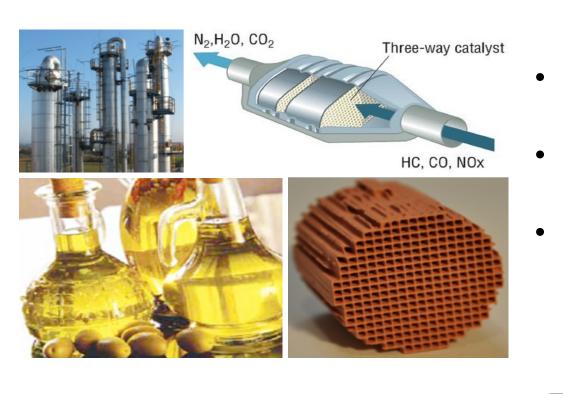
Tailoring the Physiochemical Properties of SSZ-13 Zeolite with Polyquat Growth Modifiers

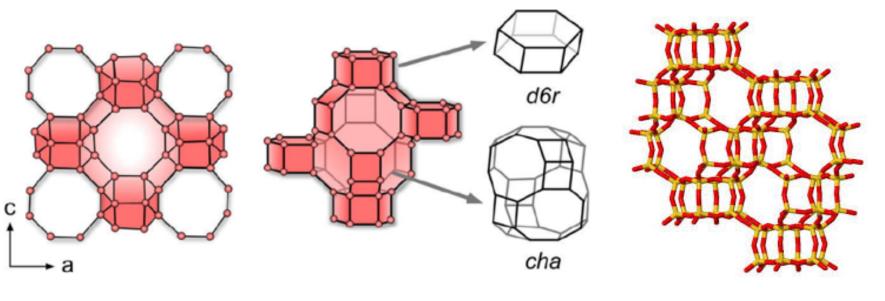
UNIVERSITY of HOUSTON

Jakob Claret, Heng Dai, and Jeffrey D. Rimer University of Houston, Department of Chemical & Biomolecular Engineering,

Background

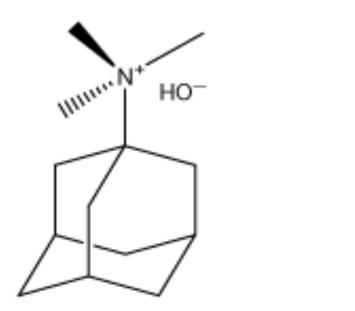


- Zeolite = 'boiling stone' in ancient Greek
- Crystalline aluminosilicate materials
- Widely used for catalysis, adsorbates and emission control



Kumar et al. J. Am. Chem. Soc. 2015.

- SSZ-13 (chabazite framework) consists of six membered rings of tetrahedral silica and alumina combined into supercages
- Commonly synthesized using TMAda-OH as the organic structure-directing agent (OSDA)
- Charged quaternary amines (e.g. PDDAC) promote crystal growth in addition to the OSDA



CI - n
H₃C CH₃

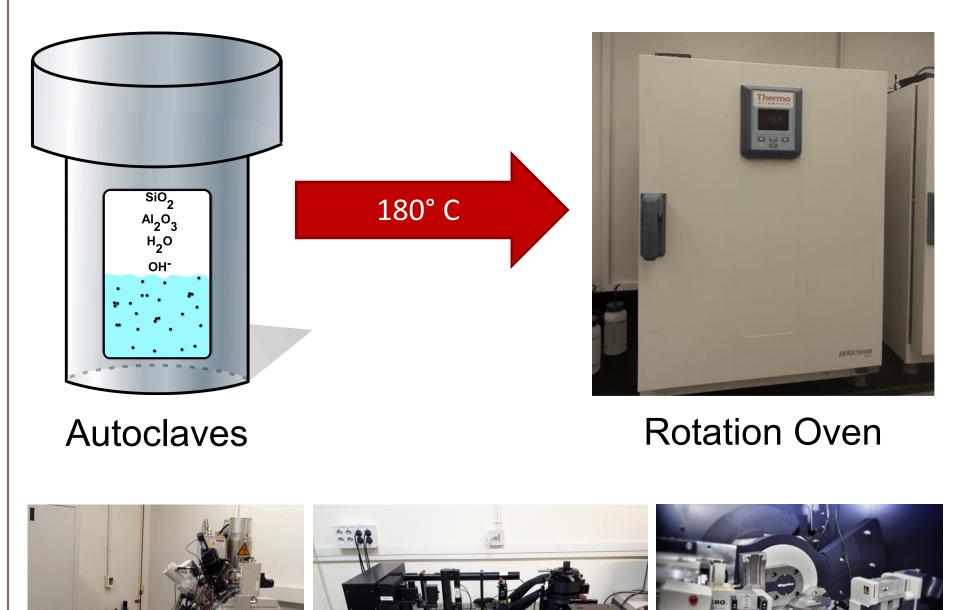
Trimethyladamantylammonium hydroxide (TMAda-OH)

Polydiallyldimethylammonium chloride (PDDAC)

Experimental Methods

- Typical zeolite synthesis:
 - Mixing and aging the reactants
 - Hydrothermal synthesis with rotation
- Powder X-ray diffraction analysis (XRD)
- Scanning electron microscopy (SEM)
- Elemental analysis: electron dispersive spectroscopy (EDS)
- Dynamic light scattering (DLS)

SEM

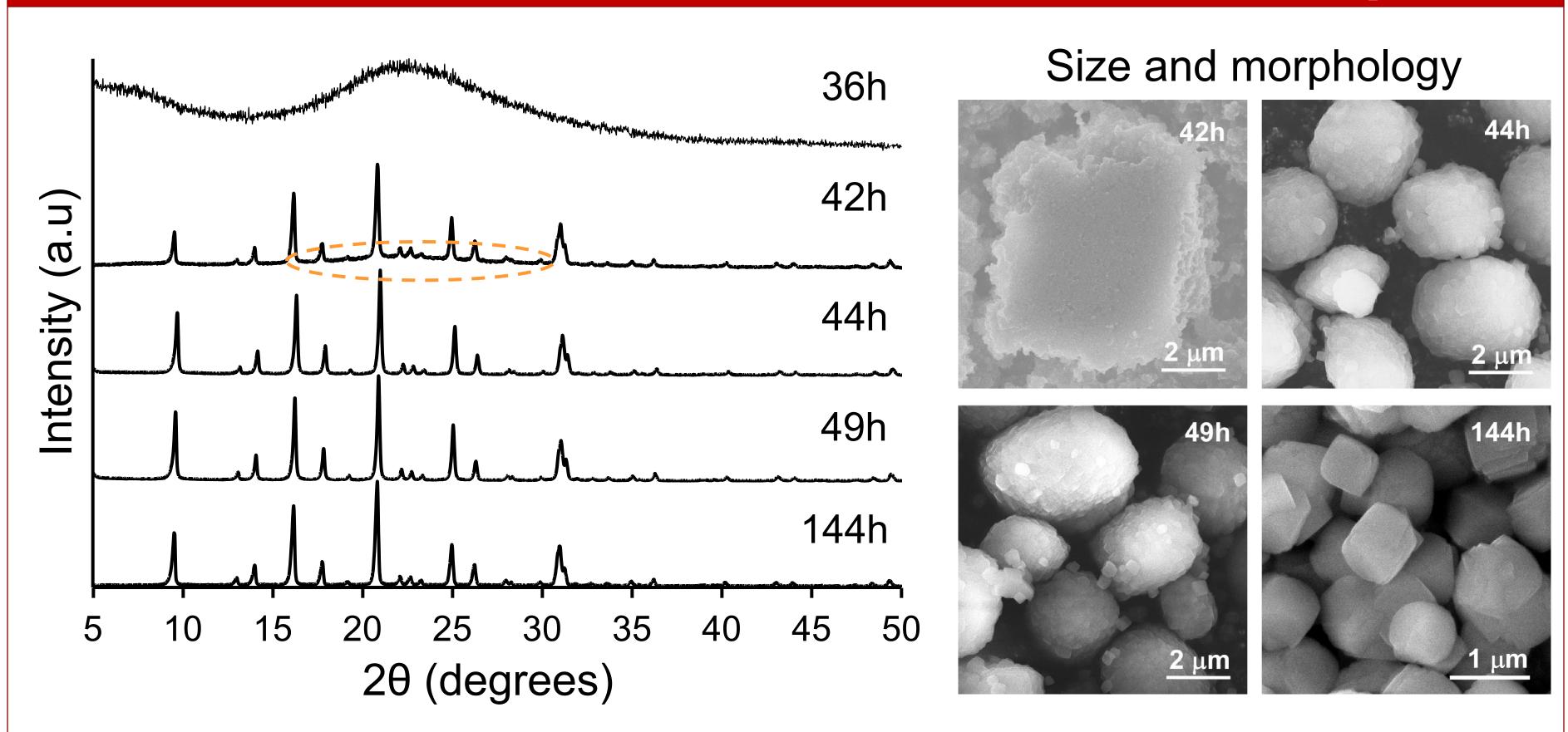


DLS

XRD

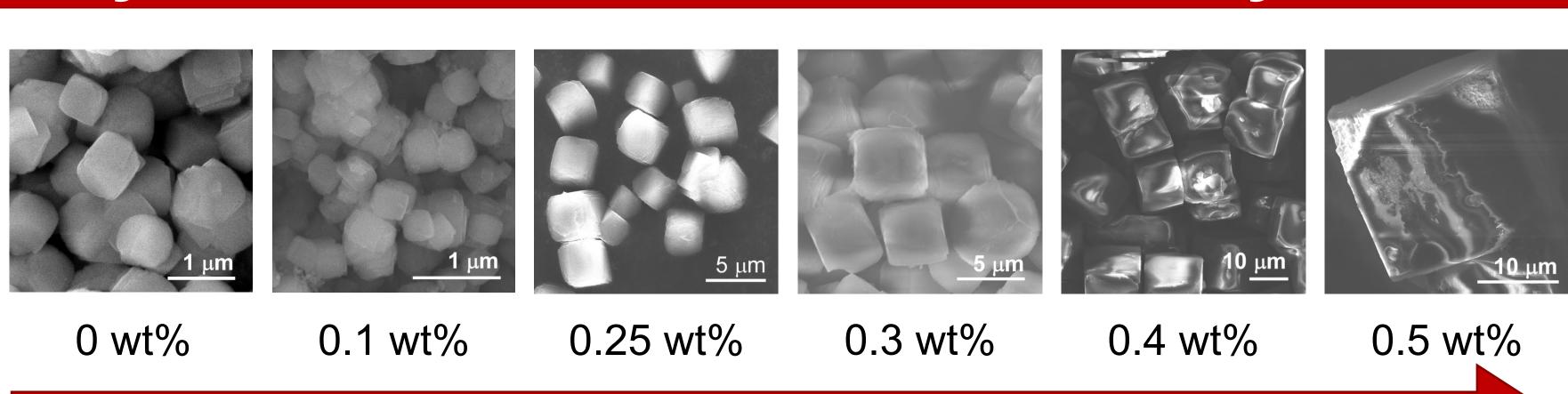
Ex situ Characterization of Control Sample

4800 Calhoun Rd, Houston, TX 77204

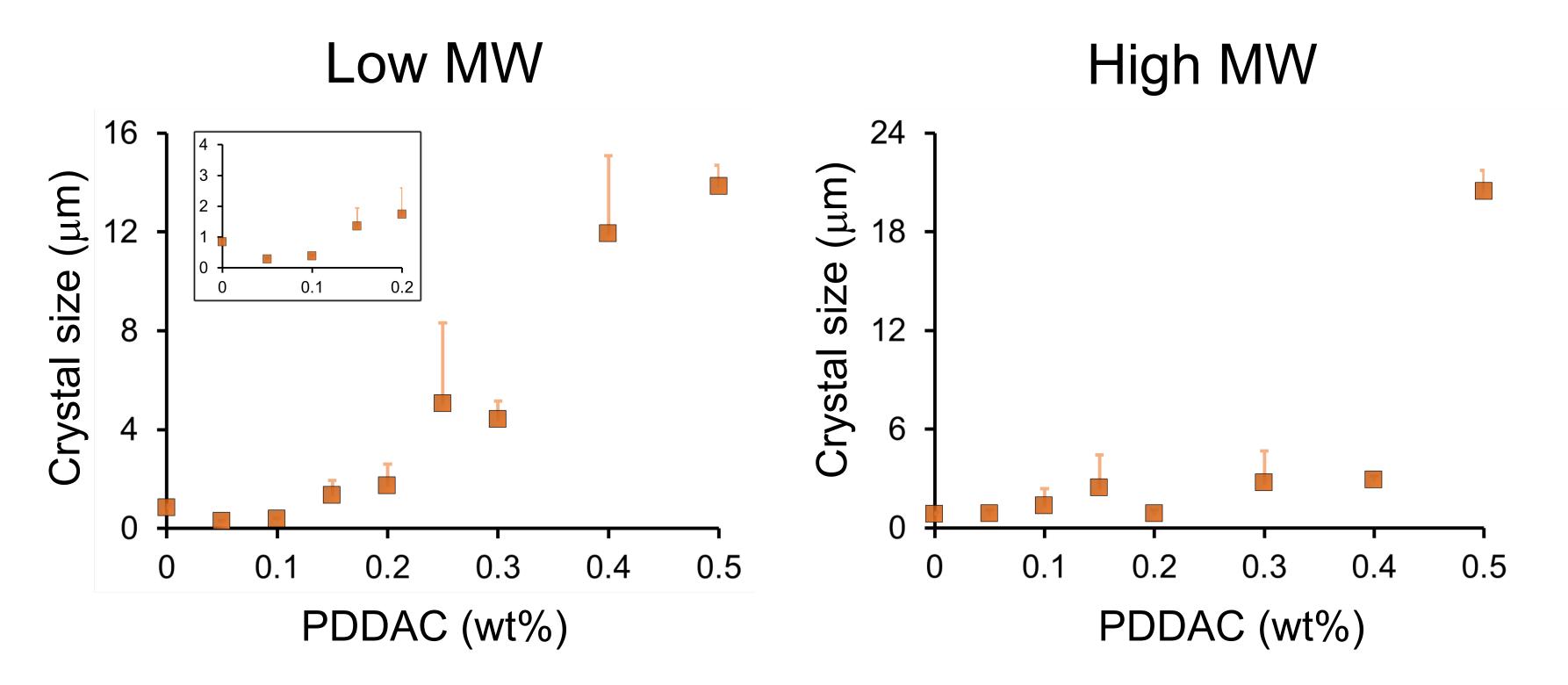


- Induction period lasts approximately 30 to 40 hours
- Smooth cubic morphology after 144 hours synthesis

Polymer Concentration Effects on Crystal Size

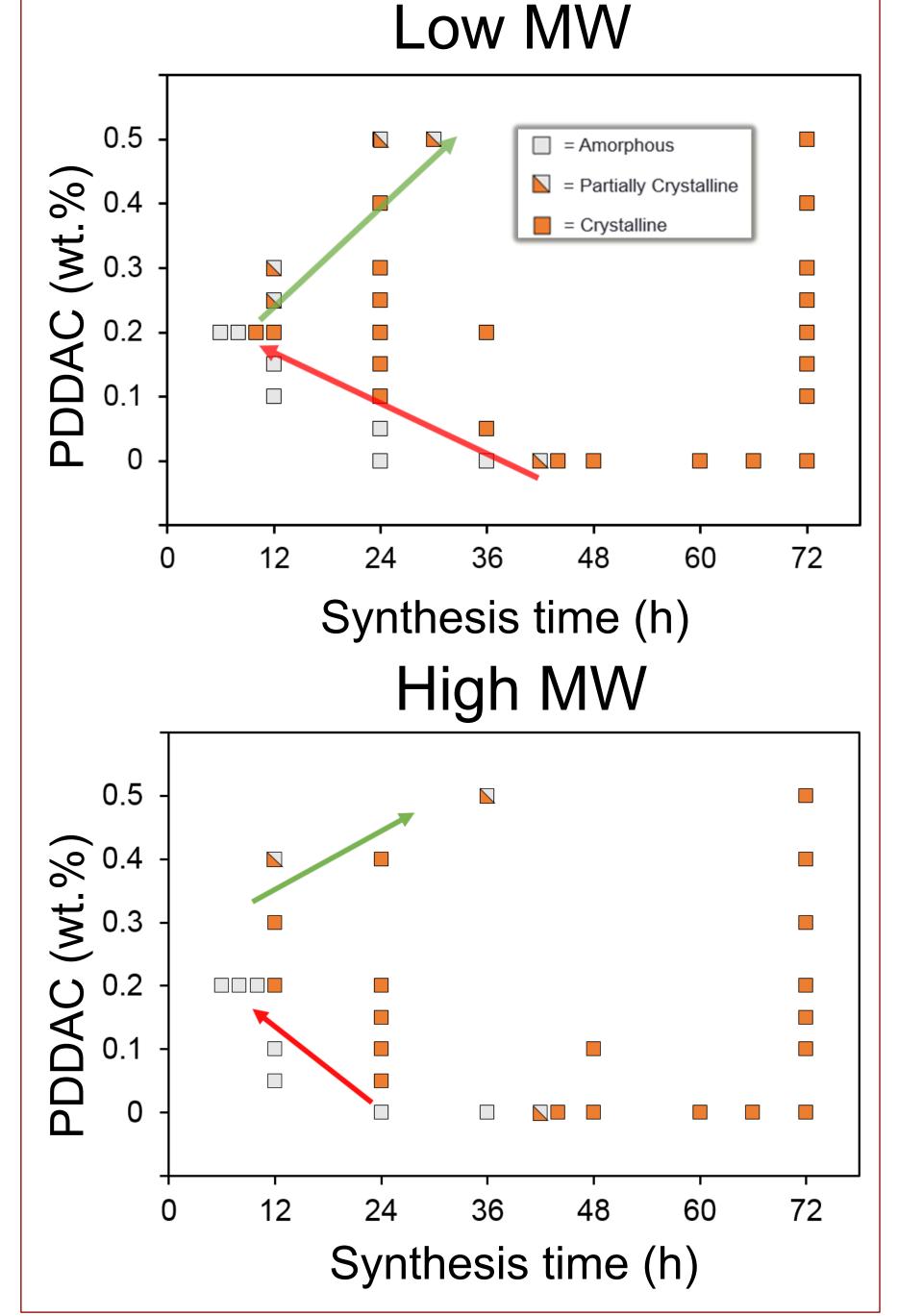


Increasing polymer concentration



- Polymer addition can tune the crystal size by up to two orders of magnitude
- Crystal size is reduced in certain polymer concentrations for low MW polymer
- Nearly no observable size reduction trend for high MW polymer addition

Crystallization Time



Summary

- Crystal size generally increases with increasing low MW polymer concentration
- Synthesis time can be shortened using certain polymer concentrations

References

- Kumar et al. *J. Am. Chem. Soc.* **2015**, 137(40), 13007-13017.
- Lupulescu et al. *J. Am. Chem. Soc.* **2013**, *135*(17), 6608-6617.

Acknowledgments





