | $z / D$ | $n$ | Avg. \% deviation | $\beta$ | Avg. \% deviation | $\frac{\eta}{\beta}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 37.3 | 1.73 | 4 | 28.9 | 7.1 | 32.9 | 5.0 | 1.14 |
| 37.3 | 1.73 | 11 | 45.0 | 5.5 | 48.4 | 4.9 | 1.08 |
| 37.3 | 1.73 | 15 | 56.2 | 4.7 | 60.0 | 4.2 | 1.07 |
| 78.0 | 2.02 | 15 | 52.5 | 3.8 | 82.5 | 7.6 | 1. 57 |
| 78.0 | 1.05 | 15 | 74.0 | 6.5 | 103.7 | 6.3 | 1. 39 |
| 687. | 2.72 | 11 | 64.8 | 7.8 | 89.4 | 7.4 | 1.37 |
| 822. | 2.68 | 4 | 22.0 | 7.1 | 38.0 | 5.0 | 1.72 |
| 822. | 2.68 | 11 | 57.0 | 8.6 | 102.0 | 7.9 | 1.77 |

It is obvious from Equations (4-3) and (4-4) that the values of $\beta$ and $\eta$ increases as initial velocity increases and as the temperature ratio $\frac{t_{0}}{t_{a}}$ decreases.

Also it is noticed that the ratio of $\eta / \beta$ is greater than 1, that is to say the characteristic radius describing the temperature distribution in a heated free jet is greater than that for the velocity distribution. Rouse et al. and Railston [22,20] have reported the value of $\frac{\eta}{\beta} \simeq 1$; Reichart's [19] experimental value for $\frac{\eta}{\beta}=2$, while our calculated value is between 1 and 2. The value of $\eta / \beta$ increases as the exit velocity increases as observed from Table 4-1.

## B. Distribution of Radial Velocity

One advantage of the present calculational method is that the calculated results for the radial distribution of the radial velocity component can also be obtained.

A radial distribution of the calculated radial velocity at various vertical position is shown in Figure 4-7. The positive radial velocity near the jet axis is thought to be caused by the spread of the jet as it moves downstream, while the negative (inward) radial velocity at larger radial distance is thought to be caused by the entrainment of ambient air. The magnitude of the inward radial velocity is decreasing as $z$ increases, that is to say the entrainment rate decreases as $z$ increases, or as vertical velocity decreases.

C. Bibliography for Chapters I, II, III and IV

1. Abbott, M.R., Computer J., 7, 47 (1946).
2. Abramovich, G.N., Turbulent Jets Theory, 278-316, ASITA, Translation AD283858 (1962).
3. Batchelor, G.K., Quart. J. Roy. Meterol. Soc., 80, 339 (1954).
4. Becker, H.A., Rosenweig, R.E., and Gwozdg, J.R., AIChE J., 12, 964 (1966).
5. Briggs, G.A., Plume Rise, U.S.A.E.C., Division of Technical Information (1969).
6. Bringfelt, B., Atmospheric Environment, 2, 575-598 (1968).
7. Carnahan, B., Luther, H.A., and Wilkes, J.O., Applied Numerical Methods, 440-442, John Wiley \& Sons, New York (1969).
8. Cohen, N.S. Rept. TM 1,3 RO, Aerojet-General Corp., Sacramento, Calif. (19964).
9. Corrsin, S. and Uberoi, M.S., N.A.C.A., Rep. No. 998 (1950).
10. Davis, O.D., Smith, G., and Klauber, G., Science, 186, 732 (1974).
11. Hinze, J.O., and B.G. van der Hegge Zignen, Applied Science Research 1A, 435 (1949).
12. Kleinstein, G., J. Spacecraft Rockets, l, 403 (1964).
13. Lapidus, Leon, Digital Computation for Chemical Engineers, 254-255, McGraw-Hill, New York (1962).
14. Mahagaokar, M., Ph.D. Dissertation, Chemical Engr. Dept., Uni. of Houston, Houston, Texas.
15. Morton, B.R., Taylor, G.E. and Turner, J.S., Proc. Roy. Soc., London, Ser. A., 234, 1 (1956).
16. Pai, S.I., Quart. Appl. Math., 10, 141 (1952).
17. Prengle, H.W., J., Morgan, C.A., Fang, C.S., Huang, L.K., Campani, P., and Wu, W.W., Environ. Sci. Tech 7, 417 (1973).
18. Priestly, C.H.B. and Ball, F.K., Quart. J. Roy. Meteorol. Soc., 81, 144 (1955).
19. Reichardt, H., Gesetzmakigkeiten der freien Turbulenz. VDI-Forschungsheft, 414, (1942); 2nd ed. 1951.
20. Railston, W., Proc. Phys. Soc., B., 67, 42 (1954).
21. Rosenweig, R.E., Hottel, H.C., and William, G.C., Chem. Engr. Sci. 15, 111 (1961).
22. Rouse, H., Yih, C.S., and Humphreys, H.W., Tellus, $\underline{4}^{\prime}$ 201 (1952).
23. Scarborough, J.B., Numerical Mathematical Analysis, Johns Hopkins Press (1962).
24. Schlichting, H., Boundary Layer Theory, 483-506, McGraw-Hill, New York (1955).
25. Schmidt, w., Z. Angew, Math. Mech. 2l, 265 (1941).
26. Sutton, O.G., Micrometeorology, 10-12, McGraw-Hill, New York (1953).
27. Szablewski, W., Intern. J. Heat Mass Transfer, 6, 739 (1963).
28. Tennessee Valley Authority, Full Scale Study of Plume Rise at Large Electric Generating Stations, Muscle Shoals, Alabama.
29. Tomich, J.F., and Weger, E., AIChE J. 30, 948 (1967).
30. Tollmun, W., Z. Agnew, Math. U. Mech., 6 , 468 (1926).
31. Warren, N.R., Ph.D. Thesis, Princeton Uni., N.J. (1957).
32. Wegstein, J.G., Commun. Assoc. Computing Mach., l, 9 (1958).

DISCUSSIONS, CONCLUSIONS, AND RECOMMENDATIONS

## A. Discussions and Conclusions

1. Finite-Difference Analysis: Starting with pipe flow conditions as intial conditions for the jet, ab initio velocity and temperature calculations were made axially and radially using a finite-difference technique (similar to Tomich and Weger), which included the modified transfer coefficient given in the literature. These calculations agree well with the experimental data for axial decay and radial distribution of temperature and velocity, and indicate the existence of a core region in which the center line properties remain unchanged; and a fully developed region in which radial and axial properties decay. The technique has eliminated the need for some of the simplifying assumptions that were required in previous investigations in order to obtain a solution. Also the finitedifference technique can be set up so that a different form of initial profiles and more complex forms of eddy transfer coefficient can be inserted when the accuracy of the available data is justified. Another advantage of this calculational method is that the distributions of the radial component of velocity are also obtained.

Conclusion : The finite-difference calculational technique has made possible a very general solution of the
conservation equations for a compressible turbulent, axially symmetric free jet.
2. The Radial Distribution of Temperature and Velocity:

The experimental results from laboratory free jets and from our infrared remote sensing measurements, and the analytical results from the finite-difference calculations show that the radial distribution of temperature and velocity is approximately Gaussian in the fully developed region . The finite-difference calculations show that the radial profiles change gradually from an initial flat to Gaussian shape. The Characteristic radius describing the temperature distribution in a hot free jet is greater than that for the velocity distribution, and both increase as the exit velocity increases; therefore similarity does not hold for hot compressible free jets.

Conclusion: The radial distribution of temperature and velocity in a hot free jet or emerging plume is approximately Gaussian in the fully developed region. And from the difference in the characteristic radius of the temperature and velocity distributions, the mechanisms for heat and momentum transfer are not the same.
3. The Plume Model: An analytical model is proposed which includes, i) core region, and ii) fully developed region solutions for the axial decay of temperature and velocity. The model contains a correlated quantity, $\alpha$ the spreading coefficient, whcih is a function of the plume Mach number and temperature ratio, $t_{o} / t_{a}$, the two most important parameters.

The model agrees well with experimental data; the largest deviation occuring in the very short transition region. Deviations can be reduced to smooth the curve in this region. The axial temperature and velovity were found to decay in a shorter distance as the temperature ratio is increased and the velocity decreased.

Coefficients for the radial distribution of temperature and velocity were correlated as function of Mach number, temperature ratio, $t_{o} / t_{a}$, and the dimensionless axial distance 2/D.

Conclusion: It is possible to construct a plume model which provides better agreement with data for the axial decay of temperature and velocity than previous models.
B. Recommendations

1. The correlation coefficients obtained in this work are based on a relatively few set of experimental data. More data at velocities between $20-40 \mathrm{ft} / \mathrm{sec}$ should be obtained to justify and improve the correlations.
2. In particular to the remote sensing of temperature measurement by infrared emission interferometry-spectroscopy technique, a more sensitive detector, e.g., the mercury cadmium telluride ( Hg Cd Te ) photodetector should be used, the field of view can then be reduced, more accurate axial and' radial temperature measurements can then be made. This improvement also provides better concentration fata.
3. So far no mass dispersion have been discussed, a turbulent transfer coefficient for concentration may be
introduced to develop a new model for mass dispersion. However, this requires experimental data to justify the theoretical model.
4. No attempt has been made to modify the finite-difference program to fit the cross wind situation. Certain boundary conditions, the eddy transfer coefficients and the schematic structure of the plume should be changed to accompany for the cross wind situations.
5. The finite-difference technique, although it is a powerful technique in solving differential equations, but may not be the best available as far as computer time is concerned, "collocation" method should be faster than the finite-difference method, and collocation may have better convergence or more stable system because it requires a smaller matrix. Collocation method should be considered in future use.

## CHAPTER VI

## EMISSIONS AS A FUCTION OF EXCESS AIR

As industrial boilers and process heaters are operated at lower excess air levels in order to improve combustion efficiency, the concentration of of pollutants in stack gases will change with flame temperature and combustion rate. The purpose of this chapter is a)- to present data from the literature on gas, oil, and coal fired furnaces indicating the levels of pollutant concentrations, b)- to discuss emissions as a function of excess air, and c)- discuss recent equipment improvements which permit operation at lower excess air levels.

## A. Combustion of Gas, Oil and Coal

Natural Gas Combustion
Natural gas is used as fuel extensively in power plants, industrial heating, and for domestic and commercial space heating. The primary component is methane, but with smaller quantities of other hydrocarbon, nitrogen and carbon dioxide. Even though natural gas is considered to relatively clean, some emissions do occur from the combustion reaction. When insufficient air is supplied, large amounts of carbon monoxide may be produced.

Emissions factors for natural gas combustion are presented in Table 6.1 [1].

## Fuel Oil Combustion

Fuel oil is classified into two major types, residual and distllate. istillate fuel oil is primarily for domestic use, but is used in some commercial and industrial applications where a high quantity oil is required. Fuel oils are classified by grades: No. 1 to No. 6 from high sulfur content.

Emissions from oil combustion are dependent on type and size of equipment, method of firing, and maintenance. Table $6.2^{[1]}$ presents emission factors for

Table 6.1-Emission Factors for Natural Gas Combustion ${ }^{(1)}$

| Pollutant | Type of Unit |  |  |
| :---: | :---: | :---: | :---: |
|  | Power Plant | Industrial Process Boiler | Domestic and Commercial Heating |
|  | $1 \mathrm{~b} / 10^{6} \mathrm{ft}{ }^{3}$ | $1 \mathrm{~b} / 10^{6} \mathrm{ft}{ }^{3}$ | $1 \mathrm{~b} / 10^{6} \mathrm{ft}^{3}$ |
| Particulates | 15 | 18 | 19 |
| Sulfur-oxides $\left(\mathrm{SO}_{2}\right)$ | 0.6 | 0.6 | 0.6 |
| Carbon Monoxide (C0) | 17 | 17 | 20 |
| Hydrocarbons ( $\mathrm{CH}_{4}$ ) | 1 | 3 | 8 |
| Nitrogen Oxides ( $\mathrm{NO}_{2}$ ) | 600 | (120 to 130) | (80 to 120) |

Table 6.2 - Emission Factors for Fuel Oil Combustion ${ }^{(1)}$

| Pollutant | Type of Unit |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | Power Plant | Residual | Distillate | Domestic |
|  | $1 \mathrm{~b} / 10^{3} \mathrm{gal}$ | $1 \mathrm{~b} / 10^{3} \mathrm{gaI}$ | $1 \mathrm{~b} / 10^{3} \mathrm{gal}$ | $1 \mathrm{~b} / 10^{3} \mathrm{gal}$ |
| Particulate | 8 | 23 | 15 | 10 |
| Sulfur dioxide | 157 | 157 | 142 | 142 |
| Sulfur trioxide | 2 | 2 | 2 | 2 |
| Carbon monoxide | 3 | 4 | 4 | 5 |
| Hydrocarbons | 2 | 3 | 3 | 3 |
| Nitrogen oxides | $\left(\mathrm{NO}_{2}\right) 105$ | (40 to 80) | (40 to 80 ) | 12 |
| Aldehydes ( HCHO ) | 1 | 1 | 2 | 2 |

fuel oil combustion. Note that the industrial and commercial category is split into residual and distillate because there is a significant difference in particulate emissions from the same equipment. It should also be noted that power plants emit less particulate matter per quantity of oil consumed, reportedly because of better design and more precise operation of equipment.

## Coal Combustion.

Coal, the most abundant fossil fuel in the USA is burned in a wide variety of furnaces to produce steam and process heat; some large pulverized coal-fired units burn 300-400 tons/hr.

Although predominantly carbon, coal contains many compounds in varying amounts. The exact nature and quantity of these compounds are determined by the location of the mine producing the coal and will usually affect the final use of the coal.

Emission from coal is dependent upon the contents of coal, the type of burner, the property and control of the burning process. The emission factors for bituminous coal and anthracite coal combustion are presented in Tables 6.3 and 6.4.

ERA Standard
In late $1971^{(2)}$ EPA established the following limits on fossil fuel boilers having a heat input greater than $250 \times 10^{6} \mathrm{Btu} / \mathrm{hr}$.

| Emission | Fuel | Lb/10 ${ }^{6}$ Btu <br> Input | Approx.ppm <br> (volumetric dry) |
| :--- | :--- | :--- | :--- |
| Particulates | All | 0.1 | 0.06 grans/scf) |
| $\mathrm{SO}_{2}$ | Liquid | 0.8 | 550 |
|  | Solid | 1.2 | 520 |
| $\mathrm{NO}_{\mathrm{x}}$ | Gaseous | 0.2 | 165 |
|  | Liquid | 0.3 | 227 |
|  | Solid | 0.7 | 525 |

For carbon monoxide CO, E.P.A. set the ambient air quality standard at a maximum of $10^{\mathrm{mg}} / \mathrm{m}^{3}$ ( 9 ppm ) for 8 -hour averaging period, and $40 \mathrm{mg} / \mathrm{m}^{3}$ ( 35 ppm ) for a 1 -hour averaging period. (3)

Table 6.3-Emission Factors for Bituminous Coal
Combustion without control Equipment ${ }^{(1)}$

| Furna | Particulates | Sulfur oxides | Carbon monoxide | Hydrocarbons | Nitrogen oxides | Aldehydes |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{aligned} & \text { size } \\ & 10^{6} \mathrm{Btu} / \mathrm{hr} \\ & \text { heat input } \end{aligned}$ | $\begin{aligned} & \text { 1b/ton } \\ & \text { coal } \\ & \text { burned } \end{aligned}$ | 1b/ton coal burned | 1b/ton coal burned | 1b/ton coal burned | $\begin{aligned} & \text { lb/ton } \\ & \text { coal } \\ & \text { burned } \end{aligned}$ | 1b/ton coal <br> burned |

Greater than 100

| Pulverized |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| General | 16 | 38 | 1 | 0.3 | 18 | 0.005 |
| Wet bottom | 13 | 38 | 1 | 0.3 | 30 | 0.005 |
| $\begin{aligned} & \text { Dry } \\ & \text { bottom } \end{aligned}$ | 17 | 38 | 1 | 0.3 | 18 | 0.005 |
| Cyclone | 2 | 38 | 1 | 0.3 | 55 | 0.005 |
| 10 to 100 |  |  |  |  |  |  |
| Spreader stoker | 13 | 38 | 2 | 1 | 15 | 0.005 |
| Less than 10 |  |  |  |  |  |  |
| Spreader stoker | 2 | 38 | 10 | 3 | 6 | 0.005 |
| Handfired units | 20 | 38 | 90 | 20 | 3 | 0.005 |

Table 6.4 - Emissions from Anthracite Coal
Combustion Without Control Equipment ${ }^{(1)}$

| Type of | $\begin{aligned} & \text { Particu- } \\ & \text { late } \\ & \hline \end{aligned}$ | Sulfur dioxide | Sulfur trioxide | Hydrocarbons | Carbon monoxide | Nitrogen oxides |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| furnace | lb/ton | 1b/ton | 1b/ton | 1b/ton | 1b/ton | 1b/ton |

Pulverized

| (dry bottom), <br> no fly-ash <br> reinjection | 17 | 38 | 0.5 | 0.03 | 1 | 18 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Overfeed stokers, no fly-

2
38
0.5
0.2
(2 to 10 )
(6 to 15 )
ash reinjection

Hand-fired units 10

36
0.8
2.5

90
3

## B. Variations with Excess Air

Particulates are defined as dispersed solid and liquid particles larger than single molecules ( $\sim 2 \times 10^{-4}$ microns in diameter) but smaller than 500 microns.

For natural gas particulate emissions are not important, and not much specific data are available in the literature, but it has been shown that increasing the
mount of excess air decreases the particulate loading since more complete combustion results. ${ }^{(1)}$

For fuel oil the normal range of particulate loading is between 0.025 and $0.060 \mathrm{gr} . / \mathrm{scf}.{ }^{(4)}$ The most commonly reported values are between 0.030 and 0.035 $\mathrm{gr} / \mathrm{scf}$. ${ }^{(4)}$ In a series of four tests, it was found that, as the oxygen concentration in the stack gas increased from 2 to 4 percent, the particulate loading decreased from 0.140 to $0.020 \mathrm{gr} / \mathrm{scf}^{(5)}$

For coal particulates consist primarily of carbon, silica, alumina and iron oxide in fly ash. The quantity is mainly a function of l) the ash content, 2) the heating value of the fuel, 3) the method by which the coal is burned, and (4) the rate of burning. No specific data relates the particulates concentration as a function of excess air is available in the literature. However, the particulates emission decreases as excess air increases because of more complete combustion. The effect of excess air on the amount of unburned combustibles is shown in Figure 6.1.


Figure 6.1 - Effect of Excess Air on Unburned Combustibles (6)

Carbon Monoxide and unburned fuel
Insufficient supply of air, poor mixing of air and fuel, low furnace temperature, and insufficient combustion time produce large amounts of carbon monoxide and hydrocarbon. Smoke formation is the characteristic feature of incomplete combustion. There are very little literature data discussed
extensively on the effect of excess of oxygen on carbon monoxide and unburned fuel.

However, carbon monoxide and unburned fuel concentrations drop rapidly as excess air is increased from $0 \%$, (4) (6)(7)(8) at about $3 \%$ to $4 \%$ excess oxygen, combustion approaches $100 \%$ completion. A number of test runs ${ }^{(7)}$ on a power house burning sulfur free natural gas showed that C0 dropped from 540 to less than $10 \mathrm{ppm} . \mathrm{CH}_{4}$ from 750 to less than 60 ppm , HCHO from 260 to less than 130 ppm , for an increase of excess oxygen from $0 \%$ to $4 \%$. The effect of excess oxygen on the concentration of the pollutants is shown in Figure 6.2.

Another set of data obtained from a steam generating coal-pulverized unit is shown in Figure 6.3; for C0, hydrocarbons, and formaldehyde for excess oxygen from $2 \%$ to $4.5 \%$.

Oxides of Sulfur
The amount of sulfur oxide is mainly dependent on the sulfur content of the fuel; effect of excess air on sulfur oxide emission is minimal. Emission can be lowered by reducing the sulfur content or installing an $\mathrm{SO}_{2}$ clean up system-a wet scrubbing or a dry system.

Oxides of Nitrogen, N0

The amount of $\mathrm{NO}_{\mathrm{x}}$ formed is a function of several variables including, nitrogen content of the fuel, flame temperature, and oxygen availablility.

A substantial portion of the ${ }^{N}{ }_{x}$ formed in combustion comes from the nitrogen in the combustion air--called thermal $\mathrm{NO}_{x}$; which is especially temperature sensitive. Figure $6,4^{(4)}$ shows this effect which occurs when some of the $N_{2}$ in the air forms atomic nitrogen and combines with oxygen to form NO and some $\mathrm{NO}_{2}$.

Emission of $\mathrm{NO}_{x}$ from combustion of natural gas is indicated to increase linearly as a function of excess air, as shown in Figure 6.2,6.5,6.6, and 6.7 for burning natural gas and propane. (10) It is estimated that as excess $0_{2}$ increased $1 \%$, the $\mathrm{NO}_{x}$ concentration increased $15 \%$. Above $5 \%$ excess $0_{2}$, $\mathrm{NO}_{x}$ concentration increases very slowly or remains constant.

For fuel oil the amount of excess air used in large modern plant is about $16 \%$ to $20 \%$ or 4 to $5 \%$ excess oxygen. This concentration varies with fuel and


FIGURE 6.2 EFFECT OF EXCESS AIR ON POLLUTANT
EMISSION FROM COMBUSTION OF NATURAL GAS ${ }^{(7)}$


FIGURE 6.3 EFFECT OF EXCESS AIR ON POLLUTANT EMISSIONS FROM A PULVERIZED COAL UNIT ${ }^{(8)}$


FLAME TEMPERATURE, ${ }^{\circ} \mathrm{F}$
FIGURE 6.4-THEORETICAL NO vs. FLAME TEMPERATURE, ${ }^{\circ} F^{(4)}$.


FIGURE 6.5 NOx vs. PERCENT OXYGEN IN FLUE GAS ${ }^{(10)}$


Figure 6-6. ${ }^{N} O_{x}$ Versus Excess $O_{2}$ For A Balanced Draft Heater With Combustion Air PreHeat [16].


Figure 6.7 NO $\mathrm{NO}_{\mathrm{x}}$ Versus Excess $0_{2}$ For A Forced Draft Heater [16]
burner design. A linear relationship has beenestablished indicating that as excess $0_{2}$ decreased by $1 \%$, the $\mathrm{NO}_{\mathrm{x}}$ concentration is reduced by $15 \%$ for tangentially fired units, and $29 \%$ for a horizontally fired unit [11]. Figure 6.6, 6.7 , and 6.8 shows the effect of excess air on $\mathrm{NO}_{x}$ emissions from two draft heaters and a large unit.

For coal, emissions of $\mathrm{NO}_{x}$ is mainly dependent on the nitrogen content and the combustion process. The overall conversion oc coal nitrogen to $\mathrm{NO}_{x}$ in a gasification/combustion process was found to be significantly lower than for direct burning of pulverised coal [12]. In Figure 6.9 [12] NO $\mathrm{N}_{\mathrm{x}}$ emissions from a diffusion flame are nomalized with respect to furnace heat input and oresented as function of furnace oxygen and measured amounts of $\mathrm{NH}_{3}$ in the fuel gas. The conversion of $\mathrm{NH}_{3}$ to $\mathrm{NO}_{\mathrm{x}}$ contributed between 60 to 180 ppm to the total emission level. Emission of $\mathrm{NO}_{\mathrm{x}}$ from different coal combustion for five front wall fired burners using the $B \& W$ coal spreader and for 4 tangentially fired burners are shown in Figure 6.10 and 6.11 [17]. For front wall-fired units, the average increase of $\mathrm{NO}_{x}$ emission is 16 ppm for $1 \%$ increase of excess air, and for tangentially fired unit is 13 ppm .
C. Equipment Developments

Burners: New 'high efficiency' burners have been designed by manufac turers to operate at low excess air levels, by improving fuel and air distribution, and shorter flame design to minimize loss of sensible heat and unburned fuel concentration. The LoNox burner, [10] a combustion oil and gas and radiant wall burner, utilizes both a tertiary air inlet in the primary tile and a special automatic recirculation oil tile. This special oil tile recirculates combustibles and combustion products back into flame region, lowering $\mathrm{NO}_{\mathrm{x}}$ production.

The HB and HEVD burners [13] have been designed to inspirate and mix a large percentage of the total combustion air. Controls are provided for both primary and secondary air adjustments.

Register burners [14] utilize the principle of admitting and controlling the swirl of combustion air through adjustable louvers; which are arranged so that all combustion air is divided into two concentric oppsitely rotating air streams. This provides violent mixing, and as the two air streams cancel rotational velocity, the result is a short, narrow flame. Register burners can also be insulated for preheated air, where used on a pressurized furnace;


FIGURE 6.8-NO $x^{-}$EMISSIONS vs. EXCESS OXYGEN ${ }^{(11)}$ (OIL FIRED UNIT)


FIGURE 6.9-NO ${ }_{x}$ EMISSION vs. EXCESS OXYGEN ${ }^{(12)}$ (FOR DIFFUSION FLAMES COAL FIRED UNIT)


special peepholes, torch loop and other equipment are available to protect the operator blasts.

Furnaces and Boilers - Large furnaces and boilers are being designed by manufacturers to satisfy the demand for lower cost power. As units were extrapolated to larger sizes certain parameters were held constant: the velocity of the combustion gases in the furnace, the temperature of the flue gas entering the convection surface, and the time required to completely burn the fuel. As the surface to volume ratio in the active combustion zone decreased the peak and average flame temperature increased, and increased $\mathrm{NO}_{\mathrm{x}}$ formation. In addition, burners that are designed for high turbulence and rapid burning rates to assure complete and stable utilization of the fuel over the widest load range possible result in high $\mathrm{NO}_{\mathrm{x}}$ generation. Two stage combustion and flue gas recirculation, as illustrated by Figure 6.12 has been found to be an effective method to reduce $\mathrm{NO}_{x}$. Figures 6.13 and 6.14 indicate how $\mathrm{NO}_{x}$ is reduced by these methods, for gas and oil fuels.
D. Conclusions.

The material presented in this report indicates that excess air (oxygen) plays an important role in energy savings and emission of certain pollutants.

To achieve high combustion efficiency and comply with current enviromental standards, equipment manufacturers have started to improve the design of burners furnaces and boilers. Sucessful combustion devices have been developed for less than $6 \%$ excess air, and some burners can operate down to $1 \%$ excess oxygen. More research and development work is needed to produce hgih efficiency combustion devices which will meet enviromental standards and operate at $0-1 \%$ excess oxygen levels.

Advanced instrumentation and monitoring devices are needed to achieve optimum operating conditions and thereby higher efficiencies.

## E. Literature Cited

(1) U.S. Enviromental Protection Agency: "Compilation of Air Pollutant Emission Factors," PHS No. 999-AP-42 (1973)
(2) Wolfe, W.C., "Control of Atmospheric Emissions from Industrial Boilers," Technical Paper, Babcock \& Wilcox, BR-961-PGTP-72-29 (1972).
(3) Prengle, H.W., Jr., "Process Plant Econdynamics," p.10, Lecture 3, preliminary ed. (1974)


FIGURE6.I2 - SCHEMATIC SHOWING METHOD OF AIR ADMISSION FOR TWO STAGE COMBUSTION PROCESS ${ }^{(15)}$.


FIGURE 6.13-EFFECT OF GAS RECIRCULATION AND TWO STAGE COMBUSTION ON COMBUSTION TEMPERATURE ${ }^{(15)}$.

## 

FIGURE 6.14-EFFECT OF GAS RECIRCULATION AND TWO STAGE COMBUSTION OF NO FORMATION ${ }^{(15)}$.
(4) Smith, W.S., "Atmospheric Emission from Fuel Oil Combustion," An Inventory Guide. U.S. P.H.S. 999-AP-2 (1962).
(5) Jefferis, G.C. and Sensenbaugh, J.D., 'Effect of Operating Variables on the Stack Emission from a Modern Power Station Boiler.' ASME (Oct. 22,1959).
(6) Smith, W.S. and Gruba, C.W., "Atmospheric Emission from Coal Combustion," An Inventory Guide. P.H.S. 999-AP-24 (1966).
(7) Mahagaokar, U., Remote Sensing of Temperature and Composition of Gas Plumes by IR Emission Interferometry-Spectroscopy, Ph.D. Dissertation, University of Houston (1976).
(8) Cuffe, S.T., Gerstle, R.W., Orwing, A.A., Schwartz, C.H., "Air Pollutant Emissions from Coal Fired Power Plants," Report No.l, JAPCA 14: 353-362 (Sept. 1964).
(9) Weidman, G.H., Utterback. P.M., "Liquid Fuel Analysis--their effect on Combustion and Emissions," Technical Paper BR-1061, Babcock and Wilcox, (April 1976).
(10) John Zink Company, Brochure: "LoNox Burner" (current).
(11) Mills, J.L.. Leudtke, R.D., Woolrich, P.F., and Perry, L.B., "Emissions of oxides of nitrogen from stationary sources in Los Angeles'; Report 3: L.A. County Air Pollution Control District, (July 1961).
(12) Lisanskas, R.A., Johnson, S.A., ' $N 0$ o Formation during Gas Combustion," Chem. Eng. Prog. 72, 76 (Aug 1976).
(13) John Zink Company, Brochures: Series HB Burner, Series HEVD Burner (current).
(14) Coen Company, Brochure: Coen Register Burners, Burlingame, California (current).
(15) James, D.E., "A Boiler Manufacturers View on Nitric Acid Formation," presented at 5th Technical Meeting, West Coast Section, APCA, Tech. Paper BR-922, TPO74, Babcock and Wilcox (October 1970).
(16) Hunter S.C. and Carter W.A., "Application of Combustion Technology for $\mathrm{NO}_{x}$ Emissions Reduction on Petroleum Process Heaters," AIChE Paper No. 64C P-186, March, 1977.
(17) Brown R.A., Mason H.B., Pershing D.W. and Wendt J.O.L., "Investigation of First and Second Stage Variables On Control of NO ${ }_{x}$ Emissions In A Pulverised Coal Furnace", presented at 83rd National Meeting, AIChE March 24, 1977, Houston, Texas.

## CHAPTER VII

## IMPROVED IRRS SIGNAL PROCESS METHOD

## A. Introduction

The application of remote sensing techniques combining infrared radiometry and spectroscopy, to measure pollutant emissions quantitatively was started in 1968 by Prengle et al. [3]. Briefly, the interferometer spectrometer method consists of receiving, by a telescope, infrared emission signals characteristic of each pollutant species in a hot gas plume. The basic optical element is the Michaelson interferometer with one fixed and one movable mirror which scans the entire infrared region ( $2.0 \mu$ to $25 \mu$ ) and produces an electrical signal from the detector called an interferogram. With adequate digital Fourier transformation of the interferogram a spectrogram of the source is obtained, intensity vs. frequency. Figures $7-1 a$ and $7-1 b$ show the infrared emission spectrum ( $0-3600 \mathrm{~cm}^{-1}$ ) from a furnace burning natural gas operated with zero excess air. The components studied include $\mathrm{CO}, \mathrm{NO}, \mathrm{NO}_{2}, \mathrm{CH}_{4}, \mathrm{C}_{2} \mathrm{H}_{4}, \mathrm{SO}_{2}, \mathrm{H}_{2} \mathrm{~S}, \mathrm{HCHO}, \mathrm{CH}_{3} \mathrm{CHO}$ and $\mathrm{C}_{2} \mathrm{H}_{6}$. In 1976, Mahagaokar [1] developed process spectral data obtained by remote sensing, to identify different species, and to determine the temperature and composition of the remote gas plume.



Special attention has been focused on the CO centration because at the characteristic frequency location there is no interference from $\mathrm{CO}_{2}$ and $\mathrm{H}_{2} \mathrm{O}$, and the well known $\mathrm{P}-\mathrm{R}$ structure spectrogram should be observed. It is the purpose of the present study 1) to reevaluate the model for calculating the concentration of $\mathrm{CO}, 2$ ) to develop a new data processing model for $C O$, and 3) to serve as a general method to improve the data processing of other components.
B. The IR Spectrum of CO

CO is a diatomic molecule which in the gas phase undergoes vibration and rotation about its center of gravity and acts like a "vibrating-rotator ". Figure 7-2 [4] shows the vibrational-rotational spectrum of $C O$ under high resolution. The band is centered at $\bar{v}_{0}=2143 \mathrm{~cm}^{-1}$, and those rotational lines with $\bar{v}<\bar{v}_{o}$ are the $P$ branch while those with $\bar{v}>\bar{v}_{0}$ are the $R$ branch, which gives $C O$ the well know $P-R$ structure. The intensity of the rotational ( $P$ and $R$ ) wings is proportional to the number of molecules undergoing transitions from a particular rotational level, $J$, which the population can be obtained from the Boltzman distribution [5]

$$
\begin{equation*}
\frac{\mathrm{N}_{J}}{\mathrm{~N}}=g_{J} e^{-\mathrm{BJ}(\mathrm{~J}+1) h \mathrm{hc} / \mathrm{kT}} \tag{2-1}
\end{equation*}
$$

The value of $J$ for which equation $(7-1)$ is a maximum, which would be the $J$ at peak of the $P$ and $R$ branches is given by,


Figure 7-2. The Vibration-Rotation Spectrum of Carbon Monoxide under High Resolution [4]

$$
\begin{equation*}
J=\left(\frac{\mathrm{kT}}{2 \mathrm{Bhc}}\right)^{1 / 2}-\frac{1}{2} \tag{7-2}
\end{equation*}
$$

The separation between the two maxima is given by,

$$
\begin{equation*}
\Delta \bar{v}_{\text {max }}=\left(\frac{8 \mathrm{kTB}}{\mathrm{hc}}\right)^{1 / 2}+2 \mathrm{~B} \tag{7-3}
\end{equation*}
$$

A computer program was set up based on Equation (7-1) to simulate the spectrogram of $C O$, and an example generated from the simulation at a $T=500^{\circ} \mathrm{K}$ is shown in Figure 7-3. The line separation, obtained from [4], is 2B $=3.94$. The maximum intensity of the line occurs at $J=9$. The half width of the spectral lines, $\Delta \bar{v}_{01 / 2}$ is given by,

$$
\begin{equation*}
\Delta \bar{v}_{{ }_{0}}=\frac{4 \sigma^{2} \bar{u}_{o} n_{0}}{c} \tag{7-4}
\end{equation*}
$$

where

$$
\begin{aligned}
\sigma & =\text { collision diameter (cm) } \\
& =3.590 \times 10^{-8} \mathrm{~cm} \text { (from [6]) } \\
\bar{u}_{0} & =\text { average kinetic velocity (cm/sec) } \\
& =\frac{8 \mathrm{RT}}{\mathrm{M}} \\
\mathrm{n}_{0} & =\text { number of molecules } / \mathrm{cm}^{3}
\end{aligned}
$$



The value of $\Delta \bar{v}_{o l / 2}$ calculated at $500^{\circ} \mathrm{K}$

$$
\Delta \bar{v}_{01 / 2}=0.227 \mathrm{~cm}^{-}
$$

## C. Equipment Characteristics

Resolution: The maximum retardation $A$ of the interferometer, the greatest optical path difference between the transmitting and reflecting beam of the Michelson interferometer, determines the resolution of the spectrogram, and is given by [7],

$$
\begin{equation*}
\text { Resolution }=\frac{l}{A} \tag{7-5}
\end{equation*}
$$

The maximum retardation operating in the experiment is 0.25 cm ; therefore, the resolution is $4.0 \mathrm{~cm}^{-1}$, which means that two lines cannot be resolved when computed from the above interferogram if their line difference is less than $4.0 \mathrm{~cm}^{-1}$.

Sampling Interval in the Interferogram: Woodward states [8] that any waveform that is a sinusoidal function of time or distance can be digitized unambiguously using a sample frequency equal to twice the band width of the system; the interferogram may then be completely reconstructed without any loss of information or signal-to-noise ratio.

In order to completely reconstruct the interferogram, the sampling interval $\delta$ should be [9],

$$
\begin{align*}
& \delta \leq \frac{1}{2\left(\bar{v}_{\max }-\bar{v}_{\min }\right)}  \tag{7-6a}\\
& \bar{v}_{\max }=\frac{10^{4}}{2.8 \mu} \mathrm{~cm}^{-1}=3571 \mathrm{~cm}^{-1} \\
& \bar{v}_{\min }=\frac{10^{4}}{25 \mu} \mathrm{~cm}^{-1}=400 \mathrm{~cm}^{-1}
\end{align*}
$$

Therefore, $\delta \leq 1.576 \times 10^{-4} \mathrm{~cm}$
It is a characteristic of the equipment to sample at multiples of 6328A; the laser is a HeNe unit which produces a line at $6328 \AA$; therefore, the maximum $\delta$ is,

$$
\delta=2 \times 6328 \AA=1.2656 \times 10^{-4} \mathrm{~cm}
$$

which satisfies Equation (7-6b). Therefore the total sampling points in the interferogram is,

$$
\begin{aligned}
\text { Sampling Points } & =\frac{A}{\delta} \\
& =\frac{0.25 \mathrm{~cm}}{1.2656 \times 10^{-4} \mathrm{~cm}} \\
& =1975 \text { pts. }
\end{aligned}
$$

Information Intervals in the Spectrogram: The spectrogram is obtained by a Fourier transformation of the interferogram, and the Cooley-Tukey algorithm is used for the
transformation. It is a characteristic of the Cooley-Tukey algorithm that the number of data points is $N=2^{n}$, when $n$ is a positive integer. Therefore the minimum number of points should be $N>1975$ to avoid loss of information; therefore, $N=2048$ pts ( $n=11$ ). The intensity of the spectrogram is then stored at an Information Interval =

$$
\frac{1}{(2 \times \delta) \cdot N}=1.929 \mathrm{~cm}^{-1}
$$

## D. CO Concentration from IRRS

The concentration model [I] for a pollutant, $C_{i}$ is given by,

$$
\begin{equation*}
C_{i}(p p m)=\frac{\left(S \bar{v}_{i}\right) \times 10^{6}}{\left(R-G A_{m}\right)\left(T_{a i}{ }^{\beta-}\right)\left(X^{2} \bar{w}_{B i} L_{e} k_{\bar{v}_{i}} \Delta \bar{v}_{i}\right)} \tag{7-7}
\end{equation*}
$$

where $S_{\Delta \bar{\nu}_{i}}$ is the integrated area of the spectrogram of the pollutant $i$ at the assigned wave number range, $\Delta \bar{\nu}_{i}$, or

$$
S_{\Delta \bar{v}_{i}}=\int_{\Delta \bar{\nu}_{i}} s \bar{v}^{-d}
$$

and $S_{\bar{v}}$ is the spectral intensity at $\bar{v} . k_{\bar{v}_{i}}$ is the average absorption coefficient of the pollutant $i$ at $\bar{v}_{i}$. It is important to point out that because of a lack of literature data, Rock's [2] absorption coefficient measurements and
correlations were used,

$$
\begin{equation*}
\frac{1}{\bar{v}} \bar{v}, m_{2}^{2}=\frac{1}{k}_{\bar{v}_{0}}^{2}=\frac{M^{\prime} x^{n}}{\bar{k}_{\bar{v}_{0}}^{A^{\prime}}}+\frac{X^{n}}{A^{\prime}} \tag{7-9}
\end{equation*}
$$

where $k_{\bar{v}, m}$ is the maximum absorption coefficient at low resolution instead of the average value. Also the other parameters in Equation (7-7) relate to the system characteristics, and the surrounding conditions, in particular are not related to the spectral structure of $C O$.

A simulated spectrogram of $\mathrm{CO}, \mathrm{S}_{\bar{v}} \mathrm{vs} . \bar{v}$ is generated, which resembles that obtainable from equipment, i.e., the spectral lines of $C O$ were simulated for a resolution of $4 \mathrm{~cm}^{-1}$, and the resulting intensity $S_{v}$ was stored and printed out at an interval of $1.929 \mathrm{~cm}^{-1}$. The spectrogram was generated in the following way. Consider any two spectral lines in Figure 7-4, at wave number $X(J)$ and $X(J+1)$. The intensity of the lines $f(J)$ and $f(J+I)$ have been stored when the true spectrogram of $C O$ was generated as given by Equation (7-1) or Figure 7-3. The base width of the spectral lines $X_{1}$ to $X_{2}$ is $0.454 \mathrm{~cm}^{-1}$, twice the calculated halfwidth of the spectral lines. The sampling point $X X(N)$ starts at around 1993. $\mathrm{cm}^{-1}$ and samples at an interval of $1.929 \mathrm{~cm}^{-1}$. The intensity $S_{v}$ is equal to the sum of the area (arbitrary units) of the spectral lines (triangles)

Case 1:

```
and XXl<Xl
    XX2<XA1
```



Case 2:
XXI<X1
and
$\mathrm{XX} 2<\mathrm{XA} 2$


Case 3:
XXI>XI
and
XA $2>x$ X2


Case 4:
$\mathrm{xx} 1<\mathrm{x} 2$
and
$X \times 2>X A 2$


Case 5:
$\mathrm{XXI}>\mathrm{X} 2$


Figure 7-4 Simulation of Spectrogram
covered by the rectangle of width $\left(\mathrm{XX}_{1}\right.$ to $\mathrm{XX}_{2}$ ) equal to $4 \mathrm{~cm}^{-1}$, which is equivalent to a resolution of $4 \mathrm{~cm}^{-1}$. Figure 7-4 shows the five different situations for calculating $S_{V} \cdot$. A computer program with an algorithm base on these five situations was set up to generate the spectrogram $S_{v}$ vs. $\bar{v}$. Two generated spectrograms with resolution of $4 \mathrm{~cm}^{-1}$ and print out interval of $1.929 \mathrm{~cm}^{-1}$ at two different starting points $1993.12 \mathrm{~cm}^{-1}$ and $1993.32 \mathrm{~cm}^{-1}$, are shown in Figure $7-5 a$ and $7-5 b$, respectively. By comparison to an actual spectrogram, Figure 7-6, obtained from a run of the experimental equipment, the $P-R$ structure can be visualized, and the ups and downs can be understood because of the equipment resolution and information interval. Also as observed from experimental spectrograms, consistent interferences appeared short of $2090 \mathrm{~cm}^{-1}$ and beyond $2205 \mathrm{~cm}^{-1}$. Figure $7-6$ also shows this effect, which is probably due to the interference of some unburned hydrocarbons in the plume. A new wave number range $2090 \mathrm{~cm}^{-1}$ to $2205 \mathrm{~cm}^{-1}$ is recommended to replace the old range 2060 to $2220 \mathrm{~cm}^{-1}$.

Also, it is observed that the value of the generated $S_{\bar{v}}$ is greater than the true $S_{\bar{v}}$ because of the system resolution; $S_{\Delta \bar{v}}$ is used in calculating the concentration of the pollutants. The true and the generated $S_{\Delta \bar{v}}$ values are calculated and compared; the values, i.e, the area of


Spectral Intensity (Arb. Units)
 Figure 7-6. CO Spectrogram from Experimental Data [1]
the spectrograms in Figure 7-3 and 7-5, are obtained simply by the summation of small rectangles method. For the complete CO spectrogram,

$$
\frac{\text { Generated } S_{\Delta \bar{v}}}{\text { True } S_{\Delta \bar{v}}}=1.11
$$

and for the range $2090 \leftrightarrow 2205 \mathrm{~cm}^{-1}$

$$
\frac{\text { Generated } S_{\Delta \bar{\nu}}}{\text { True } S_{\Delta \bar{v}}}=1.10
$$

The generated $S_{\Delta \bar{\nu}}$ used above is the average of 10 generated $S_{\Delta \bar{v}}$ from 10 different starting points. In order to account for this error, the value of $S_{\Delta \bar{\nu}}$ calculated from experimental data should use a factor of $1 / 1.10=0.909$.

It was pointed out previously that the value of $k_{\bar{v}_{i}}$ used, from Rock's correlation, is the maximum value instead of the average value. The value of $S_{\bar{v}}$ at the maximum, i.e., at $P_{9}$ or $R_{9}$, i.s 1.166 (arbitrary units), the average value of $S_{v}$ from 2090 to $2205 \mathrm{~cm}^{-1}$ (i.e., $P_{1}$ to $P_{14}$ and $R_{0}$ to $R_{16}$ ) is 0.915. As an approximation, the value of $k_{\bar{v}_{i}}$ obtained from Rock's correlation should be multiplied by a factor of $0.915 / 1.116=0.820$, for this particular wave number range in order to obtain a more correct result.

As an example, let's consider Run \#312, the spectrogram of CO is Figure 7-6 and the previously calculated concentration is 915 ppm . The value of:

$$
\begin{aligned}
& \mathrm{S}_{\Delta \bar{v}}\left(2060-2220 \mathrm{~cm}^{-1}\right)=142.72 \text { (arbitrary) } \\
& \Delta \bar{v}=160 \mathrm{~cm}^{-1}
\end{aligned}
$$

and,

$$
\begin{aligned}
& \mathrm{S}_{\Delta \bar{v}}\left(2090-2205 \mathrm{~cm}^{-1}\right)=100.53 \text { (arbitrary) } \\
& \Delta \bar{\nu}=115 \mathrm{~cm}^{-1}
\end{aligned}
$$

Assuming all parameters except $S_{\Delta \bar{v}_{i}}, k_{\bar{v}_{i}}$ and $\Delta \bar{v}_{i}$, do not change in Equation $7-7$ with the new wave number range, the corrected concentration of CO is given by,

$$
\begin{aligned}
C(C O) & =915 \mathrm{ppm} \times \frac{100.33 \times 0.090}{14272} \times \frac{160 \mathrm{~cm}^{-1}}{115 \mathrm{~cm}^{-1}} \times \frac{\mathrm{k}_{\bar{v}_{i}}}{0.820 \mathrm{k} \bar{v}_{i}} \\
& =992 \mathrm{ppm}
\end{aligned}
$$

Therefore, the following conclusions are made:

1. For a resolution of $4 \mathrm{~cm}^{-1}$, the integrated area of the concentration signal, $S_{\Delta \bar{v}}$, due to overlapping, is about 1.1 times higher than the true value.
2. Judging and comparing the theoretical to experimental spectrograms, consistant interferences from other compounds,
most likely unburnt hydrocarbon, appear on the edges of the assigned wave number range. A new range is proposed to minimize this effect.
3. The absorption coefficient used had been the maximum value. An average value is obtained as a function of the maximum value based on the theoretical spectral intensity of the spectral lines.

Finally, in the future $S_{\Delta \vec{v}}$ and $k_{v}$ for $C O$ should be calculated

$$
\begin{aligned}
& \mathrm{S}_{\Delta \bar{v}}=0.909 \mathrm{~S}_{\Delta \bar{v}} \quad \text { (observed) } \\
& \mathrm{k}_{\bar{v}}=0.820 \mathrm{k}_{\bar{v}_{i}} \quad \text { (Rock's value) }
\end{aligned}
$$

## E. Recommendations

1. Since the integrated area of the concentration signal is related to the resolution and the spectral structure of the molecule, the effect on other components should be studied.
2. Besides interferences from the expecting components, other unpredicted interferences may also exist, the spectrogram of the components should be reviewed to minimize the error.
3. The absorption coefficients of the components should be reevaluated.

## F. Symbols \& Nomenclature

## Upper Case

| A | maximum retardation (cm) |
| :---: | :---: |
| $A_{m}$ | area of the receiving mirror of the telescope, $\mathrm{cm}^{2}$ |
| B | average rotational line seperation ( $\mathrm{cm}^{-}$) |
| $C_{i}$ | concentration of component i, ppm |
| G | system gain |
| J | rotational level |
| $L_{e}$ | equivalent path length (cm) |
| N | total number of molecules |
| $\mathrm{N}_{J}$ | number of molecules in the J-th rotational state |
| $\mathrm{R}_{\bar{\nu}}$ | average spectral responsivity at - |
| ${ }^{s} \Delta \bar{v}_{i}$ | integrated area of the spectrogram of pollutant i |
| $S_{v}$ | spectral concentration signal |
| T | absolute temperature |
| $\mathrm{T}_{\text {ai }}$ | average atmospheric transmittance of component i |
| $\bar{W}_{\beta i}$ | average emissive power for the location of component i |
| X | dimensionless radius of the plume relative to the distance |

## Lower Case

| $c$ | speed of light |
| :--- | :--- |
| $g_{J}$ | degeneracy |
| $h$ | Planck's constant |
| $k$ | Boltzmann's constant |
| $k_{\bar{v}_{i}}$ | spectral absortion coefficient for component $i$ |

$\eta_{0} \quad$ number of molecules/unit volume
$\bar{u}_{0} \quad$ average kinetic velocity

## Geek Letters

| $\beta-\bar{v}$ | spectral background radiation factor |
| :--- | :--- |
| $\delta$ | sampling interval |
| $\bar{\nu}, \Delta \bar{\nu}$ | spectral frequency and bandwidth in wave numbers <br> $\left(\mathrm{cm}^{-1}\right)$, respectively |
| $\bar{\nu}_{0}$ | band centre location $\left(\mathrm{cm}^{-1}\right)$ of the co spectrogram |
| $\Delta \bar{\nu}_{\max }$ | the seperation between two maximum intensity <br> spectral lines $\left(\mathrm{cm}^{-1}\right)$ |
| $\Delta \bar{\nu}_{o l / 2}$ | half-width of band $\mathrm{cm}^{-1}$ <br> $\sigma$ |
| collision diameter |  |

## G- Literature Cited

1. Mahagaokar, U., Ph.D Dissertation, Chem. Engr. Dept., Uni. of Houston., Texas, (1976).
2. Rock, K.L., M.S. Thesis, Chem. Engr. Dept., Uni. of Houston, Texas, (1975)
3. Prengle, H.W. Jr., Morgan, C.A., Fang, C.S., Huang, L.K., Campani, P., and Wu,W.W., Enviro. Sci, Technol., Z, 417 (1973).
4. Banwell,C.N., Fundamentals of Molecular Spectroscopy, McGraw-Hill, N.Y. (1966).
5. Curtice, S., Ph.D. Dissertation, ChE. Dept., Uni. of Houston, Texas (1974)
6. Hirschfelder J.O., Curtiss,C.F., and Bird, R.B., Mole$\frac{\text { cular Theory of Gases and Liquids, }}{\text { N.Y. }}$ (1974). N.Y. (1974).
7. Bell R.J., Introductory Fourier Transform Spectroscopy, Academic Press, N.Y. (1974).
8. Woodward, M., Probability and Information Theory, Pergamon Press, N.Y. (1955).
9. Griffiths P.R., Chemical Infrared Fourier Transform Spectroscopy, Wiley, N.Y., (1975).

APPENDIX A

SOLUTION OF FINITE-DEIFFERENCE EQUATIONS

The finite-difference solution used here is a variation of the fully implicit method of Abbott [25] for constant temperature, axially symetric flow system. The equations that have been developed in Chapter II were further developed here such that the equations are readily solved by the Gaussian method of elimination. Equations (2-18), (2-19) and (2-20) are now written as $(A-1),(A-2)$ and $(A-3):$

$$
\begin{aligned}
& U_{z} \frac{\partial P}{\partial z}+P \frac{\partial U_{z}}{\partial Z}+U_{R} \frac{\partial P}{\partial R}+P \frac{\partial U_{R}}{\partial R}+\frac{P U_{R}}{R}=0 \quad(A-1) \\
& P U_{z} \frac{\partial U_{2}}{\partial Z}+P U_{R} \frac{\partial U_{z}}{\partial R}= A_{V}\left[\frac{\partial^{2} U_{z}}{\partial R^{2}}+\frac{1}{R} \frac{\partial U_{z}}{\partial R}\right] \\
&+\frac{D g}{u_{0}^{2}}\left[P_{A}-P\right] \quad(A-2) \\
& P U_{z} \frac{\partial T}{\partial Z}+P U_{R} \frac{\partial T}{\partial R}= A_{T}\left[\frac{\partial^{2} T}{\partial R^{2}}+\frac{1}{R} \frac{\partial T}{\partial R}\right] \\
&+N_{E C} A_{V}\left(\frac{\partial U_{2}}{\partial R}\right)^{2}
\end{aligned}
$$

The following finite-difference approximations for derivatives at $(Z+\Delta Z, R)$ were used:

$$
\begin{aligned}
& \frac{\partial U_{z}}{\partial Z}=\frac{U_{z}(z+\Delta z, R)-U_{z}(z, R)}{\Delta Z} \\
& \frac{\partial U_{z}}{\partial R}=\frac{U_{z}(z+\Delta z, R+\Delta R)-U_{z}(Z+\Delta z, R-\Delta R)}{2 \Delta R} \quad(A-4) \\
& \frac{\partial^{2} U_{z}}{\partial R^{2}}=\frac{U_{z}(z+\Delta z, R+\Delta R)-2 U_{z}(Z+\Delta Z, R)+U_{z}(Z+\Delta z, R \Delta R)}{\Delta R^{2}}
\end{aligned}
$$

$$
\begin{aligned}
& \frac{\partial U_{R}}{\partial R}=\frac{U_{R}(z+\Delta z, R)-U_{R}(z+\Delta z, R-\Delta R)}{\Delta R} \\
& \frac{\partial P}{\partial z}=\frac{P(z+\Delta Z, R)-P(z, R)}{\Delta z} \\
& \frac{\partial P}{\partial R}=\frac{P(z+\Delta z, R+\Delta R)-P(Z+\Delta z, R-\Delta R)}{2 \Delta R} \quad(A-8) \\
& \frac{\partial T}{\partial z}=\frac{T(z+\Delta z, R)-T(Z, R)}{\Delta z} \\
& \frac{\partial T}{\partial R}=\frac{T(z+\Delta z, R+\Delta R)-T(z+\Delta z, R-\Delta R)}{2 \Delta R} \\
& \frac{\partial^{2} T}{\partial R^{2}}=\frac{T(Z+\Delta z, R+\Delta R)-2 T(z+\Delta z, R)+T(z+\Delta z, R-\Delta R)}{\Delta R^{2}}
\end{aligned}
$$

Substituting Equations ( $A-4$ ) to ( $A-12$ ) into Equations $(A-1),(A-2)$, and $(A-3)$, the Equations then are linearlized by taking the coefficients (shown as the underlined coefficients in the equations) of the non-linear terms at their known value on the previous line of the net work or from the iteration. Equations ( $A-1$ ) to ( $A-3$ ) become:

$$
\begin{align*}
& U_{z}(z, R) \frac{P(z+\Delta z, R)-P(z, R)}{\Delta z}+P(z, R) \frac{U_{z}(z+\Delta z R)-U_{z}(z, R)}{\Delta z} \\
+ & U_{R}(z, R) \frac{P(z+\Delta z, R+\Delta R)-P(z+\Delta z, R-\Delta R)}{2 \Delta R} \\
+ & P(z, R) \frac{U_{R}(z+\Delta z, R), U_{R}(z+\Delta z, R-\Delta R)}{\Delta R} \\
+ & \frac{P(z, R) \cdot U_{R}(z, R)}{R}=0 \quad(A-13) \tag{A-13}
\end{align*}
$$

$$
\begin{aligned}
& P(Z, R) U_{z}(\dot{Z}, R) \frac{U_{z}(Z+\Delta Z, R)-U_{z}(Z, R)}{\Delta z} \\
& +P(z, R) U_{R}(z, R) \frac{U_{z}(z+\Delta z, R+\Delta R)-U_{z}(z+\Delta z, R-\Delta R)}{2 \Delta R} \\
& =A v\left[\frac{u_{z}(z+\Delta z, R+\Delta R)-2 u_{z}(z+\Delta Z, R)+u_{z}(z+\Delta z, R-\Delta P)}{\Delta R^{2}}\right. \\
& \left.+\frac{1}{R} \frac{U_{z}(z+\Delta z, R+\Delta R)-U_{z}(z+\Delta z, R-\Delta R)}{2 \Delta R}\right] \\
& +\frac{D g}{u_{0}^{2}}\left[P_{A}-P(z, R)\right] \\
& \text { (A-14) } \\
& P(Z, R) U_{z}(Z, R) \frac{T(Z+\Delta Z, R)-T(Z, R)}{\Delta z} \\
& +P(Z, R) U_{R}(Z, R) \frac{T(Z+\Delta z, R+\Delta R)-T(Z+\Delta Z, R-\Delta R)}{2 \Delta R} \\
& =A_{T}\left[\frac{T(z+\Delta z, R+\Delta R)-2 T(z+\Delta z, R)+T(z+\Delta z, R-\Delta R)}{\Delta R^{2}}\right. \\
& \left.+\frac{1}{R} \frac{T(Z+\Delta Z, R+\Delta R)-T(Z+\Delta Z, R-\Delta R)}{2 \Delta R}\right] \\
& +N_{E_{c}} A_{v} \frac{U_{z}(z+\Delta z, R+\Delta R)-U_{z}(z+\Delta z, R-\Delta R)}{2 \Delta R} \\
& \frac{U_{2}(Z, R+\Delta R)-U_{2}(2, R-\Delta R)}{2 \Delta R} \quad(A-15)
\end{aligned}
$$

Supposedly, we are going to calculate the values of $T, U_{Z}, U_{R}$, and $P$ of the points from 1 to $N$ on line $Z=1$, the following procedure is used. First we calculate the $T$ values for the points. Consider point 1 , as shown in Figure $A-1$; from the boundary conditions as given in Chapter II, subsitute the values


Figure A-l. The Finite-Difference Grid System
into Equation (A-15); after some arrangement it become

$$
\begin{aligned}
& P(0,1) u_{2}(0,1) \frac{T(1,1)-T(0,1)}{\Delta Z}=A_{T}\left[\frac{2 T(1,2)-2 T(1,1)}{\Delta R^{2}}\right. \\
&\left.+\frac{1}{R} \frac{T(1,2)-T(1,1)}{\Delta R}\right] \\
&+N_{E_{<}} A_{v}\left[\frac{u_{2}(0,2)-u_{2}(0,1)}{\Delta R}\right]^{2} \quad(A-16)
\end{aligned}
$$

Futher rearrangement, the Equation can be put into the following form:

$$
\begin{equation*}
\mathrm{b}_{1} T(1,1)+\mathrm{c}_{1} T(1,2)=\mathrm{d}_{1} \tag{A-17}
\end{equation*}
$$

where

$$
\begin{aligned}
\mathrm{b}_{1}= & P(0,1) \mathrm{U}_{\mathrm{Z}}(0,1) \Delta \mathrm{R}^{2} \mathrm{R} \\
\mathrm{c}_{1}= & 2 A_{\mathrm{T}}(\Delta \mathrm{ZR}+\Delta \mathrm{Z} \mathrm{\Delta R}) \\
\mathrm{d}_{1}= & -\mathrm{N}_{E C^{A}} \mathrm{~A}_{\mathrm{V}}\left[\mathrm{U}_{\mathrm{Z}}(0,2)-U_{Z}(0,1)\right]^{2} R \Delta Z \\
& -P(0,1) U_{Z}(0,1) T(0,1) \Delta R^{2} R
\end{aligned}
$$

For point 2, Equation (A-15), after substitution and rearrangement becomes

$$
\begin{align*}
& a_{2} T(1,1)+b_{2} T(1,2)+c_{2} T(1,3)=d_{2}  \tag{A-18}\\
& a_{2}= 2 A_{T} R \Delta Z-A_{T} \Delta Z \Delta R+P(0,2) U_{R}(0,2) \Delta Z \Delta R \\
& b_{2}=-A_{T} R \Delta Z-P(0,2) U_{Z}(0,2) \Delta R^{2} R \\
& c_{2}= 2 A_{T} R \Delta Z+A_{T} \Delta R \Delta Z-P(0,2) U_{R}(0,2) \Delta R \Delta Z \\
& d_{2}=-0.5 N_{E C} A_{V}\left[U_{Z}(0,3)-U_{Z}(0,1)\right]^{2} R \Delta Z \\
&-P(0,2) U_{Z}(0,2) \Delta R^{2} R
\end{align*}
$$

where

The system of Equations for each line can be written in the following form:

$$
\begin{array}{rlrl}
b_{1} T(1,1)+c_{1} T(1,2) & & =d_{1} \\
a_{2} T(1,1)+b_{2} T(1,2)+c_{2} T(1,3) & =d_{2} \\
a_{3} T(1,2)+b_{3} T(1,3)+c_{3} T(1,4) & =d_{3} \quad(A-19) \\
\ldots \ldots & \ldots & \\
a_{N-1} T(1, N-2)+b_{N-1} T(1, N-1)+c_{N} T(1, N) & =d_{N-1} \\
& a_{N} T(1, N-1)+b_{N} T(1, N) & =a_{N}
\end{array}
$$

The $a, b, c$, and $d$ values are known scalers. The matrix of coefficients $a, b$, and $c$ alone is called a tridiagonal matrix. The system is readily solved by the Gaussian elimination method [32]. The alogarithm for the solution of the system is

$$
\begin{align*}
& \beta_{1}=b_{1}, \quad \gamma_{1}=d_{1} / \beta_{1} \\
& \beta_{i}=\frac{b_{i}-a_{i} c_{i-1}}{\beta_{i-1}}  \tag{A-20}\\
& \gamma_{i}=\frac{d_{i}-a_{i} \gamma_{i-1}}{\beta_{i}} \\
& T(1, N)=2,3, \ldots, N \\
& T(1, i)=\gamma_{N} \\
& T
\end{align*} \quad i=2,3, \ldots, N \quad .
$$

After solving $T$, the values of $P$ is calculated

$$
P(1, i)=\frac{1}{T(1, i)\left(1.0-t_{a} / t_{0}\right)+t_{a} / t_{0}}
$$

The values of $U_{Z}$ and $U_{R}$ were then solved similarly to $T$ using Equations $(A-14)$ and ( $(A-13)$ respectively. The computational accuracy was then improved with an iterative procedure until the deviations of the assumed and the newly cal-
was less than $5 \%$ on $U_{Z}$ and $T$, and the solution is carried downstream to $Z+2 \Delta Z$, etc. A procedure developed by Wegstein [27] :

$$
\begin{equation*}
\bar{T}(\text { New })=0.33 T(\text { Old })+0.67 T(\text { New }) \tag{A-22}
\end{equation*}
$$

was used to accelerate the iterative process.

## APPENDIX B

## FINITE-DIFFERENCE PROGRAM AND RESULT

The finite- difference program was set up to calculate the thermal and momentum structure of gas plume (or free jet) exhausting into quiesent air. The input data required for the system are the exit velocity, exit temperature, the ambient temperature, and the stack diameter ( or the nozzle diameter). Other input data required for calculation purpose are described in the comment statements. The initial profiles of temperature and velocity are step functions. Also initial profiles of arbitary shape can be input into the program by changing the read format of the initial profiles. The computer program and the result print out are included in this APPENDIX.

```
\& WWATFIV IIME = 30 THIS PROGRAB SULVES THE FREE JET PRUHLEM BY USI NG
AN ABBUTT-TYPF FINITE UIFFERENEE SOLUTIUN UF THE
COISERVATIINSS.
ITHFIT UATA
INPUT UATA
TO-JET EXIT TEMPERATURE, UEG•R
TA-AMBLEMT TEMPERATURE, DEG. R
TA-AMBLENT TEMPERATUREDDEGOR
UU-JET
UIT VELICITY,FT. SEC.
    DFLLI-FIRST Z-UIREGTION FIVITF DIFFERENCE STEP SIZE
    UELZ2-SECDNO Z-DIFFCTION FIUIT: DIFFERENCE STEP SIZE
    DELZSOIHIRU Z-UIRECTION FINITE DIFFERENCE STEP SIZE
DELRI-FIRSI R-UIRECTIDN FINITE DIFFERENCE STEP SIZE
    DELRZ-JECUIYD R- DIKECTIUNFINITE DIFFERENCE STEP SIZE
    DELR 3-IHIRG R-DIRECTION FINITE
PR.JU-TLRBULENT PRANDTI NUMBER
    TUL-TULEREVEEFDF ITERATIVE CONVERGEIVCE
        II -NUNHER OF LINES UF THE FIRST Z-DIRECTION STEP SILE
        IS - FUR THE SECOND S
        IT-I FQR THE THIRL STEP
        JINGZ-Z JTEFOK, THE RE STEP SIZE
        LIMIT-LIMITING NUMEER OF ITERATIONS
        ULMENSIIN OLOVZ (501), OLDVK(501), OLDRO(501), VZ (501) , VR(501)
        XT(501), RC( 201\(), A(501), ~ 己(501), C(501), 0(501), T E S T(501), O L D T(501)\)
```



```
        XDELR3, リRNの, TOL
    \(20 U\) FURMAT (EF:O.O)
        WRITE (O, 1 L1) TO,TA, UO, DELZI, DELZ2, UELZ3, DELR1, DELR2,
        XDELR3, PKNO, TOL
    111FCNMAT (11 (2X,FR.3))
    REAO (כ, 200) II, IS,IT,JI,JS,JT,KNUZ,LIMIT
    WRITE ( \(5, \dot{\circ} \mathrm{O} 0)\) II,IS,IT,JI,JS, JT, KNOZ,LIMIT
    200 FURMAT (81:)
        RL:U (5,3CO) CUNST, CPC, DIA
    30? FURiAAT (3F:0.0)
    WKITE \((5,369)\) CDIST, CPC
    363 FORMAT (2 (5x,Flis.5))
    WRITE (t, it.79) UO, TO, TA,DIA
767Y FJMMAT (IH:,/////, 30X, FINITE DIFFERGHCE PROCRAM , /,
    12OX, 'CALCULATIGN THE THERMAL AVD MOMENTUM STRUCTURE UF GAS ,
    2'PLUME',
\(3 / 1, ~ 30 X, ~ ' X I T ~ V E L O C I T Y, ~ F T / S E C=', ~ F L O .5, ~ /, ~\)
```




```
        \(P A R=C .333\)
        \(\mathrm{J}=\mathrm{J} \mathrm{I}\)
        KAPPA=UELRO/DELRI
        NU=DtL \(3 / 0\) IR2
        PKAP=0:LZ2/חELZ1
        UMFGA= UFLL3/EELZl
        \(K N(1) L=K N C Z-1\)
: INITIAL PRIFILG VALULS
    MG \(1 \quad \mathrm{~A}=1\), K UUZ
    ULUVZ(1) \(=1,0\)
    \(O L 1) \Gamma(N)=1.1\)
    GLOVR(i) \(=0.0\)
    - ULDPO (iv) \(=1, \because\)
    DU 2. \(\mathrm{N}=\mathrm{KMOZ}\),
        HLDVZ( \(\because 1=0.0\)
        ULUT (N) \(=0 . \mathrm{C}\)
        OLOVR (N) \(=0.0\)
    - ULBPrl \((: u)=T: / T A\)
    DU \& \(N=1, J\)
    \(\vee 2(N)=[\) LLCV \((N)\)
    \(T(J)=0\) LDT \((\because)\)
    \(\operatorname{VR}(N)=U L D V R(N)\)
    \(\therefore\) RO(ij)=uLDRU(N)
    \(\mathrm{Pj}=1.0\)
    \(M=6\)
```

```
20n1 IF \(([I-M)+203,2003,3003\)
1003 IF (IS-M) 4003,5003,6003
4003 IF \((I T-M) ~ ¿ 000,2000,7003\)
\(3003 \quad 2 \mathrm{PI}=\mathrm{M}-1\)
    \(Z=Z P 1 * U E L L\)
    jEL \(2=0\) UER1
    DELR=UELRI
    \(J=J I\)
    POS = POS \(+1 . \dot{S}\)
    \(S E T=1.0 / \mathrm{CELZ}+1.0\)
    GO TO 1001
2003 DO 13 , N= \(1, J I, K \wedge P P A\)
    \(N: W=(1-1) / n A P P A+1\)
    OLOVZ(NEW)=OLDVZ(N)
    OLUT (NEW) \(=\) ULDT (IJ)
    OLDVR(VEW)=OLDVR(N)
    13 OLIRU('IFW)=OLDRO(N)
    [10 14 N=NE: 1 , JS
    OLDVL(ij) \(=0.0\)
    CLi) \(\Gamma(N)=0.0\)
    \(N M=. v-1\)
    \(p \mathrm{P}=\mathrm{ij}\)
    \(R \mu N_{1}=N-1\)
    \(O L \operatorname{OLR}(4)=C L D V R(N M) \nleftarrow R P M / R P\)
    14 OLURG (w) =TL/TTA
    0? \(48 \quad i=1,35\)
    \(V Z(v)=\| L O L(N)\)
    \(T(: N)=0 L \cup r(, d)\)
    \(\operatorname{VR}(i j)=11 L D V:(i)\)
    4\% RU(: \(\because=0\) =ULT (N)
6003 \(\quad 2 P 1=(I I-2)\)
    \(2 P-M-I I+1\)
    \(7=\angle P I * D E L Z I+Z P 2 * D E L Z 2\)
    DEL \(\angle=\mathrm{DELZ}\)
    LeLR=GELQ2
    \(J=J S\)
```



```
    GOTO 1001
5003 DU \(\ddagger=1 N=1 ; J S, N U\)
    NEW=(N-1)/UU+1
OLUVZ (ivEW) \(=\) ULUVZ (N)
    ULDT(iN[w)=i:LDT(ij)
    OLOVR (NW) = GLDVR (N)
    \(1 弓\) OLUKO (N: W) = 二LDRO(N)
    DU \(16 . V=N E W, J T\)
    うLUVZ(N)=0.0
    OLDT \((N)=0.1\)
    \(\mathrm{N} \mathrm{N}=\mathrm{N}-1\)
    \(R P=, V\)
    RPR: \(=1-1\)
    DLDVR(N)=OLLVR(NM)*RPM/RP
    IU OLDRU(J)=TG/TA
    DU \(38 \quad N=1, J T\)
    \(V Z(N)=\) r)LDVZ (N)
    \(T(v)=\) OLIT (H)
    \(\operatorname{VR}(H)=\operatorname{LLDVi}(N)\)
    30 RU(:N)=ULCRU(N)
/1;03 2 \(\mathrm{PL}=\mathrm{IS}-\mathrm{I}\) I
    \(2 P 3=M-1 S+1\)
    \(Z=\angle P 1 * D E L Z 1+\angle P 2\) * \(D E L Z 2+Z P 3 * D E L Z 3\)
    UELZ \(=0 E L Z 3\)
    DELR=DELR3
    \(J=J T\)
    PLS = PUS +UM T.GA
100: INCR=0.0
    \(J 1=J-i\)
\(J 2=J-2\)
    \(J=J-2\)
\(2 P: J T=4.73 /((T U / T A) * * 0.5)\)
    2POTC=0. ค
    IF (Z-2POTC) 1004,i004,2004
1004
    FVA=CP(SCTU 1004,1004,2004
    GOTO 1004
```

```
200% FVL=1.0
200; REYK=CUNST*((TU/TA)**0. 2)*, LURRU(1)*FVZ
    ZTPC=3.43/((TO/TA)**0.5)
    LTPC=U.8
    IF (Z-\angleTPC.)1005,1005,2005
1005 FT=CPC%PRNO
    GO 10 3005
200'2 FT=1 [ 0
    Cl=REYK/(D!LR*UELR)
    CC1=PRK/(D=LR*EELR)
3001 WRITE (0,50), %X,'UZ(Z,R)', 8X,'UR(Z,R)', 8x,'T(Z,R)',10x,
    1:RU(Z,R)!,11x,iZ:,14x,:R')
        VZ(J)=O.0
        Rrj(j)=[D/TA
H00. A(1)=RU(1):=VZ(1)/DELZ+4.0*CC1
    B(1)=-4.0*CC1
    D(1)=RUC(1)*VZ(1)*(1.0/OLDRO(1)-TA/TO)/((1.0-TA/TO)*DELZ)
    DO ; 9 : = =2,J1
    RP=v-1
    R=RP*L):LP
    CC<=PRK/(2.0*R*1)ELSN
    CC3=RO(N)*VR(N)/T2.O*DELR)
    CC4=RU(N)*VZ(N)/DELZ
    NP=N+1
    NN=11-1
    C(泣)=CC2-CL3-CC
    B(i,)=CC 3-C!, - -CC 1
    AT=(1.0/RO(j) ) =TO
    CP=171.5+U.02096*aT-0.0000リ396*AT*AT
    THIETA = (1.O/DLDKO(N)-TN/TO)/(1.0-TA/TII)
    &D(G)=CC4*THTTA+UO*UO/(32. 己*CP*(TO-TA))*REYK*(VZ(NP)*VZ
    X(NP)-L.0*VZ(NP)*VZ(NM)+VL(NM)*VZ(NM))/(4.0*DELR*UELR)
            DO 22 v=1, 1
    2- TEST (N)=T(N)
    C(i)=C(1)/A(1)
    DO 19 {1=2,.1
    NL=v-1
    B(NL)=b(N1)/A(vl)
    A(A)=A(N)-C(N)*B(N:)
    1) D(H)=(O(N)-C(N)*D(N1))/A(N)
    T(Jl)=D(Jl)
    RU(J1)=1.0/(T(J1)*(1.0-TA/TO)+TA/TO)
    DO 20 N=1,N2
    Niv= j2-!i+1
    Niv1=NN+1
    20) T(NN)=L(NN)-B(NN)*T(NN1)
    nu oG N=1,J
    T(.j)=PAR*T S [(j) + (1.0-PAR)*T(iv)
    6:2RO(N)=:.O/TT(N)*(1.0-TA/TO)+TA/TO)
    DG 23 H=1,GTEST(iv)-T(N))
    IF (DIIF-TUL)<3,23,9001
    23 COTTINUE
    ER<< =0.0
```



```
    74A(i)=R!(1)*VZ(i)/UELZ+4.0*Cl
    D(1)=RU(1)利L(1)*OLDVZ(1)/UELZ
    OH 3 N=2,J1
    RP=iv-1
    R=RP%UELR
    C2=REYK/(2.0)*R*DELR)
    (3=RO(Iv)*VR(N)/(2.0*DELR)
    C4=RU(iv)*VZ(N)/DELZ
    C(1) =C <-C3-Cl
    A(N) =C4+2.0*C,1
    8(iv)=Cs-C2-C1
    3U(N)=C4*OLLVZ(N)
    00 17 :=1,J
    17 TEつT(rN)=VZ(id)
        D(1)=D(1)/A(1)
```

```
            \(004 \mathrm{~N}=2, \mathrm{Jl}\)
            \(N 1=1-1\)
            \(B(i v l)=u(N 1) / A(N 1)\)
            \(A(N)=A(N)-6(N)+B(N 1)\)
            \(4 \quad D(\because)=(\mathrm{C}(\mathrm{N})-\mathrm{C}(\mathrm{N}) * 0(N \mathrm{~N}) / \mathrm{A}(\mathrm{N})\)
            \(V Z(J 1)=D(J 1)\)
            DU \(5 N=1, J_{1}\)
            \(1 N N=\sqrt{2} 2-i N+1\)
            N(N) =NiN+1
    \(5 \quad V Z(N N)=U(N N)-B(W N) * V Z(N N L)\)
    DO 6 6 6 iv \(=1, j\)
\(666 V Z(V)=P A R * T E S T(N)+(1 . O-P A R) * V Z(N)\)
    DU \(6 \quad N=2, J\)
    N \(1=1 N-1\)
    \(R P=N 1\)
    \(R=? P * D E L R\)
    \(C E 1=1.0 /((\therefore 0 * R U(N)) / D E L R-R U(N 1) / D E L R+R U(N) / R)\)
    \(C E \angle=R O(N) *(O L D V Z(N)-V Z(1)) / C E L Z\)
    \(C E 3=V Z(N) *(C L O N O(N)-R O(N)) / D E L Z\)
    \(C E^{\prime}=R U(N) * ?(N 1) / D E L R\)
    \(V R(1)=C L 1 *(C E 2+C E 3+C E 4)\)
    DO 10 N \(1, \mathrm{~J}\)
    ERKIJR \(=A H S\) ( \(F\) FST (N) \(-V Z(N))\)
    IF (ERHOR-7DL) \(10,10,5001\)
    IU GOVTINUE
    IF (ERRI) 26,20,5001
    24 GDiNTIIJE
        IF (POG-S.T) \(3002,3002,2000\)
\(300 . P O S=1.0\)
    DU 7 iv \(=1\), J
    \(R P=1-1\)
    \(R=2 \rho+C=L R\)
    \(R F=H\)
    IF (M.LE. 3 , ANU.M. 1, E. 27) GO TO 699
    If (M,LE3, G! TO 69
    कノ \(\mathrm{RF}=2 \mathrm{x}_{5} 5\) 。
    IF (M.LE.4*) G才) TO 699
    ; 7, RF二к*2
    Gij 10.399
    6: \(R F=P\)
        WRITE(S, 15U) VL(N), VR(N), T(N), RD(N), Z,RF
        \(\mathrm{GL}=10\)
        \(L L=L \dot{+} 2+1\)
        IF (LL.GT:i) G[J TO, 7
    7 Cutit INUE
    150 F(1KMAT (4F:5.5, \(2 F 15.3)\)
        \(M N=M / 2\)
            \(M N: C_{1}=M M+2\)
        IF (MMN.LT.M) GO TO 4001
    \(15<\) WRITE 6,121\()\)
400.
    OLUنZ( \(\dot{i})^{\prime}=\mathrm{V} /\) (N)
    OLUT (! (N) = T (. 6 )
    OLUVR (id) = VK (iN)
    + OLIJKO(H)=RI(H)
    \(M=i+1\)
    GOTC \(\angle O Q 1\)
9001 INCF=ITICR+
IF (INCR-LIMIT) 1009,1009,7001
1009 GO TO 3001
700 D DO \(12,11=1, \mathrm{~J}\)
    RP=ill
    \(R=R P * D t L R\)
    250 WRITE \((6,250) \quad V Z(N), V R(N), T(N), R U(N), M, N, Z, R\)
    WRITE \((6,35)\) TSST(v)
    7. FURMAT (F20.9)
2000
    COINT INUE
CUINT INUE
    CUNTINUE
    STUP
END
```

FINITE DIFFEマENCE PRJGRAM
CALCUL TIGV THE THERMAL ANG MBNEVTUM STRUCTURE OF GAS PLUME

```
#XIT VELUCITYMET/SEC= 37.30COC O- R = 912.0U000
#XIT PLUME TEMPERATURE, DES=R ROE.012ODO
```



| UZ (2, 2 ) |
| :---: |
| 1.coos |
| - 7 coor |
| 1.1000:j |
| 1.0000 |
| 1.9000 |
| $0.7447)$ |
| 0.79793 |
| 0.97967 |
| 0.70737 |
| 0.39547 |
| 0.9933: |
| 0.8130 |
| $0.14463 d$ |
| 0.61710 |
| 0.70951 |
| 0.11057 |
| 0.JC27 |
| 0.1 .6000 |

$U R(Z, R)$
0.00000
$C .0 .000$
0.0 .000
0.00000
0.00 .000
0.0000
0.00003
0.00013
0.06124
0.00869
0.00306
-0.00971
-0.0750
-0.10426
-0.13563
-0.10413
-0.10038
-0.03462
$T(Z, 2)$
1.00300
1.00000
1.00000
1.00300
1.00900
0.99 .79
0.97775
0.79709
1.99789
0.98548
0.89333
0.18955
0.04703
0.01737
0.00760
0.00777
0.00275
0.00700
$R O(Z, R)$
1.00000
1.00000
1.00000
1.000100
1.00070
1.00000
1.00002
1.00013
1.00039
1.00618
1.04476
1.52261
1.67597
1.71201
1.72171
1.72633
1.73035
1.733 .54
2
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100
0.100

[^0]|  |  |  |  |  | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| UZ (2, ${ }^{\text {a }}$ |  | 1-20.20 | $1.0 \cup 030$ | $0 .<00$ | $0.000$ |
| 1.000\% | O-OCCC | 1.00000 | 1. 30.30 | $0 \cdot 200$ | O-1CO |
| 1-006, | 0.cuco | 1-00000 | 1.00000 | $0 \cdot 200$ | O. 200 |
| 1.50000 | 0.05000 <br> 0.0 <br> 100 | 1.00ron | 1.0CC? | O. 200 | 0.300 |
| 5. 3597 | $0.0 \cdot 000$ |  | $1.000 \because 5$ | C-200 | 0.500 |
| 0.79970 | $C-C, 002$ | 0.97735 | 1.000. 2 | - - 200 | 0.600 |
| 0.77771 | 0.0010 | $0.7937 ?$ | 1.0007 | C. 200 | 0.700 |
| 0.79891 | 0.0053 | 1-99330 | 1.000=1 | 0.200 | 0.800 |
| 0.79331 | C.C $¢ 278$ | 0.79338 | $1 \cdot 015-1$ | 0.200 | 0.700 |
| 0.7651 | $0.0 \div 3: 4$ | 9.7636 | 1.077.1 | 0.20 | 1.000 |
| 0.6 | 0.04292 | ). $3<7<2$ | 1.0434? | $0 \cdot 200$ | 1.100 |
| 0.2838 r | -0. 0.259 | $0 \cdot 2447$ | 1.43415 | 0.200 | 1.200 |
| 0.53140 | - C.02375 | $\bigcirc \cdot 0930$ | 1.59178 | $0 . \angle 00$ | 1.300 |
| 0.63383 | -0.0,234 | - 03303 | 1.71450 | 0.200 | 1.400 |
| $0.1755 ?$ | -C.0.349 | 0.01537 | 1.72425 | 0.200 | 1.500 |
| 0.10767 | -0.00079 | 0.0575 | 1.73001 | 0.200 | 1.600 |
| $0 . C 30$ | $-0 \cdot 0.754$ | $\begin{aligned} & 0.00301 \\ & 0.00200 \end{aligned}$ | 1.73334 | 0.200 | 1.70 C |
| 0.1 COOL | -0.0.423 | 0.00200 | 1.73304 | 0.200 |  |
|  | UR(L,R) | 1(2, 21 | $R).(2, R)$ | 200 | $R$ |
| $U Z(2, R)$ 1.0000 | 0.05000 | 1.00200 | 1.00000 | $0 \cdot 300$ | - 0.100 |
| 1.0.00? | $0.0 \leq 000$ | $\cdots .00000$ | 1.00090 | 0.300 0.300 | 0.200 |
| 1. OCOU | $0 . C 9 C O O$ | 1.00000 | 1.0000 | 0.300 | 0.300 |
| 1.90000 | 0.00000 | - 00000 | 1.00001 | $0 \cdot 300$ | 0.400 |
| 0.79978 | 0.00001 | 0.77778 | 1.00054 | 0.300 | 0.500 |
| 0.99771 | 0.00003 | 0.79731 | 1.000? | 0.300 | 0.600 |
| 0.29754 | 0.00015 | 0.99754 | 1.00078 | 0.300 | 0.700 |
| 0.97769 | 0.0006't | 0.99768 | 1-00473 | 0.300 | 0.800 |
| $0.7889=$ | 0.02290 | 0.48388 | 1.02156 | 0.300 | 0.700 |
| 0.75030 | 0.01023 | 0.75314 | 1.09239 | $\bigcirc .300$ | 1.000 |
| $0.7377=$ | 0.02484 | 9.73773 | $1.397<3$ | 0.300 | 1.100 |
| 0.32887 | -0.0.378 | 0.32823 | 1.39 256 | 0.300 | 1.200 |
| 0.12831 | -C.C3530 | - 0.12745 | 1.66851 | 0.300 | 1.300 |
| 0.35404 | -0.04864 | 0.05335 | 1.703 .50 | 0.300 | 1.400 |
| 0.02440 | -0.0,253 | 0.02403 | 1.7035 | 0.300 | 1.500 |
| 0.01093 | -0.C-192 | 0.01071 | 1.728 .74 | 0.300 | 1.600 |
| 0.00375 | $-0.04934$ | $0.00366$ |  | 0.300 | 1.700 |
| 0.00000 | -0.04647 | 0.5 cos | 1.733 .14 |  |  |


|  |  |  | $R O(z, R)$ |  | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\cup Z\{2, \mathrm{Q})$ | $U R(i, ~ R ~$ | $1(2,2)$ $1.40) 0$ | $1.01000$ | $0.4013$ | $0.000$ |
| 1．OCCO「 | C－C＝OO | 1．00000 | 1．00090 | 0.400 | $0 \cdot 100$ |
| 1－J00\％ | C．0̌030 | $\frac{1}{1.00000}$ | 2.00030 | 0.400 | 0.200 |
| 1．9000 | $0 \cdot 0 . j 000$ | 1．0．9797 | 1．OOCin | 0.40 | 0.300 |
| $0.1777 \%$ | $\because$－ $2 \times \sim$ | ，27773 | 1．jnu．？ | 0.400 | 0.430 |
| 0.19714 | O．C．OM1 | －777） | 1．00022 | 0.400 | 0.500 |
| $0.3978<$ | 勺．CU心う | 0.77332 | 1．00035 | C .400 | 0.000 |
| 0.37713 | $\bigcirc \cdot C .23$ | － 097017 | 1．00152 | 0.400 | 0.700 |
| 0.79017 | $0 \cdot 0 \cdot 096$ | 0．79017 | 1.00701 | 0.400 | 0.800 |
| 0.13362 | $0 \cdot 0-3=4$ | － 723375 | 1．0224 | 0.400 | 0.900 |
| 0.73512 | $0.0 \pm 106$ | $0.73+75$ | 1．029．39 | 9.400 | 1.000 |
| $0.7725 i$ | $\cap \cdot 0.46$ | $0.77-46$ | 1.30945 | 0.400 | 1.100 |
| 0． $2=33 \mathrm{H}$ | －0．0．0208 | 1． 065775 | 1.55131 | 0.400 | 1.200 |
| 0 －6935 | －0．0－6\％5 | $0 \cdot 15775$ | 1.554549 | 0.400 | 1． 300 |
| 0.97430 | －C．04C63 | 0.07315 | 1.64549 $1.570 \div 9$ | 0.400 | 1.400 |
| 0.113527 | －0．04560 | $0 \cdot 03402$ | 1.71376 | 0.400 | 1．500 |
| 0.21613 | －0．04790 | $\begin{aligned} & n .01561 \\ & 0.027 t \end{aligned}$ | ． 726 － 4 | 0.400 | 1.600 |
| $\therefore \therefore C 583$ | -0.54058 -0.04382 | $\begin{aligned} & 0.0 .76 \\ & 0.0000 \end{aligned}$ | $1.7<36.4$ | 0.400 | 1.700 |
| $0.18000$ | －0．04382 | －0iJ00 | 1.73334 | 0.400 |  |
| UZ（Z，R） | UR（L，R） | T（ 2,0 ） | $R O(2, R)$ | 2 | $R$ |
| 1.30000 | $0.03 C 30$ | 1．00300 | $1.00 C 00$ | $0 \cdot 50$ | 0.100 |
| 1． 1.0000 | 0.00000 | 1．00000 | － 00000 | 0 | 0.200 |
| 0．7997\％ | O．CVOOO | 1．00300 | 1.0000 | 0.500 | 0.300 |
| 0．7939！ | 0.00000 | 0.99798 | 1－0000 | －． 500 | 0.400 |
| 0.7979 | $0.0 \bigcirc 02$ | 0.99793 | 1.0003 | 0.500 | 0.500 |
| 0.9996 ¢ | 0.09008 | 0.79768 | 1．00059 | 0.500 | 0.600 |
| 0.77853 | 0.04033 | 0.79363 | 1.00024 | 0.500 | 0.700 |
| 0.79432 | 0.02123 | 0．73427 | 1.0024 | 0.500 | 0.800 |
| $0.9777 \%$ | 0.04410 | 0.47701 | 1．039？ | 0.500 | 0.900 |
| 0.92032 | 0.0 .137 | 0．71） 78 | $1.132=4$ | 0.500 | 1．000 |
| 0.75096 | 0.02154 | － 74763 | 1.11893 | 0.500 | 1.100 |
| 0.34037 | 0.011083 | O． 38863 | $1 \cdot 34909$ | 0.500 | 1.200 |
| 0.18645 | －0．01915 | 0.18053 | 1.52508 | 0.500 | 1.300 |
| 0.39207 | $-0.03143$ | 0.07068 | 1.52506 | － 0.500 | 1.400 |
| $0.0453 i$ | －0．03731 | 0.04451 | 1.67900 | 0.500 | 1.500 |
| 0.02119 | －0．03903 | 0.02079 | 1.70779 | －． 500 | 1.600 |
| 0.00785 | $-0.03923$ | 0.00767 | $1 \cdot 72414$ | 0.500 | 1.700 |
| 0.20000 | －0．03592 | 0.00000 | $1.733 \cup 4$ | － 50 | 1.70 |


|  |  | T（l，） | $R \square(2, Q)$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\bigcup 2(2, R)$ | UR（\％，R ） | 1．00200 | $1.000 ; 0$ | 0.600 | 0．100 |
| 1.5000 | 0.30000 | 1． 20000 | 1－00030 | 0．600 | 0．200 |
| 1．）COO： | 0.6 .000 0.0000 | 1． 0.9777 | 1． 1.000 | 0．600 | $\begin{aligned} & 0.200 \\ & 0.300 \end{aligned}$ |
| 0．3972 | O．0．） 20 | C．3ワ27 | 1． 20001 | 0．60？ | $0.400$ |
| C－77？ | 0．0） 0 ¢1， 3 | 1）． 34438 | 1． 1.00 ¢ | 0.600 0.600 | 0．500 |
| 0．7993－ | O．CiUl2 | 2．39747 | 1． 1.0009 | C．600 | 0.600 |
| $\begin{aligned} & 0 .-9994 . \\ & 0.93793 \end{aligned}$ | $0.0 习 44$ | 0.99794 | 1．0CC年 0038 | －． 600 | 0.700 |
| $0.79<04$ | C．0＇15 50 | $0.39<04$ | 1．0123 | 0.600 | 0.900 |
| $0 .+7140$ | C．Cu458 | 0.9711 | 1．0417 | 0.600 | 0.900 |
| 0.90607 | 0.01151 | 0.70241 | 1.1293 | 0.600 | 1.000 |
| 0.13235 | $0 \cdot 0=454$ | O． 0.4087 | 1.33379 | 0.600 | $1 \cdot \frac{1}{2} 00$ |
| $0 .+1064$ | 0.0320 | 0.40514 | 1.50220 | 0.600 | $1 \cdot \frac{200}{300}$ |
| $0.122 ?$ | －0．0 0.05 | $0 \cdot 10805$ | 1.60645 | 0.600 | 1．300 |
| 0.4076 | $-0.02673$ | －0．05563 | 1.66504 | 0.600 | 1． 400 |
| 0．$) 5650$ | $-C \cdot 02410$ | －027？ | 1.7005 | 0.600 | 1.500 |
| 0.1270 | $-0.02717$ |  | 1.720 .6 | 0.600 |  |
| $0.01003$ | －0．03717 | $0.01044$ | 1.73354 | 0.600 | $1.700$ |
| 0.16000 | －0．03480 | －．U040 |  |  |  |
|  |  |  | RO（2．2） | 2 | $\stackrel{R}{2}$ |
| $U Z(Z, R)$ | $U R(Z, R)$ | T（2，2） | 1.0006 | 0.700 | $0.000$ |
| 1． 0000 | 0.00000 | 1．00200 | 1.00910 | 0.700 | － 200 |
| 0.9979. | 0.03000 | －． 0.99797 | 1.00000 | 0.700 | $0 \cdot \frac{200}{300}$ |
| 0.7979 | $0.6 \div 000$ | 0.97375 | 1.00002 | 0.700 | 0.300 |
| 0.73973 | 0.02001 | 0.7973 | 1.00008 | 0.700 | $0 \cdot 400$ |
| 0.79790 | 0．02c04 | O．97721 | 1.00032 | 0.700 | 0.5 CO |
| 0.79924 | O．C心O16 | － 3972 | 1．00124 | 0.700 | 5．600 |
| 0.3972 | 0.00055 | －． 29707 | 1.00443 | 0.700 | － 100 |
| $0.7895<$ | $0 . C 9177$ | 0．98744 | 1．01529 | 0.700 | 0.800 |
| $0.9(459$ | $0 \cdot \mathrm{C} 550$ | O．95442 | 1.04801 | 0.700 | 0.700 |
| 0.9250 | O．Cil52 | 0.39177 | 1.13594 | 0.700 | 1－C00 |
| $0.7167 \pm$ | $0 . C=786$ | － 0.42443 | 1．32206 | 0.700 | 1． 100 |
| 0.42643 | $0.0: 477$ | 0.42443 | $1.483 i 9$ | 0.700 | 1．200 |
| 0.32240 | $-0 . C 1047$ | －-232343 | 1.58933 | 0.700 | 1.300 |
| 0.2507 | －0．0＜152 | $0 \cdot 12343$ | 1.65439 | 0.700 | 1．400 |
| $0.1664 \%$ | －0．0i788 | O． 03324 | 1.69338 | 0.700 | 1－500 |
| 0.03312 | －0．03051 | －0．01272 | 1.71731 | 0.700 | 1.600 |
| 0.01294 | －0．03046 | 0.01200 | 1.73314 | 0.700 | 1.700 |
| 0.00000 | －0．02852 | －000．00 |  |  |  |





|  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\cup Z(2, R)$ | $\cup \mathrm{UP}(L, R)$ | $\begin{aligned} & 1(2,2) \\ & 5.97534 \end{aligned}$ | $1.01033$ | $1.400$ | $0.000$ |
| $0 \rightarrow 8790$ | O-CuOOO | 0.97344 | 1.01137 | $\therefore 400$ | 0.100 |
| 0.78650 | 0.60043 | 0.76732 | 1.014 0 | 1.400 | 0.200 |
| O.7823+ | $0 \cdot 0 ? 114$ | $0 \cdot 76732$ | 1.01975 | 1.400 | 0.300 |
| 0.37415 | C. C, $3 \rightarrow$ | 0.95000 | 1.01275 | 1.400 | 0.400 |
| $0.30!2$, | $C \cdot C: 456$ | -93- 20 | 1.037i | 1.4 CC | 0.500 |
| 0.7379 | $0 \cdot \mathrm{Cr} 782$ | 5.910, 02 | 1.05726 | 1.400 | 0.600 |
| $0.9036)$ | 9.0 025 | 0.87204 0.61887 | 1.08303 | 1.400 | 0.700 |
| 0.95360 | O. $0+253$ | 0.01387 0.74779 | 1.11944 | 1.400 | 0.800 |
| 0.79472 | 0.02433 0.096 | 0. 0.60547 | 1.164.4 | 1.400 | 0.900 |
| 0. 0.9642 | O. 0.9285 | O. 56992 | 1.222 .4 | 1.400 | 1.000 |
| 0.45432 | 0.01756 | 0.47011 | 1.239 .2 | 1.400 | - 200 |
| 0.37771 | 0.021 C | 0.37332 | 1.300?9 | - -400 | 1.300 |
| 0.850 | O.C1327 | 0.29305 | 1.43511 | 1.400 | 1.400 |
| 0.50 .19 | $0 . C 6538$ | $0 \cdot 20327$ | 1-20944 | 1.400 | 1.500 |
| 0.12700 | 0.09046 | 0.12366 | 1-28426 | 1.400 | 1.000 |
| $0.1) 6061$ | $-0.01247$ | 0.06165 | -65830 | 1.400 | 1.700 |
| 0.00000 | -0. C - 225 | $0.60 J C O$ | 1.733:4 | 1.400 |  |
|  |  |  |  |  |  |
| UZ (2, 2 ) | UR(L,R) | T (2, 2 ) | RO( $2, \mathrm{R})$ | Z |  |
| 0.98354 | 0.0 O00 | 0.96749 | 1.01375 | 1-500 | 0.00 |
| 0.92197 | 0 - CU057 | $0.96=21$ | 1.01474 | 1.500 | $0 \cdot 100$ |
| 0.37651 | 0.0 Cl 45 | 0.75311 | 1.01805 | 1.500 | $0 \cdot 2 \mathrm{CO}$ |
| C. 35723 | 0.0290 | 0.94525 | 1.02372 | 1.500 | $0 \cdot 300$ |
| 0.35127 | $0 . C-520$ | $0.72 う 10$ | 1.03274 | 1.500 | - 4.500 |
| 0.72646 | 0.05802 | 0.99567 | 1.04620 | - 50 | 0.600 |
| 0.8897 C | 0.01327 | ?. 35474 | 1.06531 | $1 \cdot 500$ | 0.700 |
| 0.93733 | $0.0: 835$ | $0 \cdot 80041$ | - 1922 | 1-500 | 0.800 |
| 0.76875 | 0.0.446 | 0.73185 | 1-12RU2 | 1-500 | - .900 |
| $0.68{ }^{0} 7$ | $0.0-858$ | 0.65025 | 1.11315 | - 500 | 1.000 |
| 0.28451 | 0.0 .965 | ?. 25935 | $1 \cdot 229<6$ | - 500 | 1. |
| 0.48157 | 0.02713 | 0.46484 | 1.29233 | $\cdots-500$ | - 200 |
| $0.39 \pm 82$ | $0 . C 2193$ | 0.37235 | 1.36174 | - 200 | - 200 |
| $0 \cdot 28966$ | $0.0+560$ | 0.28533 | 1.43366 | - - 500 | -300 |
| 0.20645 | 0.60755 | 0.20500 | 1.50711 | 1.500 | 1.400 |
| 0.3144 | 0.02484 | 0.13122 | 1.58135 | 1.500 | 1.500 |
| 0.06313 | 0.00216 | $0.663 \geq 0$ | 1.65699 | 1. 500 | 1.600 |
| 0.00000 | $0.0<196$ | $0 . \cos 0$ | 1.73354 | 1.500 | 1.700 |
| $0 \cdot 00000$ |  |  |  |  |  |


|  |  | T (2, 21 | $R O(Z, R)$ |  | $R$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $02(7,2)$ | 0. C¢C\% | 0.75300 | $1.01733$ | $1.600$ |  |
| 0.77833 | 0.3:073 | 0.75502 | 1.01877 | 1.600 | 0. 200 |
| 0.7706? | $0.0=178$ | 0.74301 | 1.02250 | 1.600 | 0. 300 |
| $0.75 \%$ | U.C $=340$ | 0.93371 | 1.0? 0 - 7 | -.600 | 0.400 |
| へ. 3 ¢ \% | $0 \cdot C=386$ | - $31-15$ | 1.03 | 1.60 | 0.500 |
| 0.71425 | 0.C.735 | O-80942 | 1.07355 | 1.600 | 0.600 |
| $0 . \therefore 757 \pm$ | $0 \cdot 0-397$ | $0.837-33$ | 1.10177 | 1.600 | 0.700 |
| 0.22273 | $0.0-707$ | O.78 $0^{1} 74$ | 1.13757 | 1.600 | 0.800 |
| 0.753 .7 | $0 \cdot 0-410$ | 0.61518 | 1.18204 | 1.600 | 0.900 |
| 0.5794 | O.C.770 | 0.63013 | 1-235:0 | i. 600 | 1.000 |
| $0.5764 \%$ | $0 . C \leq 871$ | - 0.54317 | 1. 29676 | 1.600 | 1.100 |
| 0.47332 | $3.0=677$ | $0.457 C 2$ | 1.36346 | 1.600 | $1 \cdot 200$ |
| $0 \cdot 2025 \div$ | ]. $0 \div 6 \div$ | 0-28-2 26 | 1.43346 | 1.500 | $1 \cdot 300$ |
| $0.6927 i$ | 0.01741 | 0.28255 | 1. 50518 | 1.600 | 1.400 |
| 0.21025 | $0.0-237$ | 0.20539 | 1.57970 | 1.600 | 1. 500 |
| $0.1347 \%$ | 0.09520 | $\bigcirc \cdot 13273$ | 1.65531 | 1.600 | 1.6 CO |
| $0.06507$ | $\begin{aligned} & 0.01509 \\ & 0.03515 \end{aligned}$ | $\begin{aligned} & 6.00421 \\ & 0.00000 \end{aligned}$ | 1.65334 | 1.600 | 1.700 |
| - . 000 |  |  |  |  |  |
| U ( 2 , R) | UR (2, ${ }^{\text {U }}$ ) | T(2, $)$ | FU) $(2,2)$ | 1.700 | 0.800 |
| $0.97 \dot{C l}$ | 0.00000 | 0.94381 | 1.02215 | 1.700 | 0.100 |
| 0.77054 | $0 . C 6090$ | 0.94593 | 1.023 21 | 1.700 | 0.200 |
| 0.76381 | 0.00212 | 0.93713 | 1.02734 | 1.700 | 0.300 |
| 0.9515 | 0.00371 | 0.96157 | 1.03433 | 1.700 | 0.400 |
| 0.93193 | 0.00649 | 0.593 .98 | 1.04550 | 1.700 | 0.500 |
| 0.70203 | 0.0 .000 | $0 \cdot 862 \leq 1$ | 1.03169 | 1.700 | 0.600 |
| $0.3625+$ | 0.04437 | $0.8 \div 159$ | 1.001975 | 1.700 | 0.700 |
| 0.30844 | 0.01922 | 0.70632 | 1.14570 | 1.700 | 0.800 |
| 0.7401 | $0 \cdot 04377$ | 0.69734 | 1.18936 | 1.700 | 0.900 |
| 0.55897 | 0.25698 | O- 022234 | 1.1.242?4 | 1.700 | 1.000 |
| 0.26859 | 0.02790 | - 23727 | 1.342141 | 1.700 | 1.100 |
| 0.4746 | 0.02650 | - 45279 | 1.365 .70 | 1.700 | 1.200 |
| 0.39214 | $0.0 \div 310$ | 0.36700 | 1.43425 | 1.700 | 1.300 |
| $0 \cdot 29440$ | O-Ci888 | 0.50457 | 1.50547 | 1.700 | 1.400 |
| 0.21287 | $0 . C 1458$ | $\begin{aligned} & 0.20571 \\ & 0.13352 \end{aligned}$ | 1.57912 | 1.700 | 1.500 |
| 0.13716 | 0.01100 | $0.13352$ | 1.65515 | 1.700 | 1.600 |
| 0.0665 | 0.05965 | $\begin{aligned} & 0.06479 \\ & 0.00500 \end{aligned}$ | 1.733 .4 | 1.700 | 1.700 |


|  |  | $T(7, \dot{*})$ | $R \square(Z, R)$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| U2（2， 5 ） | UR C C C C | C． 93320 | 1.026 .36 | 1．800 | O． 000 |
| 0.76635 | C．CLCOC | C－93－320 | 1．02824 | 1.800 | 0.100 |
| 0.36397 | O． $061-27$ | 0.93210 | 1．03253 | 1．800 | 0.200 |
| 0.45643 | $0 \cdot C \cdot 246$ | －．925．9 | 1．04009 | 1.800 | 0.300 |
| O． 7428, | C．C． $\mathrm{C}^{4} \mathrm{C}$ |  | 1．051 -05 | 1.800 | 0.400 |
|  | O－0U700 | O－32＋17 | 1.05774 | 1.800 | 0.500 |
| 0.69125 | $0 \cdot 0.057$ | － 0 － 0270 | 1．08ヲう1 | 1.800 | 0.600 |
| 0.84934 | 0.01481 | 0．80370 | 1．118：0 | 1.800 | 0.700 |
| 0.79472 | C． $0-934$ | 0.75043 | 1.15379 | 1.800 | 0.800 |
| 0.12704 | C． $0 \leq 343$ | ？． $3647 \%$ | 1．157．88 | 1．800 | 0.900 |
| 0.04767 | $0.0 \div 638$ | $\bigcirc \cdot 61019$ | － 24800 | 1.300 | 1.000 |
| 0.26589 | $0.6-737$ | 0． 52372 | 1． 20612 | 1.800 | 1－1C0 |
| 0.47344 | 0.02530 | －44， 25 | 1－35890 | 1.200 | 1．2C0 |
| 0.38073 | O．C $<352$ | $0.363<9$ | 1．358978 | 1.800 | 1.300 |
| 0． | O．C：009 | $0 \cdot 23<37$ | $\frac{1}{1} \cdot 50575$ | 1.80 C | 1.400 |
| 0.2145 | O．01044 | 0． 204321 | － 579 ¢ | 1.000 | 1.500 |
| 0.1398 | $0.0+330$ | $0 \cdot 13363$ | 1.57989 | 1.800 | 1.600 |
| 0.30752 | O．C1107 | 0．心o ú2 | 1． 1.73484 | 1.80 C | 1.700 |
| 0.20000 | 0.01002 | C．CCOCO | 1.733 R |  |  |
|  |  | T（ $7, R$ ） | $Q \cup(Z, R)$ | 7 | ${ }^{R}$ |
| $U Z(2, R)$ | $0 \mathrm{CR}=20$ | 0.92648 | $1.031+3$ | 1.900 | 0.000 |
| 0.95934 | 0.60000 | － 92356 | 1.03343 | 1.900 | 0.100 |
| 0.75653 | 0.00124 | C． 32350 | 1．03854 | 1．900 | $0 \cdot 200$ |
| 0.44644 | 0.00279 | ก． 71342 | 1．04610 | 1.900 | 0.300 |
| 0.33375 | 0.06487 | 0.89588 | 1．04610 | 1.900 | 0.400 |
| 0.71120 | $0 . C 6763$ | $0 \cdot 37003$ | 1－053 | 1.900 | 0.500 |
| 0.87730 | 0.01111 | $0.8351 \frac{1}{2}$ | 1－09744 | 1.903 | 0.600 |
| 0.83647 | $0.0: 518$ | $0 \cdot 79522$ | 1． 1.12644 | 1.900 | 0.700 |
| 0.7015 | $0 . C 1943$ | 0.73518 | 1．126201 | 1.900 | 0.800 |
| 0.71471 | $0 \cdot 0=323$ | O．07028 | $1 \cdot 20500$ | 1． 900 | $0.9 C 0$ |
| 0.3375 | $0.6 \leq 590$ | C－59805 | 1．25471 | 1.900 | 1.000 |
| 0.55334 | $0 . C 5690$ | 0.52006 | 1－25491 | 1.900 | 1．100 |
| 0.46534 | 0.02616 | 0．43748 | 1．37233 | 1.900 | 1．200 |
| 0.37904 | $0.0<403$ | $0 \cdot 35397$ | 1.372790 | 1.900 | 1.300 |
| 0.69512 | 0.02110 | 0.28046 | 1.450705 | 1.900 | 1.400 |
| 0.21534 | $0 \cdot 0+300$ | $0 \cdot 20307$ | 1．57941 | 1.900 | 1.500 |
| 0.1339 | 0.01521 | $0 \cdot 13324$ | 1． 1.65494 | 1.900 | 1.600 |
| 0.06226 | 0．6 6309 | 0.06497 | 1．0733，4 | 1.900 | 1.700 |
| 0.06000 | 0.6 .186 | 0.60000 | 1．133－4 | 1.900 |  |


|  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |


| UZ ( $2, ~$ 只) | UP( 2,8$)$ | $T(Z, R)$ | $R \cup(2, R)$ | 2 | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0.70327 | C.COOOC | 0.55357 | 1.06607 | 2.400 | 0.000 |
| $0.3057=$ | $0.0: 210$ | 0.34707 | 1.06734 | 2.400 | 0.100 |
| 0.2949 ; | 0.05430 | 0.33303 | 1.9730C | 2.400 | 0.2 CO |
| $0.16 \%$ | 0.06 .093 | $\bigcirc .81344$ | 1.0¢3? 4 | 2.400 | 0.300 |
| 0.3500 | $0 . C u 377$ | $0.730: 1$ | 1.037:4 | 2.400 | 0.400 |
| 0.31473 | 0.01233 | 0.75.j20 | 1.11504 | 2.400 | 0.5 SCO |
| 0.77053 | $0 \cdot \mathrm{C}, 593$ | 0.71154 | 1.13007 | 2.400 | 0.600 |
| 0.71754 | 0.01879 | 0.50371 | 1.16708 | 2.400 | 0.700 |
| 0.6566 | 0.02178 | 2. $603 \leq 1$ | 1.20157 | 2.400 | 0.800 |
| 0. 3874 \% | 0.0247 | 2. 34169 | 1. 24056 | 2.400 | 0.900 |
| 0.1191. | O.C:こ73 | 0.47572 | 1. 28449 | 2.400 | 1.000 |
| $0 . \div 45 \%$ | C.0.175 | $0.41 . j 58$ | 1.33237 | 2.400 | 1.100 |
| 0.3727 | $0.0-751$ | 0.34510 | 1.38348 | 2.400 | 1. 200 |
| 0.26 .283 | $0 \cdot 0 \div 514$ | 0.28184 | 1.43670 | 2.400 | 1.300 |
| $0 .-3710$ | 0.05186 | 0.22203 | 1.47031 | 2.400 | 1.400 |
| 0.17645 | 0.04705 | 0.10657 | 1.54479 | 2.400 | 1.5C0 |
| $0.217{ }^{\circ}$ | $0 . C 3224$ | 0.11603 | 1.59779 | 2.400 | 1.600 |
| 0.9733 | -0.CU213 | $0 . C 7076$ | $1.643: 5$ | 2.400 | 1.700 |
| 0.03401 | -0.00389 | O.03317 | 1.57254 | 2.400 | 1.800 |
| O. icucs | -0.00362 | 0.00300 | 1.73304 | 2.400 | 1.900 |
| $U Z(Z, R)$ | $U R(L, R)$ | $T\left(Z, Q^{2}\right)$ | RO(2,R) | 2 | R |
| 0.36910 | 0.0 OCOO | 0.32007 | 1.07945 | 2.600 | 0.000 |
| 0.38545 | $0 . C 1233$ | 0.32217 | 1.09139 | 2.600 | 0.100 |
| 0.1742 | 0.04480 | 0.81045 | 1.03722 | 2.600 | 0.200 |
| $0 \cdot 5527$ | $0.0 \% 749$ | 0.77070 | 1.07757 | 2.000 | 0.300 |
| 0.22924 | C.01040 | C. 76353 | $1.111 \geq 2$ | 2.600 | 0.400 |
| 0.77234 | $0.0 \div 347$ | C. 72349 | 1.129\%3 | 2.500 | 0.500 |
| 0.14706 | $0.0 \div 053$ | 0.58617 | 1.15317 | 2.600 | 0.600 |
| 0.5973 | 0.01037 | 0.63726 | 1.13138 | 2.600 | 0.700 |
| $0.6385 \%$ | O.CE172 | 0.58200 | 1.21445 | 2.600 | 0.800 |
| $0.744{ }^{\circ}$ | $0.0<332$ | 0.52415 | 1.25213 | 2.000 | 0.900 |
| 0.2067 | $0.0 \bigcirc 397$ | n-46,293 | 1. 27418 | 2.600 | 1.000 |
| 0.43161 | 0. $\mathrm{C}-357$ | 0.40080 | 1.33978 | 2.600 | 1.100 |
| 0.36927 | 0.02203 | 0.33733 | 1.38817 | 2.600 | 1. 200 |
| 0.30322 | $0.0 \div 754$ | 0.27736 | 1.43843 | 2.600 | 1.300 |
| 0.24083 | 0.0 isc 8 | 0.22340 | 1.48953 | 2.600 | 1.400 |
| 0.18291 | 0.01191 | $0 \cdot 17052$ | 1. 24091 | 2.600 | 1.500 |
| 0.12980 | $0.0573 \%$ | 0.12183 | 1.59155 | 2.600 | 1.600 |
| 0.08171 | 0.0 .2310 | 0.07708 | 1.64102 | 2.600 | 1.700 |
| 0.03904 | 0.00061 | 0.03675 | 1.58807 | 2.600 | 1.800 |
| 0.00000 | 0.00056 | 0.00000 | 1.733 .4 | 2.600 | 1.900 |


| UZ $(7, Q)$0.368130.6430.35290.30 |
| :---: |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |

$U 2(2, R)$
0.84655
0.3427
0.33119
0.31191
0.78485
0.75012
0.70807
0.05337
0.00502
0.54643
0.48525
0.42293
0.36114
0.30095
0.24320
0.18350
0.13709
0.08865
0.04312
0.00000


$T(7,2)$
0.77037
0.76076
0.75537
0.73051
0.71037
0.67740
0.63310
0.59329
0.5437
0.4735
0.43069
0.38123
0.32525
0.27255
0.22085
0.17158
0.12491
0.08085
0.03933
0.00000

| KO(2, R) |
| :---: |
| 1.07332 |
| 1.07528 |
| 1.10121 |
| 1.11120 |
| 1.12541 |
| 1.14344 |
| 1.16739 |
| 1.17475 |
| 1. 226.41 |
| 1. 26333 |
| 1.303,9 |
| $1.347-0$ |
| $1.373: 7$ |
| 1.44104 |
| 1.48979 |
| 1.53940 |
| 1.58877 |
| 1.53779 |
| 1.68600 |
| 1.733 ¢ 4 |



R
0.000
0.100 0.100
0.200
0.300
0.40 C
0.40 C
0.500
0.500
0.600
0.600
$C .700$
0.800
0.800
0.900
0.900
1.000
1.100
1.200
1.200
1.300
1.400
1.500
1.500
1.7000
1.700
1.800
1.700
1.700
$Z$
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000
3.000

R
0.100
0.200
0.300
0.400
0.400
0.500
0.500
0.800
0.900

1 - CCO
1.100
1.200
1.200
1.200
1.300
1.400
1.500
1.600
1.700
$1: 800$
1.800
1.900

|  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| UZ（ $1, R)$ | UR（ $2, R)$ | T 0.68488 | $\begin{aligned} & R D(27 R \\ & 1.15370 \end{aligned}$ | 3.500 | 0.100 |
| 0.1754 | O－atrol | 0.67142 | $1.161 \% 3$ | 3.500 | 0.200 |
| 0.76080 | 0.0 .537 | 0.63204 | 1．15444 | 3.500 | 0.400 |
| 0.5421. | 0.0 ¢ 203 | 9.57315 | 1．22239 | 3.500 | 0.600 |
|  | U．0．799 | 11.4743 | 1．27432 | $3 \cdot 5 C ?$ | C．800 |
| 0.45035 | $0 . C 2775$ | C．40336 | 1．337：3 | 3.500 | 1.000 |
| 0.35284 | $0 . C<732$ | 0.31509 | 1.40823 | 3.5 CO | － 200 |
| $0.2570 t$ | 0.0007 | 0.13330 | 1.43039 | 3.500 | 0 |
| 0．2755； | C．C5037 | 0.16332 | 1.54827 | 3.200 | 1.600 |
| 0.119. | －0．0．397 | 0.10774 | 1．6C6：0 | 3.500 | 1.800 |
| $0.0753)$ | －0．0こので | $\bigcirc .07043$ | 1．64803 | 3.500 | 2.0 CO |
| 0.04622 | －0．05353 | 0.14743 | 1．075 3 | 3.500 | 2 |
| 0.3296 | $-0.0 \cup 070$ | 0.03033 | $1.595+8$ | 3.500 | 2．400 |
| $0 \cdot \mathrm{n} 174 \%$ | －0．0－338 | $0.0182 ?$ | 1．71038 | 3.5 CO | 2.6 |
| 0．：0779． | －0．0u24i | 2． 20.331 | 1.72333 | 3.500 | 2.800 |
| 0.6 COC | －0．00875 | $0 . \operatorname{coOCO}$ | 1.733 .4 | 3.500 | ． 000 |
|  |  |  |  |  |  |
| UZ（Z－R） | UR（Z，R ） | T（2， 2 ） | RO（Z，R） |  | 0．${ }^{\text {R }}$（00 |
| $0 \cdot 10584$ | 0. cruco | 0.60025 | 1． 29938 | 4.000 | 0.200 |
| $0.0927=$ | O．CÓ24 | 0.59471 | －$\cdot 2063$ | 4.000 | 0.400 |
| 0.65436 | 0.01625 | 0.56202 | 1．22696 | 4.000 | 0.600 |
| 0.59400 | 0.02230 | － 510.8 | 1． $30 \frac{1}{575}$ | 4.000 | 0.800 |
| 0． 21749 | $0 . C 2705$ | 0.44570 | 1.35819 | $4 . \mathrm{COO}$ | 1．000 |
| $0.4323 \div$ | 0.03310 | $0 \cdot 37013$ | 1.41630 | 4.000 | 1．2C0 |
| 0.3468 ＇t | $0.0 \div 54 \mathrm{C}$ | ¢．30573 | 1.47375 | 4.000 | 1.400 |
| 0.26803 | 0.0 .894 | O． 18371 | 1． 1.52786 | 4.000 | 1.600 |
| 0.20047 | －0．06926 | $0 \cdot 1337 \frac{1}{8}$ | 1.57536 | 4.000 | 1.800 |
| 0.24591 | －0．00252 | $0 \cdot 13758$ |  |  | 2.000 |
| $0 \cdot \mathrm{C} 434$ | －0．01421 | $0 \cdot 10043$ | 1.6448 | 4.000 | 2.200 |
| 0.07322 | －C．C－280 | 0.07172 | 1．04672 | 4.000 | 2.400 |
| 0.049 CH | －0．0．827 | 0.04897 | 1．67359 | 4.000 | 2.600 |
| 0.1297 | －0．0．696 | 0.03002 | 1．716？8 | 4.000 | 2.800 |
| 0.61367 | －C．C -0.120 | 0.01374 | 1．733－4 | 4.000 | 3.000 |
| 0.00000 | $-0.02941$ | 0.00000 | 1．733：4 | ． |  |


|  |
| :---: |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |
|  |  |



$$
\begin{aligned}
& z \\
& 5.000 \\
& 5: 000 \\
& 5: 000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5: 000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5.000 \\
& 5.000
\end{aligned}
$$

| UZ ( $2, \mathrm{Q}$ ) | UR1L, R) | T ( $2,{ }^{2}$ ) | $R O(2, R)$ | Z | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $0.352^{\circ}$ | $0 . C .000$ | $C .44737$ | $1.303 \% 1$ | 5.500 | 0.000 |
| 0.54540 | 0.0565 | J.44312 | 1.30834 | 5.500 | $0 . \angle 40$ |
| 0.52121 | C.C-C74 | 0.42382 | 1.322;2 | 5.500 | 0.400 |
| 0.4520 | $0.04-1$ | 0.39379 | 1.345:3 | 5.500 | -.600 |
| 9.4込73 | $0 \cdot n-754$ | 9.35503 | $1.374 \% 8$ | 5.500 | 0.800 1.000 |
| 0.37804 | 0.0 .373 | ). $31332^{2} 5$ | 1.40926 | 5.500 5.500 | 1. 200 |
| 0.3220 .5 | $0.0 \div 342$ | 0.27334 | 1.44631 | 5.500 5.500 | 1.400 |
| 0.20827 | $0.0: 585$ | 0.22735 | 1.4. 52338 | 5.500 | 1.600 |
| $0 \cdot 41883$ | $0.0 \div 440$ | $0 \cdot 103 \leq 7$ | 1.5 -5078 | 5.500 | 1.800 |
| $0.1747 \%$ | $0 \cdot C=146$ | 0.11353 | $1.594 \%$ ? | 5.50 | 2.000 |
| 0.13572 |  | 0.11953 | 1.62573 | 5.500 | 2.200 |
| 0.1020 | 0.01324 $0 . C U 329$ | 0.00437 | 1.65503 | 5.500 | 2.400 |
| 0.536 | 0.00154 | 0.64038 | 1.69334 | 5.507 | 2.600 |
| $0.3217 \%$ | 0.00052 | 9.01954 | $\begin{aligned} & 1.70933 \\ & 1.73384 \end{aligned}$ | 5.500 5.500 | 3.000 |
| 0.0000 | 0.00023 | C.C0000 | 1.73384 |  | 3.00 |
| UZ (Z,R) | UR (Z,R) | T12.2) | RO(Z,R) | 6.200 | $0 . \mathrm{R}_{\mathrm{R}}^{\mathrm{C} 0}$ |
| 0.51673 | $0.0<000$ | 0.41232 | 1.331, ${ }^{\text {R }}$ | 6.000 | 0.200 |
| 0.50915 | $0 \cdot C .544$ | 0.40033 | 1.33536 | 6.000 | 0.400 |
| 0.48707 | $0 . C 1033$ | 0.39013 | 1.34.3776 | 6. 0.00 | 0.600 |
| 0.45285 | 2.01427 | O. 05448 | 1.39414 | 6.000 | 0.800 |
| 0.40972 | 0.01670 | $0.33<04$ | 1.424? | 6.000 | 1.000 |
| $0 \cdot 26154$ | 0.01827 | 0.29348 | 1.45845 | 6.000 | 1.200 |
| 0.31177 | 0.01624 | 0.25731 | 1.49327 | 6.000 | 1.400 |
| 0.26325 | $0 \cdot \mathrm{C}-710$ | O. 0.18347 | 1.52808 | 6.000 | 1.600 |
| 0.21784 | -. 51516 | -.14793 | 1.56199 | 6.000 | 1.800 |
| $0 \cdot 17633$ | $0.0-278$ | -14393 | 1.59448 | 6.000 | 2.000 |
| 0.13703 | 0.01029 | 0.11910 | 1.52525 | 6.000 | 2.200 |
| $0 \cdot \pm C 557$ | C. 0795 | O.Otis? | 1.65459 | 6.000 | 2.400 |
| 0.0754. | -0.0ヶ4 42 | 0.04174 | 1.68231 | 6.000 | 2.6C0 |
| 0.02813 | C.C. 344 | 0.02907 | 1.70857 | 6.000 | 2.800 |
| -0.000u | $0 . C \leq 304$ | 0.00000 | 1.73334 | 6.000 | 3.000 |


| $U Z(2, Q)$ | UR（ 72.2$)$ | T（2，${ }^{\text {P }}$ ） | RO（Z，R） |  | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0.48087 | O．CECOO | 0．37179 | 1．35734 | $6.500$ | $0.000$ |
| 0.47426 | 0.0 ¢ 020 | ©．37315 | 1.36112 | 6.500 | $0 \cdot 200$ |
| 0.45503 | $0 \cdot C=390$ | $0 \cdot 35908$ | $1.372 \therefore 4$ | 6.500 | C．400 |
| $0.42 \% 25$ | $\mathrm{C}-\mathrm{C}-27 \mathrm{y}$ | 0.33711 | $1.339^{\prime} 7$ | 6.500 | 0.600 |
| 0.3375 | $0.0: 7.44$ | 1）．3C71 | 1.41322 | 6.500 | 0.300 |
| $0.3447 \%$ | 0．0：743 | 0.27734 | $1.440 \% 3$ | t． 500 | $1 \cdot C C O$ |
| 0.36057 | $0 \cdot 0+825$ | 0.24371 | $1.470 \div 0$ | 6.500 | 1.200 |
| $0 .<565:$ | 0.0 .700 | 0.20770 | $1.502+2$ | $0 \cdot 500$ | 1.400 |
| 0.31475 | C． 12024 | D．17709 | 1． 23443 | 6.500 | 1.600 |
| $0.1757 こ$ | $0 . C=445$ | 0． 14.478 | 1． 50607 | 6.500 | 1.800 |
| 0.3737 | － 0.0251 | C． $115 \pm 9$ | 1．59655 | 6.500 | 2.000 |
| 0.18722 | 0.01063 | 0.08790 | 1.62634 | $6=500$ | $2 \cdot 200$ |
| 0.9772 | $0.01 ; 495$ | $0.06+84$ | 1．65503 | 6.500 | $2 \cdot 400$ |
| 0.04357 | C．C：75E | 0.04171 | 1.63235 | $6 \cdot 500$ | 2.600 |
| 0.22394 | $0.0 i 659$ | 0.02014 | 1．7085？ | 6.500 | 2.800 |
| O．＇）COCis | $0.0<599$ | 0.00900 | 1．72334 | 6.500 | 3.000 |
| UZ（ $2, ?$ ） | $U R(1, R)$ | $T(2, R)$ | RO（ $2, \mathrm{R})$ | 7200 | R ${ }^{R}$ |
| $0 \cdot 14740$ | $0.0 才 000$ | 0.34547 | 1.39235 | 7.000 | 0.000 |
| 0.44173 | 0.04496 | $0 \cdot 34<36$ | 1.38570 | 7.000 | $0 \cdot 200$ |
| 0.42500 | 0.00947 | 0.33034 | 1.37554 | 7.000 | 0.400 |
| 0.39903 | 0.01320 | 0.31143 | 1.41126 | 7.000 | － 6.600 |
| 0.36600 | 0.01593 | 0.28730 | $1 \cdot 43194$ | 7．000 | 0.800 1.000 |
| 0.32840 | $0 . C-759$ | $0 \cdot 25951$ | 1.45647 | 7.000 | 1． 200 |
| 0.2996 | 0.01820 | 0．22777 | 1.43357 | 7.000 | 1－200 |
| $0.3488 i$ | 0.01775 | $0 \cdot 19745$ | 1.51246 | 7.000 7.000 | 1.400 1.600 |
| 0.11016 | C．0 -705 | $0 \cdot 16757$ | 1.54176 | 7.000 | 1．600 |
| 0.17357 | 0.01572 | O． 14079 | 1.57148 | 7.000 | 1－800 |
| $0 \cdot \dot{3} 337$ | 0.21419 | 0.11348 | 1.60055 | 7.000 | 2.000 |
| $0 \cdot 2 C 754$ | O． $0: 26: 2$ | 0.08777 | 1.62872 | 7.000 | $2 \cdot 200$ |
| 0.07794 | $0.0 \pm 115$ | 0.06308 | 1． 55644 | 7.000 | 2.400 |
| $0.5503<$ | 0．0 0.78 | 0.04110 | 1.69307 | $7.0 C C$ | 2.600 |
| 0.32442 | 0.0 .1984 | $0.01+73$ | 1.708 .55 | 7.000 | 2.800 |
| 0.0000 | 0．0isl1 | 0.60000 | 1.73334 | 7.000 | 3.00 |



| $U Z(Z, R)$ | $\cup R(L, R)$ | T ( 2,2 ) | $R \cup(2, R)$ | $Z$ | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0.35986 | o.cicoc | $0.26,03$ | $1.451) \mathrm{C}$ | 8.500 | 0.600 |
| $0.3561 \%$ | 0.05431 | 0.26301 | 1.45334 | 8.500 | 0.200 |
| 0.34512 | 0.02831 | C. 25 つ32 | 1.46 C 24 | 8.50 C | 0.4 CO |
| $0 \cdot 32772$ | 0.0.179 | ก. 24309 | 1.47136 | 8.500 | 0.600 |
| 0.3 CJI | l.C.460 | 0.2こ705 | 1.43512 | 8.500 | 0.800 |
| 0.? 7879 | $0 . C 1007$ | ).2031? | 1.50405 | 8.500 | 1. COO |
| $0 .<4990$ | 0.Cig01 | 0.13731 | 1.52431 | 8.500 | 1.200 |
| $0.2178{ }^{\circ}$ | 0.02871 | 0.10327 | 1.54630 | 8.500 | 1.400 |
| $0.2895<$ | 0.0 .936 | 0.14275 | 1. $5691+3$ | 8.500 | 1.600 |
| 0. 5747 | $0 \cdot 0-453$ | 0.12028 | 1.293:1 | 8.500 | 1.800 |
| 0.1302j | C.01800 | 0.09824 | 1.517:5 | 8.500 | 2.CCO |
| 0.10202 | $0.0: 720$ | $0.076: 7$ | 1. 34125 | 8.500 | 2.200 |
| C.0747n | 0.01028 | 0.05033 | 1.66501 | -8.500 | 2.400 |
| 0.04890 | 0.01527 | C.030.7 | 1.68840 | 8.500 | 2.600 |
| 0.12397 | 0.04423 | 0.01791 | 1.71135 | 8.500 | 2.800 |
| 0.30000 | 0.01317 | 0.00000 | 1.733.54 | 8.500 | 3.000 |
| $U Z(Z, R)$ | $\operatorname{UR}(2, R)$ | T(2, ${ }^{\text {( }}$ ) | $R O(2, R)$ | Z | $R$ |
| 0.3341 | 0.00600 | 0.24227 | 1.47212 | 9.000 | 0.000 |
| $0.3304=$ | 0.00413 | 0.24900 | 1.4742 C | 9.000 | 0.200 |
| 0.32121 | 0.00000 | 0.23332 | 1.43037 | 9.000 | 0.4 CO |
| 0.30596 | 0.01142 | 0.22206 | 1.49033 | 9.000 | 0.600 |
| 0.23593 | $0 . \mathrm{C1424}$ | C. 20361 | 1. 20365 | 3. CCC | 0.800 |
| 0.20237 | 0.01642 | 0.19191 | 1.51940 | 9.000 | 1. CCO |
| $0 \cdot 23637$ | 0.01795 | 0.17330 | 1.53822 | 9.000 | 1. 200 |
| 0.20392 | 0.01887 | 0.15343 | 1.55833 | 9.000 | 1.400 |
| 0.18074 | 0.01928 | 0.13305 | 1.57942 | 9.000 | 1.600 |
| 0.15294 | 0.01927 | 0.11248 | 1.50154 | 9.600 | 1.800 |
| 0.12537 | C.0.892 | 0.09214 | 1.62403 | 9.000 | 2.000 |
| 0.19853 | 0.01832 | 0.07229 | 1.64649 | 9.000 | 2.200 |
| 0.07557 | 0.01753 | 0.05309 | 1.66882 | 9.000 | 2.400 |
| 0.34747 | $0.0 \pm 650$ | 0.03403 | 1.690. 7 | 9.000 | 2.600 |
| $0.0233 \%$ | 0.01556 | 0.01694 | 1.71235 | 9.000 | 2.800 |
| 0.00000 | 0.01444 | 0.00000 | 1.73334 | 9.000 | 3.000 |


| U2 ( $2, R$ ) | $\cup R(2, R)$ | $T(Z, R)$ | $\mathrm{RO}(\mathrm{Z}, \mathrm{Q})$ | Z | $R$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0.30973 | C.CSOOO | 1). 22034 | 1.492, 1 | 9.500 | 0.000 |
| $0.2063 \%$ | 0.60298 | C. 21837 | 1.43427 | 9.500 | 0.333 |
| 0.2740 | O.CO 773 | 0.21255 | 1.47989 | 7.500 | 0.667 |
| $0 .-8472$ | 0.0.109 | i). 20322 | 1.508:3 | 9.500 | 1.CCO |
| 0.66714 | 0.0 .374 | 0.17027 | 1. $220 \leq 2$ | 9.500 | 1.333 |
| 0.34605 | $0 . C \in 20$ | 0.17bc9 | 1.535-3 | 7.500 | 1. 657 |
| $0 \cdot 22251$ | C.01789 | 0.15950 | 1.55217 | 9.500 | 2. CCO |
| 0.19756 | 0.01902 | 0.14109 | 1.57C=4 | 9.500 | 2.333 |
| 0.17179 | 0.0.365 | 0.12319 | 1.53007 | 9.500 | 2.667 |
| 0.14576 | 0.01796 | 0.10444 | 1.61042 | 9.500 | 3. CCO |
| 0.11987 | 0.01972 | ).0¢577 | $1.5311 \%$ | 9.500 | 3.333 |
| 0.6944. | $0.0+927$ | J.C6744 | 1.55208 | 9.500 | 3.667 |
| 3.0697: | 0.CLEA2 | 0.04252 | 1.67202 | 9.500 | 4.C00 |
| 0.114574 | O.C!776 | 0.03242 | 1.693 .35 | 9.500 | 4.333 |
| $0.02250$ | $0.0: 673$ | 0.01588 | 1.71387 | 7.500 | 4.667 |
| $0.00500$ | $0.0 .555$ | 0.60000 | 1.733 ct | 9.50 C | 5.000 |
| UZ ( $2, ~$ ) | UR ( $Z, R$ ) | $T(Z, R)$ | $\mathrm{RO}(2, \mathrm{R})$ | $Z$ | R |
| $0 \cdot 28652$ | O.COOCO | $0.19 \rightarrow 68$ | 1.51224 | 10.000 | $0 . \operatorname{CCO}$ |
| 0.29393 | O.0.325 | 0.19796 | 1.513.72 | 10.000 | 0.333 |
| 0.27653 | 0.07750 | C.19237 | 1.51836 | 10.000 | 0.667 |
| $0 \cdot 20450$ | 0.01081 | 0.18470 | 1.526\%9 | 10.000 | 1. CCO |
| 0.24877 | 0.01367 | 0.17332 | 1.53769 | 10.000 | 1.333 |
| $0 \cdot 2278 \cdots$ | 0.01502 | 0.16074 | 1. 55090 | 10.000 | 1.667 |
| 0.20863 | 0.01784 | 0.14596 | 1.56609 | 10.000 | 2.000 |
| 0.18583 | 0.61914 | 0.13000 | 1.582E4 | 10.000 | 2.333 |
| 0.16209 | 0.01790 | 0.11330 | 1.60075 | 10.000 | 2.667 |
| $0 \cdot 13774$ | $0 \cdot C<037$ | 0.09627 | 1.61943 | 10.000 | 3.000 |
| 0.11379 | 0.02040 | 0.07923 | 1.638 .8 | 10.000 | 3.333 |
| 0.08971 | $0.0<011$ | 0.06241 | 1.65771 | 10.000 | 3.667 |
| 0.06651 | 0.0 .955 | 0.04397 | $1.677<3$ | 10.000 | 4. 600 |
| 0.10436 | O.C 675 | 0.03307 | 1.69639 | 10.000 | 4.333 |
| $0.0215 \%$ | 0.01772 | 0.01475 | 1.71527 | 10.000 | 4.667 |
| 0.00006 | 0.0 .647 | 0.00000 | 1.73304 | 10.000 | 5.000 |


| $U Z(Z, R)$ | $U R(Z, R)$ | T (2, 2) | RO(2, R) | 2 | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $0 . ? 6432$ | c.00000 | 0.18016 | 1.53138 | 10.500 | $0 . \mathrm{COO}$ |
| $0.3620:$ | C.C C-374 | 0.17365 | 1.23248 | 10.500 | 0.333 |
| 0.5545 | 0.00731 | 0.17421 | 1.53731 | 10.500 | 0.657 |
| 0.24433 | 0.7 .758 | 0.16703 | 1. 54452 | 10.500 | 1. CCO |
| $0 \cdot \square 3 \bigcirc 77$ | 0.0 .346 | -.ivフ4 | 1.55424 | 10.500 | 1.333 |
| 0.2137) | $0 . C-507$ | 0.14303 | 1.566-7 | 10.500 | 1.567 |
| 0.13450 | 0.01780 | 0.13275 | 1.57993 | 10.500 | 2.000 |
| 0.17383 | 0.01924 | 0.11 s47 | 1.59516 | 10.500 | 2.333 |
| 0.15204 | $0 . C 2023$ | 0.10347 | 1.61148 | 10.500 | 2.667 |
| 0.1297 | 0.0 .077 | 0.04508 | 1.52857 | 10.500 | 3.000 |
| 0.1072 | 0.0 .073 | 0.07261 | 1.546.3 | 10.500 | 3.333 |
| 0.6447 | $0.0<082$ | $0.057-8$ | 1.66370 | 10.500 | 3.667 |
| $0.0627^{\circ}$ | 0.02035 | 0.04226 | 1.69163 | 10.500 | 4.000 |
| 0.1414 | 0.01960 | 0.02768 | 1.69932 | 10.500 | 4.333 |
| 0.0204. | C. 01858 | C.01357 | 1.71672 | 10.500 | 4.667 |
| 0.00000 | $0.0: 732$ | 0.00000 | $1.733: 4$ | 10.500 | 5.000 |
| UZ (Z,R) | $U R(2, R)$ | $T(2, P)$ | RO( $2, R)$ | 11.2 | - ${ }^{\text {R }}$ |
| $0.2429 \%$ | 0.00000 | 0.16107 | 1.54935 | 11.000 | 0.0 CO |
| $0.24 i 0 u$ | $0 . C 0365$ | 0.16036 | 1.55129 | 11.000 | 0.333 |
| 0.23517 | 0.00716 | 0.15047 | 1.55526 | 11.000 | 1.060 |
| $0 \cdot 2257$ | 0.01039 | 0.15018 | 1.56172 | 11.000 | 1.333 |
| 0.21312 | O.C1323 | 0.14177 | 1.57046 | -1.000 | 1.333 |
| 0.1978 | 0.01574 | 0.13155 | 1.58120 | 11.0co | 1. 667 |
| 0.18053 | 0.01776 | $0 \cdot 11970$ | 1.59352 | 11.000 | 2.033 |
| 0.16164 | 0.01933 | 0.10719 | 1.60740 | 11.000 | 2.333 |
| 0.1417 | $0.0<045$ | 0.09376 | 1.62222 | -1.000 | \%.860 |
| 0.12115 | $0.0 \geq 115$ | 0.07994 | 1.63776 | 11.000 | 3.333 |
| 0.1 CO 30 | $0.0<146$ | 0.06597 | 1.65376 | 11.000 | 3.337 |
| $0.6795)$ | $0.0<141$ | $0 \cdot 05212$ | 1.6697 | 11.000 | 4.000 |
| $0.0590 \%$ | 0.02102 | $0.033=0$ | 1.63620 | 11.000 | 4.333 |
| 0.03890 | 0.02031 | 0.02724 | 1.70231 | 11.000 | 4.667 |
| 0.01921 | O.C1930 | 0.01240 0.00000 | $1.7335 \frac{1}{4}$ | 11.000 | 5.000 |


| UZ（ $2, R$ ） | $L R(l, R)$ | $T(2,2)$ | $R O(Z, R)$ | Z | R |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $0.1223 *$ | C．COnOO | 0.14416 | 1.56797 | 11.5 C 3 | $0 . \mathrm{CcO}$ |
| 0． 206 | $0 . C) 358$ | 0． 14301 | 1.559 .6 | 11.500 | 0.333 |
| 0.21543 | $0 . \mathrm{C} 703$ | 0.13751 | 1.57271 | 11.500 | 0.667 |
| 0．0704 | 0.01724 | ก．13412 | 1.57849 | 11.500 | 1．OCO |
| 0．－467 | C．C1？ $\mathrm{Cl}_{13}$ | （）． 12073 | 1．54631 | 1：．500 | 1.333 |
| $0 . \overline{\mathrm{E}} 21 \mathrm{C}$ | C．C－5 54 | ）． 11775 | 1． 59574 | 11.500 | 1.667 |
| 0.16646 | $0.0+773$ | 0．1C746 | 1.6071 C | 11.500 | 2.000 |
| 0.14735 | 0.0 .739 | $0.09 .2<0$ | 1.61931 | 12.500 | 2.333 |
| 0.3110 | $0 . C 2063$ | 0.02425 | 1.0324 .7 | 11.500 | 2.607 |
| 0.1123. | $0 \cdot C-144$ | 0.07193 | 1．064 21 | 11．500 | 3.000 |
| 0.99320 | C．0 0136 | 2．05944 | 1．661？7 | 11．500 | 3.333 |
| 0.1740 | $0 . C-199$ | 1）．049．97 | 1.67604 | 11.500 | 3.667 |
| 0.55495 | $0.0-157$ | 0.03474 | 1.67074 | 11.500 | 4.000 |
| 0.03625 | $0.0<091$ | 0.02279 | 1.70532 | 11.500 | 4.333 |
| 0.2179. | 0.02741 | 0.01120 | 1.71970 | 1 i .500 | 4.667 |
| 0.0 CCO | $0 . C: 960$ | 0.00000 | $1.733=4$ | 11.500 | 5.000 |
| $U Z(Z, R)$ | $U R(\angle, R)$ | $T(2, R)$ | $R \cup(Z, R)$ | Z | R |
| $0 \cdot \therefore 024 c$ | O．C．000C | 0.12755 | 1．58544 | 12．000 | O．CCO |
| $0 \cdot 20090$ | C．CL352 | 0.12 う 6 | 1－596ヶ0 | 12.000 | 0.333 |
| 0.19623 | 0.00692 | 0.12360 | 1． 50965 | 12．00C | 0.6 .57 |
| 0.18377 | 0.01211 | 0.11381 | 1．59479 | 12.000 | 1．CCO |
| 0.1787, | $0 . C 2300$ | 0.11237 | 1.60176 | 12.000 | 1.333 |
| 0.16647 | $0 \cdot 0 \div 555$ | 0.10450 | 1.61035 | 12.000 | 1.667 |
| 0.15243 | 0.01770 | $\bigcirc .07547$ | 1.62032 | 12.000 | 2．000 |
| 0.13646 | $0 \cdot 61944$ | 0.08556 | 1.63141 | 12.000 | 2.333 |
| $0 \cdot 12047$ | 0． $\mathrm{C}=077$ | 0.07502 | 1.64337 | 12.000 | 2.667 |
| 0.10331 | 0.02167 | 0.06410 | 1.55575 | 12.000 | 3.000 |
| 0.08583 | 0.02218 | 0.05302 | 1.66831 | 12.000 | 3.333 |
| $0.0 ¢$ e23 | C．C $<229$ | 0.04195 | 1.68236 | 12.000 | 3.667 |
| 0.65073 | 0.02202 | 0.03103 | 1.69524 | 12.000 | 4．COO |
| 0.03347 | $0.0<139$ | 0.02937 | 1.70831 | 12.000 | 4.333 |
| $0.0165 \%$ | $0.0 \div 040$ | 0.01002 | 1.72119 | 12.000 | 4.657 |
| 0.00000 | －0．0こ908 | $0 . \mathrm{COOCO}$ | 1.733 － 4 | 12.000 | 5.000 |


|  |  |  | $R \cap(2, R)$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $U L$ 0.1 0 | CR(0.tr) c.00 | ¢.11183 | 1. DC C2:4 | 12.500 | 0.000 |
| 0.18164 | O.CO347 | 0.11097 | 1.503 .9 | 12.500 | 0.400 |
| $0 \cdot 775=$ | O. Cr 0 O 3 | 0.10542 | 1.50606 | 12.500 | 0.800 |
| $0: 1034$ | 0.05973 | 0.10427 | 1.610t1 | 12.500 | 1.200 |
| 0.10201 | 0.0.249 | C: 09507 | 1.51677 | 12.500 | 1.6C0 |
| 0.5107 | $0 . C-546$ | 0.69134 | 1.624;7 | 12.500 | 2.CCO |
| 0.1334 | 0.0.766 | 0.08377 | $1.633 \geq 0$ | 12.500 | 2.400 |
| 0. $1245 \%$ | 0.2-74 | 0.07331 | 1.54323 | 12.500 | 2. 9.0 |
| 0. -977 | 0.0 .025 | 0.00509 | 1.05 5304 | 12.500 | 3.200 |
| 0.99413 | $0.0=133$ | 0.05065 | 1.66430 | $12 \cdot 500$ | 3.600 |
| $0 \cdot \sim 783$ | 0.02024 | 2.04577 | 1.67631 | 12.500 | 4.000 |
| 0.0623. | O.C Ci< 58 | ).03703 | 1.547)7 | 12.う00 | 4.400 |
| 0.0453. | 0.02235 | 0.02741 | 1.69756 | 12.500 | 4.300 |
| $0.0306:$ | 0.02175 | 0.01760 | 1.71124 | 12.500 | 5.200 |
| 0.31514 | $0 . C-078$ | 0.20345 | 1.722.5 | 12. 200 | 5.600 |
| O. ¿CCOs | O.C. C 445 | $0 . \mathrm{COOCO}$ | 1.73334 | 12.500 | 6.000 |
| UZ (Z,R) | $U R(L, R)$ | T ( $2, R$ ) | RO( $2, R$ ) | 13.8 | - ${ }^{\text {R }} \mathrm{COO}$ |
| 0.18414 | $0.0 C D O C$ | 9.09599 | 1.61804 | 13.000 | O. 400 |
| 0.16273 | $0 . C 0342$ | 0.07623 | 1.61946 | 13.000 | -. 8.800 |
| $0 \cdot 5733$ | 0.00675 | 0.09405 | 1.62190 | 13.000 | - 1.800 |
| 0.15347 | $0 \cdot 00989$ | 0.09049 | 1.62558 | 13.000 | 1. 200 |
| 0.14550 | 0.C:278 | 0.0978 | 1.63178 | 13.000 | 1.000 |
| 0.1358 | $0.0 \div 536$ | 0.07778 | 1. 63734 | 13.000 | 2.400 |
| $0 \cdot 22454$ | $0 \cdot 61759$ | 0.07299 | $1.645 \% 9$ 1.65431 | 13.000 | 2.400 2.800 |
| 0.11223 | 0.0.944 | 0.1065 0.157 | 1.65431 | 13.000 | 3.200 |
| 0.39593 | $0 . C 2089$ | 0.65752 | 1.66362 | 13.000 | 3.600 |
| 0.08501 | $0 \cdot 02173$ | 0.04921 | 1.67341 1.68350 | 13.000 | 4. CCO |
| 0.67673 | C.0. | 0.04072 | 1.69373 | 13.000 | 4.400 |
| 0.0563. | C. C -277 | 0.03227 | 1.69373 1.70396 | 13.000 | 4.800 |
| 0.04172 | $0.0<259$ | 0.02399 0.01570 | 1.714:0 | 13.000 | 5.2C0 |
| 0.01376 | 0.02105 | 0.00772 | 1.72457 | 13.000 | 5.600 |
| 0.0 COO | 0.05772 | 0.00000 | 1.73334 | 13.000 | 6.000 |


| UZ ( $2, R$ ) |  | T (2, 2) | RO(2,R) | 13.50 | - |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0.14574 | 0.0 .000 | 0.05302 | 1.53439 | 13.500 | 0.100 |
| 0.14400 | $0.0-338$ | 0.03537 | 1.63419 | 13.500 | 0.400 |
| 0.1413 | $0 . C$ Coto | 0.08032 | 1.53710 | 13.300 13.500 | 1.200 |
| $0 \cdot 36.3{ }^{0}$ | -0. 0.00 | $\bigcirc .07340$ | 1.345 3 | 13.500 | 1.500 |
| 0.:2081 | 0.0 .525 | 0.00830 | 1.551:0 | 13.500 | 2.c00 |
| 0.11077 | 0.01750 | 0.06253 | 1.65711 | 13.500 | 2.400 |
| 0.29973 | 0.01937 | O.05017 | $1.565 \pm 8$ | 13.500 13.500 | 3:200 |
| 0.07583 | C.Cig | 0.04224 | 1.58171 | 13.500 | 3.600 |
| 0.0631 | 0.02260 | 0.0349 | 1.590'3 | 13.500 | 4.000 |
| 0.05021 | $0 \cdot 0.285$ | 0.02772 | 1.69927 | 13.500 | 4.460 |
| 0.0374. | 0.02270 0.02215 | 0.02053 0.01349 | 1.7081 $1.716 \frac{1}{5}$ | 13.500 | 5.800 |
| $0.0247 \%$ | 0.0 .215 | 0.01349 | 1.716254 | 13.500 | 5.600 |
| 0.00000 | 0.01788 | 0.00000 | 1.73304 | 13.500 | 6.000 |
| UZ (Z,R) | UR ( $2, R$ ) | $T(Z, R)$ | RO(Z,R) | 4.00 | - 00 |
| $0 \cdot 279$ | 0.09000 | O.06794 | 1.64919 | 14.000 14.000 1 | O.C00 |
| 0.1269 | $0 . C 15333$ | $C .00741$ 0.06785 | 1.6490 1.65100 | 14.000 | 0.800 |
| 0.11760 | 0.00966 | 0.00531 | 1.65434 | 14.000 | 1.200 |
| 0.1136 | 0.0252 | 0.06187 | 1.558 54 | 14.000 | 1.600 |
| 0.10615 | 0.05310 | 0.05755 | 1.66346 | 14.000 14.000 | 2.000 |
| 0.8975 0.0879 | 0.01725 | -0.04740 | 1.67555 | 14.000 | 2.8c0 |
| 0.07756 | $0 \cdot \mathrm{CLC74}$ | 0.04165 | 1.682't | 14.000 | 3.200 |
| 0.0667 | 0.02185 | 0.03565 | 1.58953 | 14.000 | 3.600 |
| 0.05553 | $0 \cdot 0 \leq 254$ | O.02754 | 1.69705 | 14.000 | 4.400 |
| 0.04425 | $C$. 0.0 0.0262 | 0.01733 | 1.71206 | 14.000 | 4.880 |
| $0: 92177$ | 0.02216 | 0.01139 | 1:71947 | 14.000 | 5.200 |
| 0.01073 | $0 \cdot 0<224$ | 0.00501 | 1.72674 | 14.000 14.000 | 5.600 6.000 |
| 0.0 cccu | 0.01792 | 0.00000 | 1.73334 | 14.000 | 6.000 |


| UZ（ $2, R)$ | UQ（1， 21 | T（ 2,2 ） | $\operatorname{RO}(2, R)$ | 14.500 | R ${ }^{R}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $0 \cdot 11040$ | 0．CいOOO | 0.05750 | $1.603=9$ | 14.500 | $0 . C 00$ |
| 0.10960 | 0．Cい328 | 0.05736 | 1.6633 C | 14.500 | 0.400 |
| 0.1072 |  | 0.05608 | 1．055？1 | 14.500 | 0.8 CO |
| $0.1034 ?$ | つ． 06052 | C． 05378 | 1.66778 | 14.500 | 1.200 |
| 0.93013 | 0.0 .234 | $\cdots .25115$ | 1．57112 | 14.50 C | 1．600 |
| $0.0917 \%$ | $0 . C-490$ | 0.04757 | $1.675 \% 4$ | 14.500 | 2．CCO |
| 0.08432 | 0.01713 | 0.04365 | 1．03CC3 | 14.500 | 2.400 |
| 0.07602 | 0.01702 | C．03920 | 1.68536 | 14.500 | 2.8 CO |
| 0.11609 | $0 . \mathrm{C}_{-} 053$ | O．0344 ${ }^{0.0}$ | 1．591）9 | 14.500 | 3．200 |
| $0.0577 \%$ | 0.02154 | 3．02749 | 1． 27711 | 14.500 | 3.600 |
| 0.04501 | 0．0． 035 | 0.02442 | 1.7633 C | 14.500 | 4.000 |
| 0.03824 | $0.02-65$ | त． $011+35$ | $1.709 \div 5$ | 14.500 | 4.400 |
| 0.0287 | 0．0 054 | 0.01434 | 1.71579 | 14.500 | 4.800 |
| 0.018 C | $0.0-203$ | 0.00742 | 1.72174 | 14.500 | 5.200 |
| 0.20 .732 | 0.0112 | 0．0こ4 04 | 1.72776 | 14.500 | 5.600 |
| 0.1 ccoo | $0.0-483$ | 0.00000 | 1.73384 | 14.500 | 6.000 |
| $U Z(Z, R)$ | $U R(L, R)$ | $T(Z, R)$ | $R O(Z, R)$ | Z | R |
| 0.09352 | $0 . C=900$ | C．04503 | 1.67647 | 15．000 | $0 . \mathrm{CCO}$ |
| $0.0928 ;$ | 0.01 .321 | 0.04628 | 1．676．9 | 15.000 | 0.400 |
| 0.0908 | 0．Cu634 | 0.04525 | 1.67812 | 15.000 | 0.800 |
| 0.0976 | 0.00332 | 0.04355 | 1.68014 | 15.000 | 1．200 |
| 0.09320 | 0．0．210 | 0.04127 | $1.682 \pm 7$ | 15.000 | 1.500 |
| 0.07777 | C．C 1461 | 0.03846 | 1.68625 | 15.000 | 2.000 |
| 0.07140 | 0.02682 | 0.03522 | 1.67016 | 15.000 | 2.400 |
| $0 . \cap 644 ;$ | $0 . C 1809$ | 0.03153 | 1.69451 | 15.000 | 2．800 |
| 0.05697 | 0．02018 | 0.02779 | 1.69719 | 15.000 | 3.200 |
| 0.04992 | $0.0<129$ | 0.02377 | 1.70403 | 15.000 | 3.600 |
| 0.04074 | $0.0 \leq 200$ | 0.01971 | 1.70912 | 15.000 | 4．000 |
| 0.13245 | $0 \cdot 0<231$ | 0.01502 | 1．714－0 | 15.000 | 4.400 |
| 0.02410 | $0.0<222$ | 0.01 .56 | 1．71925 | 15．000 | 4.800 |
| 0.01530 | 0.02173 | 0.00750 | 1.72423 | 15.00 C | 5．200 |
| 0.00740 | $0.0<085$ | 0.00374 | 1．72910 | 15.000 | 5.600 |
| 0.0 coc | 0．C：958 | 0.00000 | $1.733: 4$ | 15.000 | 6.000 |

## APPENDIX C

## ANALYTICAL SOLUTION OF AXIAL TEMPERATURE

 AND VELOCITY DECAYThe present analysis is based on the assumption that the rising air is in the turbulent motion, the turbulence being generated by the motion in the column itself, and is confined to conditions of no wind so that the axis of the column is aligned with the force of buoyance. The process taken into account are the buoyance of the column, its spreading and the entrainment of air into it, and the loss of momentum and heat by lateral diffusion.

Considering a ring-shaped volume of air as shown in Figure 2-2, in steady state yields an equation of continuity

$$
\begin{equation*}
\frac{\partial}{\partial z}\left(r u_{z} \rho\right)+\frac{\partial}{\partial r}\left(r u_{r} \rho\right)=0 \tag{c-1}
\end{equation*}
$$

an equation of verticle motion

$$
\begin{equation*}
\frac{\partial}{\partial z}\left(r u_{z}^{2} \rho\right)+\frac{\partial}{\partial r}\left(r u_{z} u_{r} \rho\right)=r \frac{\Delta t}{t a} \rho g+\frac{\partial}{\partial r}(r \tau) \tag{c-2}
\end{equation*}
$$

nad an equation of heat conservation.

$$
\begin{equation*}
\frac{\partial}{\partial z}\left(r u_{r} t \rho\right)+\frac{\partial}{\partial r}\left(r u_{z} t \rho\right)=\frac{1}{c p} \frac{\partial}{\partial r}(r F) \tag{c-3}
\end{equation*}
$$

All the quantities relate to mean values for the ring, and the verticle turbulent mixing has been neglected in comparision with the horizontal. The pressure is auumed to be undisturbed, the density is treated as constant except in so far as it affects the buoyance.

From Equation ( $\mathrm{C}-1$ ) and ( $\mathrm{C}-2$ ) may be derived the kinetic energy equation

$$
\begin{align*}
\frac{\partial}{\partial z}\left(\frac{1}{2} r u_{z}^{3} \rho\right)+\frac{\partial}{\partial r}\left(\frac{1}{2} r u_{r} u_{z}^{2} \rho\right)= & r u_{z} \frac{\Delta t}{t_{a}} \rho g+ \\
& u_{z} \frac{\partial}{\partial r}(r \tau) \tag{c-4}
\end{align*}
$$

and from Equation $(C-1)$ and $(C-3)$, yield

$$
\begin{align*}
\frac{\partial}{\partial z}\left(r u_{z} \Delta t \rho\right)+\frac{\partial}{\partial r}\left(r u_{r} \Delta t \rho\right)= & -\frac{1}{c_{p}} \frac{\partial}{\partial r}(r F) \\
& -r u_{z} \rho \frac{\partial t_{a}}{\partial z} \tag{c-5}
\end{align*}
$$

Equations ( $\mathrm{C}-2$ ), ( $\mathrm{C}-4$ ), and ( $\mathrm{c}-5$ ) are integrated from $\mathrm{r}=0$ to $r=\infty$ with the boundary conditions given in Chapter III which implies that at a sufficient radial distance, there will be no momentum transfer and heat flow. The Equations become,

$$
\begin{align*}
& \frac{d}{d z} \int_{0}^{\infty} r u_{z}^{2} \rho d r=\int_{0}^{\infty} r \frac{\Delta t}{t_{a}} \rho g d r  \tag{c-6}\\
& \frac{d}{d z} \int_{0}^{\infty} \frac{1}{2} r u_{z}^{3} \rho d r=\int_{0}^{\infty} r u_{z} \frac{\Delta t}{t_{a}} \rho g d r-\int_{0}^{\infty} r \tau \frac{\partial u_{z}}{\partial r} d r(c-6)  \tag{c-7}\\
& \frac{d}{d z} \int_{0}^{\infty} r u_{z} \Delta t \rho d r=-\int_{0}^{\infty} r u_{z} \rho \frac{\partial t_{a}}{\partial z} d r \tag{c-8}
\end{align*}
$$

Equation (C-6) states that the verticle gradient of momentum flux is equal to the buoyance of a horizontal stratum of unit thickness. Equation ( $C-7$ ) states that the verticle gradient of flux of kinetic energy is equal to the rate at which work is done by buoyant force less that rate which
work is done by the turbulent shear. Equation ( $C-8$ ) expresses the incremental weight flux past all successives planes; when the ambient air is constant, neutral condition, i.e. $\frac{\partial t_{a}}{\partial z}=0$, the increment is constant.

In order to solve the equations analytically, it is assumed that the shearing stress is a quadratic function of the relative velocity, i.e.,

$$
\begin{equation*}
\tau=1 / 2 \rho u_{z m}{ }^{2} j\left(\frac{r}{R}\right) \tag{c-9}
\end{equation*}
$$

and the similarity hypothesis is good for the radial profiles of $\Delta t$ and $u_{z}$. Following the argument of Sutton [10] that the profiles are Gaussian and the measures of dispersion are approximately at least, the same for $\Delta t$ and $u_{z}$, for which experimental confirmation in the laboratory had been provided by Rouse et al. [9] and by Railston [5], $\Delta t$ and $u_{z}$ are taking as

$$
\begin{align*}
& \Delta t=\Delta t_{m} \exp \left(-\frac{r^{2}}{2 R_{m} 2}\right)  \tag{C-10}\\
& u_{z}=u_{z m} \exp \left(-\frac{r^{2}}{2 R_{m}^{2}}\right) \tag{C-11}
\end{align*}
$$

Substituting Equation ( $C-9$ ), ( $C-10$ ), and ( $C-11$ ) into $(C-6),(C-7)$, and $(C-8)$ and then integrating gives

$$
\begin{align*}
& \frac{d}{d z}\left(R_{m}^{2} u_{z m}^{2}\right)=2 R_{m}^{2} \frac{\Delta t_{m}}{t_{a}} g  \tag{C-12}\\
& \frac{d}{d z}\left(R_{m}^{2} u_{z m}^{2}\right)=3 R_{m}^{2} u_{z m} \frac{\Delta t_{m}}{t_{a}} g-\alpha R_{m} u_{z m}{ }^{3} \tag{C-14}
\end{align*}
$$

No assumption has been made about the form of the function j, i.e. the value of $\alpha$; to the extent that is justifiable to assume that the relation between $\tau$ and $u_{z m}{ }^{2}$ is
independent of the stability of the enviroment, $\alpha$ becomes a universal constant. Multiplying Equation ( $\mathrm{C}-12$ ) by $3 \mathrm{u}_{\mathrm{zm}}$ and (C-13) by 2 and subtracting it follows immediately that

$$
\begin{equation*}
\frac{d R_{m}}{d z}=\alpha \quad \text { or } R_{m}=\alpha z^{\prime} \tag{c-15}
\end{equation*}
$$

where $z^{\prime}=z+z_{0}$. Further $\Delta t_{m}$ and $u_{z m}$ are solved by integrating factor method from the equations, the final form are:

$$
\begin{align*}
& \Delta t_{m}=\frac{A_{1}}{\alpha^{2} z^{2}}\left[\frac{3}{2} \frac{A_{1} g}{t_{a} \alpha^{2} z^{\prime}}+\frac{A_{2}}{z_{3}}\right]^{-1 / 3}  \tag{C-16}\\
& u_{z m}=\left[\frac{3}{2} \frac{A_{1} g}{t_{a} \alpha^{2} z},+\frac{A_{2}}{z^{1}}\right]^{1 / 3} \tag{C-17}
\end{align*}
$$

## APPENDIX D

## NON-LINEAR LEAST SQUARE FIT PROGRAM AND RESULT FOR THE SPREADING COEFFICIENT

The program set up is based on Scraborough's [23] non-linear least square fit method. Equation (3-33) is the key equation to solve for the spreading coefficient, $\alpha$, of the free jet from the experimental data of axial temperature and velocity decay. An iterative procedure was used until the absolute difference of the new $\alpha$ and the old $\alpha$ is less than $5 \%$ of the new $\alpha$.

```
B $WATFIV
    MoO FORMAT, (IIC,7FiOQ3)
    REAU(, (1Ci)
    10:
    FUKMAT(&F 1).3)
        DT=TD-TA
        RO=0.5*U
        Mil=0.25*[新㘯UU*[T
        M2=0.3/j*口**?*U0*口T*32.2/TA
        M3=0.1-5*0**!
        M4=U心**3
        M5=U.75*D*UD*DT*32.2/TA
        TAFO=AFO
        NNN=0
        20
            ST=^.
            DO 10 I= 1:
            TYA(I)= N+**3/((T(I)-TA)**3)
            LI(I)=i(I)-ZCT
            2IA(I)=R[)+ lT(1)*TAFO
            TYC(I)=M2*/TA(I)**5./TAFU +M 3*M4*ZTA(I) %*3.
            1-M3*M5:LTA(I)*** * T & E
            WRITE (6,5,2) ZTA(I),ZT(1), TAFO
    22 FDR:AT (3F:U.3)
            DTY(I)=-N2*ZTA(I)**5/TAFU*TAFU
            1+5.*ZT(I)*MO*ZTA(I) **4./TAFO
            3-2*!3**4*2l(1)*2TA(I)**2/TAFט
            ST=',T+ITYC(I)-TYA(I))*ETY(I)
            STu=STU+ 1)TY(I)*DTY(I)
        L" CGATIPIUE
            TAF=TAFO+ \thereforeT/STP
            TLiA= "PS((TAF-TAFU)}/TAFO
            IF (TDA.LE.D.Ci) GU TU 30
            NN=1!+?
            TAF]=TAF
            IF (NJ.GF..5) GO TO 31
            G1)T0,0
```



```
        3: L=U
            U^FI= \FO
    40) SU=0.
            SUL=0.
```



```
            ZU(I)=L(I)--ZCV
            ZU4(1)=RO+ZU(I) *UAFO
            WKIT: (o,5三) ZU(I),ZUA(I), UNFO
            UYC(I)=M2/(1)&FU*ZUA(I))+M3*(M4-M5/UAFO)/(ZUA(I)**3.)
            UUY(I) = - M**(c.*UAFU*LU(I)+RO)/(UAFO*ZUA(I)*
            GUAFO*ZUA(1)
            1-M3xM5/(UAFG和AFU
            I}**ZUA(I)**3)-3*M3*(M4-M5/UAFU)*ZU(I)/
            22UA(I)**4
            SU=SU+(UYC(I)-UYA(I))*DUY(I)
            SUU=SUIf+DUY(I)*EUY(I)
        j) CD.jT INUE
            UAF=UAFU+SU/SUM
            UCA=ABS((if,F-1AFO)/TAFO)
            IF (UDA.LE.C.O%) GO TU OO
            L=L+1
```

```
    UAFT=UAF
    IF (L.1,F.Is) GU TO
        4 1
    G0 T0 40
    4. WRITL (6.3')
    7% FURMAT (20X,'L=',I3)
```



```
    ICCEFICIENT OF FRFE JET',
    1/:2OX,GY WUN-LINEAR LIAST SQUARE FIT METHIJD',
    3//, 30X, :XIT VELOCITY, FT/SEC=',F10.5./.
    4 30X, 'EXIT PLUME TEMPEPATURE, DEG.R R=',F10.5,%,
    530X,'AMBIENT TEMPE:SATU?E, DEG. R=',F10.5,/,
    O30X,'DIAMETFR UF THE STACK, FT =',F10.5,1)
    WRITE (6,う\niG) UAF, TAF
99) FORMAT (/, 30x, 'SPREADIING CUEFFICIENT FOR VEL.',
    ''DLCAY=1,FG.3,/,
    <30X,'SPREAOIMG COEFFICIENT FOR TEMP. DECAY=1,F6.3)
        WRITE (0,1<0)
L2{ FORMATI/I/, iOX,'Z',10X,:UZ(Z)',8X,'UL(Z) CAL.',
    16x,'T(Z):,OX,'P(Z)'CAL.:'
        SDT=O.
        SDU=?:
        ADJ=C.
        UU 10 1=1, %
        TC(I)=AI/(TYC(I)*0.333)-TA
        ULC(I) =UYC(1)**0.333%
        SDT=(ABS(TC(I)-T(I)})*2+SD
```



```
        ADT=ABS((TSíI)-T(I))/T(I))+ACT
        ADU=ABS ((UZC(I)-UZ(I))/UL(I))+ADU
        WRITF (0,1}0) L(I),UZII),UZC(I),T(I),TC(I)
    i3O FORMAT (5FL5.こ)
70 COVTINU:
    SUT=(SUT/N)*れ0.5
    AUT=AUT/N*:O:)
    SUU=(SUU/N) ir:0.5
    A[)H=(a\Gamma(U/N )*100.
    WRITE (o,l4n) `UU,ACU,SDT,ADT
```



```
    L 30x, 'STO. NEVIATION=,FFG.2,%,
    230x,'AVE. FERCINT UEVIATIGiv=',F6.2,//,
    अ3UX,' \triangleXIAL DECNY 'JF TEMPERATURE',/,
    &30X:SID.DEVIATION=:,FG.2,IN,F'F6.2)
        STUP
        END
```

CALCULATE THE SPREADINS COEFICIENT OF FREE JET by non-linear least square fit method

EXIT VELOCITY, FT/SEC= 37.30 .00
EXIT TEMPERATURE, UEGOR $=912.00 .00$
CIAMETER OF THE JET,FI=0.340E OI
SPREADING COEFFICIENT FOR VEL. DECAY= 0.145
SPPEADING COEFFICIENT FUR TEMP. CECAY= $0.14:$
$U Z(Z) \quad C A L$.
36.17
33.00
$31 \cdot 11$
26.23
21.37
17.89
14.72
12.10
9.23
8.35

1121
879.20
850.20
823.20
701.50
722.90
688.100
665.00
641.80
622.50
603.20

T(Z) CAL.
870.37
845.96
845.96
811.07
711.07
763.78
730.01
694.56
672.89
655.92
637.40 637.40 631.40
618.54

AXIAL DECAY DF VELOCITY STO. CEVIATION= 2.11
AVG: PERCENT DEVIATION= 4.51

AXIAL DECAY OF TEMPERATURE STU. UEVIATION= 12.73
AVG. PERCENT DEVIATIUN= 5.39

CALCULATE THE SPREADING COEFICIENT OF FREE JET
BY NON-LINFAR LEAST SQUARE FIT METHOD
EXIT VELOCITY, FT/SEC $=35.60$
EXIT TEMPERATURE,DEG. R $=916.20$


SPREADING COEFFICIENT FOR VEL.CECAY= O. 150
SPREADING COEFFICIENT FDR TEMP. DECAY= C. 147

| 2 | UL(Z) | UZ(Z) CAL. | T(2) | T(z) CAL. |
| :---: | :---: | :---: | :---: | :---: |
| 2.00 | 34.18 | 35.111 | 877.00 | 8855.12 |
| 2.50 | 32.04 | 31.37 | 885.10 | 8810.47 |
| 3.00 4.00 | 27.70 24.92 | 24.00 | 700.00 | 755.57 |
| 5.00 | 21.36 | 21.20 | 724.70 | 720.31 |
| 6.00 | 18.15 | 17.51 | 689.30 | 692.87 |
| 7.00 | 15.66 | 16.68 | 662.50 643.00 | 655.20 |
| 8.00 9.00 | 11.04 | 13.05 | 623.50 | 640.10 |
| 10.00 | 9.26 | 11.48 | 611.80 | 632.13 |
|  |  | CAY OF VELOC gent deviat | 4.23 |  |
|  |  | ecay of tem ATIOV= $11 \cdot 2$ CENT DEVIAT | TURE $4.43$ |  |

CALCULATE THE SPREADING COEFICIENT OF FREE JET
BY NCN-LINEAR LEAST SQUARE FIT METHOD
EXIT VELOCITY, FT/SEC= 78.00
EXIT TEMPERATURE,DEG. R $=1058.00$ ANBIENT TENPERATURE, CEG. R=523.80 DIAMETER GF THE JET,FT=0.830E-O1

SPREADING CDEFFICIENT FUR VEL. DECAY $=0.105$
SPKEADING CDEFFICIENT FOR TEMP. DECAY= 0.097


$$
\begin{aligned}
& U L(Z) \\
& 75.60 \\
& 70.20 \\
& 62.40 \\
& 54.80 \\
& 46.80 \\
& 42.12 \\
& 37.40 \\
& 32.00 \\
& 30.42 \\
& 27.30
\end{aligned}
$$

UZ(Z) CAL.
76.44
68.64
59.28
53.04
48.36
42.90
39.01
34.32
32.76
29.64
$9(2)$
1004.50
940.47
881.70
822.95
790.70
764.19
737.31
710.77
694.74
684.06

T(Z) CAL.
871.03
833.63
801.58
801.58
769.53
769.53
748.16
726.79
705.43
694.74
AXIAL CECAY OF VELOCITY
STD: DEVIATION= 6 OD
AVG. PERCENT DEVIATICN= 6.43
AXIAL DECAY OF TEMPERATURE
STD.DEVIATION $=20.50$


CALCULATE THE SPREADING COEFICIENT OF FREE JET BY NON-LINEAR LEAST SRUARE FIT METHOD

EXIT VELCCITY, FT/SEC $=822.00$
EXIT TEMP
AMBIEJT TEMPERATURE, DEG. R $=527.40$
DIAMETER UF THE JET,FT=0.2DBE-Oi
SPREADING COEFFICIENT FOR VEL.DECAY $=0.032$
SPREADING COEFFICIENT FGR TEMP. DECAY= 0.070


## APPENDIX E <br> COMPUTER PROGRAM FOR IRRS <br> SIGNAL PROCESSING

The first program, 'Experimental Spectrogram of $\mathrm{CO}{ }^{\prime}$ recalls the experimental data of Mahagaokar [14] and print it out in the way as it was digitised in the tape. The second program, 'Simulated Spectrogram of $C O$ ' generates the $\mathrm{P}-\mathrm{R}$ structure of $C O$ and also the spectrogram that resembles the way as it was obtained from experiment.

```
                            UINFASION YSPEC(2100), NFILE(20),
        LN(20), TEMP(20),NPK(20)
            DIMENSION Y(12G), X(126)
```



```
            DATA LIST/',
            DATA H'312,322'
            DATA NFILE/1,5/
            DATA NPK/256,1181
                DATA TEMP/510.,510./
            OU 1000 I= 1,2% N(I),NFILE(I),NPK(I)
    1U4 FORMATTIHL,INX,'WAVE-NUMBER'VS SPECTROGRAM VALUE',I,
    l10X,'RUNNH=',',
    MEAD (1,102) (YSPEC(J),J=1,2046)
102 FOFMAT (11F12.5)
            NNPK=NPK(1)
            MNEAC (1,1D2) (YSPEC(J),J=1,NINPK)
    1 READ(1,102,ENC=103)
        W.KIT= (6,111)
111 FORMAT (ix, IDLMMY')
            1;0 Ti l
103 CCVTINUE
            SV=0.90
            SV S5V+ J= 1067,1151
    998 C.CNTINUE
            wRITL (6,11), SV
            FDRMAT (<OX,'SV=', F10.5)
            0u 393 J=1030,1162
            ANU=1:92781*J
            JJ=J-1035
            Y(J.J)=YSPEC(J)/SV
            x(JJ) = & N| 
            MKITE (6,105) J, ANU,YSPEC(J),Y(JJ)
973 CCNTINUE
    105 FURNNT (2nx,I5,3(5x,F12*5) )
            VMAX=3.830*(SGRT(0.180527*TEMP(I)) +0.b)
            PVMAX=2143.26- VMAX
            RVMAX=2143.26 + VMAX
                            \mathrm{ RIT-(6,112) TEMP(I),PVMAX,RVNAX}
    112 FOR4AT (%H1,20X,'TEMP,K=',F10.5,/,20X,'PVMAX=',
```



```
    11.IMUNT, IXUNTI
1000 CONTINUE
```



```
                            EEMP(1)=510.
                            \(V(1)=0\) 。
                            \(v(2)=1\).
    \(11 \mathrm{C}=1.7876 .8 \mathrm{E}-23\)
    \(\mathrm{B}=1.32\)
    \(\mathrm{mO}=2143.28\)
    \(F K=1.39054 E-23\)
    \(1=1\)
```



```
    1 FRRMAT (IH1, 1OX, 'TEMP, K', F10.2, /, \(20 X, J^{\prime}, 15 X,{ }^{\prime} E(V, J) '\),
    \(1 \because 2 x, 1 F(V, J)\)
    [1: \(337 \mathrm{~J}=1,40\)
    \(\mathrm{N}=\mathrm{j}-1\)
        \(E(K, J)=H C * B * M *(M+1)\)
    \(F(K, J)=(2 x M+1) * E X P(-E(K, J) /(R K * T E M P(I)))\)
    , RKITE \((6,5) \mathrm{A}, \mathrm{E}(\mathrm{K}, \mathrm{J}), \mathrm{F}(\mathrm{K}, \mathrm{J})\)
    FORMAT (20X,13,13X,E12.5,10X, :12.5)
377 LNTI:H1E
993 GONTINUE
    \(F F(40)=0\).
0095
    \(J \geq=4 l-J\)
    \(.11=41:-J\)
    \(\mathrm{JK}=\mathrm{CG} \mathrm{C}-\mathrm{J}\)
    \(F F(J)=F(i, J 2)\)
    \(F F(J!)=F F(J)\)
\(9 . j\) CDNTINUE
\(S F T=0\),
    UE 5354 VN \(=1,79\)
\(5 F T=S F T+F F(N)\)
5354
CONTINUE
    CC 396 N \(=1,77\)
    \(Y(N)=F F(N) / S F T\)
\(X(N)=19 马 3.20+(N-1) * 3.84\)
```



```
    Ij TURNAT (
Gio LDNTINUE
        \({ }^{2} \mathrm{~F} 1=0\)
        UC 5:5力 iv=18, 60
        \(S F I=S H 1+Y(N)\)
5355 CONTINUE
            \(S \mathrm{~F} 2=\) ?
```



```
            \(S F 2=S F ?+Y(N)\)
53.36 CONTINIE
```




```
    11., I Yリ!!T, IXU!IT)
```



```
        AN \(=J-1032\)
        \(x \times\left(N_{i j}\right)=1.9287 * J-0.1 * I I I\)
        \(\times \times 1=\times \times\left(N^{\prime} 1\right)-2\).
        \(x \times 2=x \times(N . i)+2\)
        \(X N=(x \times(N: j)-1) 93.20) / 3.84+1\).
        \(i=I F I X(X i d)\)
\(K=N+1\)
```

```
                            \((N)=1293.20+(N-1) * 3.94\)
\(x 1=x(i v)-0: 227\)
\(x_{2}=\times(14)+0.227\)
\(\times A 1=\times 1+3.8\)
```



```
    IF (xXI.LT:XE:AND:XXL:GT:XAZ) GO TO 702
    IF \((x \times 2 . G E \times A 2)\) GO TU 502
\(530 \quad A(N N)=0.227 * Y(N)\)
    701
    \(\mathrm{G}(\mathrm{J} T \mathrm{~T}=60027 * Y(N)+0.5 *((X \times 2-X A 1) * 2) * Y(N+1) / 0.227\)
        GC TO 600
    คา1 \(0 \Delta 1=\left(X X_{1}-X 1\right) / 0.227\)
```



```
    00 Ti 552
    \(.5141=(1) \Delta 1-1) * Y.(N) *(X 2-X X 1)\)
    552 MA2 \(=(X A Z-X \times 2) / 0.227\)
    IF (CA2-1.) \(560,560,561\)
    万力 0 A \(2=0.227 * \dot{Y}(K)-Y(K) * D A 2 *(X A 2-X X Z) * 0.5\)
    (0) TO 570
\(501 \stackrel{4}{2}=(C A 2-1.1 * Y(K) *(\times \times 2-\times A 1)\)
```



```
    \(A(N N)=0.227 * Y(K)\)
    \(\therefore\) !2 \(A(N N)=0.227 * Y(K)\)
    600 CONTINUE
    \(\triangle S F T=0\).
    0062 i NH \(=1,152\)
    \(\angle S F T=A S F T+A\) (NN)
    col CoNTINDE
    \(4 \mathrm{SFl}=0\) 。
    \(D 0,5359 N=35,117\)
    \(A S F I=A S F I+A(\cdot 1 H)\)
5359 CCNTINUE
    \(4 S F 2=0\).
        DO, 360 : N \(V=47,107\)
    \(A S F 2=\Lambda S F 2+A(1 N)\)
    5350 CONTINUE
    WRIT \((6,12)\) ASFT, \(\triangle S F 1, A S F 2\)
        12 FIRMAT \(1 / 1 /, 20 X, 1 A S F T=1, F 10.5,1,20 \mathrm{X}, \mathrm{ASFi}=1, F 10.5,1\),
            \(12 C X,{ }^{\prime} A S F 2=1, F 10.5\) )
            CO GRZ N:I=1; 152
\(R S(N Y)=A(N N) / A S F T\)
            RS \((N y)=A(N N) / A S F T\)
WRTG \((6,10)\) NN, XX(NJ), A(NN), RS(NN)
        60L CONTINUET1 (1953.20,i.9273,RS,152,0.,0.,1,LIST,1.,
            2IYUNT, IXUNTO
    STO CCNTINUE
STOP
        SEOP
ENE
```


[^0]:    R
    0.000
    0.100
    0.100
    0.200
    0.200
    0.300
    0.405
    0.500
    0.600
    0.700
    0.800
    0.900
    1.000
    1.100
    1.200
    1.300
    1.400
    1.400
    1.500
    1.500
    1.600
    1.700

